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Black Carbon Incorporation Effect on Optical Properties of Poly-**Methyl Methacrylate Films**

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ABSTRACT

The optical properties of poly-methyl methacrylate (PMMA) films with difference filling levels of black carbon (BC) were studied in the visible and ultraviolent wavelength regions. It was found that the optical absorption is due to allowed- and, forbidden- direct transitions and the energy gaps decrease with increasing BC content for all transitions, while the width of the tail localized states increase with increasing BC content. The optical constants refractive index n, extinction coefficient k, the real ϵ_r and imaginary ϵ_i parts of the dielectric constant have been also calculated. The dielectric constant increased in the composite samples as compared with the pure PMMA sample prepared by the same method.

Keywords: polymer, composite, fillers, optical properties, dielectric constant

INTRODUCTION

It is well known that the physical properties of polymers may be affected by filling. Certain structural, optical, mechanical, electrical and magnetic properties of the selected polymer can be controllably modified owing to the type of the filler used, its concentration and the way in which it penetrates and interacts with the chains of the polymer. Detailed studies of polymer filled with different filling levels of a certain filler allow the possibility of choice of the desired properties [1-5].

Poly-methyl methacrylate (PMMA) has been widely used in architecture, automobile, air and railway transport systems due to its superior optical and mechanical properties [6-7]. This wide range of applications of PMMA can be even more extended by incorporation of filler into PMMA matrix, because well dispersed filler may enhance various physical properties of PMMA. The aim of the present work is too concerned with the investigation of the effect of black carbon filler of various filling levels on the optical properties of poly-methyl methacrylate (PMMA) films.

OPTICAL BAND GAP

Practically the optical absorption coefficient \propto can be calculated from the relation [8]:

$$ln\left(\frac{I_o}{I_t}\right) = 2.303 \, A = \propto d \tag{1}$$
 where I_o and I_t are the intensities of the incident and transmitted beams respectively, A the optical absorbance

and d is the film thickness [5,9].

The extinction coefficient K, is related to the absorption coefficient α , through $K = \alpha \lambda / 4\pi$, where λ is the wavelength of light [10-12].

The refractive index as a function of wavelength can be determined from the reflection coefficient data R and the extinction coefficient *K* using equation [13]:

$$n = \left(\frac{4R}{(1-R)^2} - k^2\right)^{\frac{1}{2}} - \frac{(R+1)}{(R-1)} \tag{2}$$

The absorption edge for direct and non-direct transitions can be obtained in view of the models proposed by Tauc et al [14-17]:

$$\propto hv = C_o(hv - E_g^{opt})^n \tag{3}$$

where C_o is a constant related to the properties of the valance and conduction bands [18], hv is the photon energy, \propto is the absorption coefficient, E_g^{opt} is the optical energy band gap of the material and n = 1/2, 3/2, 2, or 3 for direct allowed, direct forbidden, indirect allowed and indirect forbidden transitions respectively [19-22].

A plot of $(\propto hv)^{1/n}$ versus (hv) often yields a reasonably good straight line fit to the absorption edge and the extrapolated (hv) at which $(\propto hv)^{1/n} = 0$ provides a convenient experimental benchmark for the optical band gap E_q^{opt} [23].

The optical absorption coefficient \propto (v) near the band edge shows an exponential dependence on photon energy (hv) and obeys an empirical relation due to Urbach [24-26]

$$\alpha (v) = \alpha_o \exp(hv/E_t) \tag{4}$$

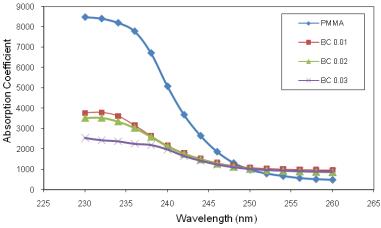
where \propto_0 is a constant and E_t is related to width of the band tails of localized states in the forbidden band gap. it should be mentioned that this equation is applicable only in the low absorption region ($\propto 10^3 - 10^4 cm^{-1}$).

EXPERIMENTAL DETAIL

Films of PMMA-BC composite were prepared by solution casting: 2g of pure poly-methyl methacrylate (PMMA) supplied from Sigma-Aldrich company has been dissolved in 20ml of dicholoro methylene to obtain a pure PMMA film, then the same amount of PMMA was dissolved with 0.01, 0.02, and 0.03 of CB to prepare the different filling levels of the PMMA-BC composites, these homogeneous solutions were spread on a glass plate and allowed to evaporate the solvent slowly in air at room temperature for 24h. The thickness of the films was in the range of $(100 - 470)\mu m$, it was determined using micrometer at different places in each film and an average was taken.

RESULTS AND DISCUSSION

The optical absorption spectra of all films were recorded at room temperature, by UV-VIS double beam spectrometer (Model: Lambda 25) in the wave length range from 190 to 1100 nm. The optical band gap of these samples was evaluated from the photon energy absorption plot.



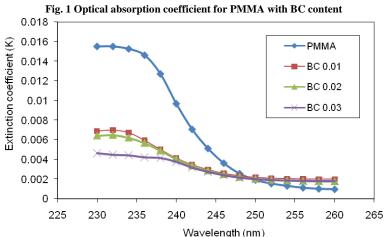


Fig. 2 Extinction coefficient K as a function of wavelength for PMMA polymer with BC content

Figure 1 shows the absorption coefficient as a function of the wavelength for pure PMMA and different concentration of BC filers, for low wavelengths, in the UV region the absorption coefficient is decreasing exponentially. The extinction coefficient K over the exponential absorption region (230 - 260)nm for all polymer samples are given in Fig. 2. The variations of refractive indices n of all films with photon energy are shown in Fig. 3. It is clear from these figures, that the refractive indices n and extinction coefficient K are decrease with increase of wavelength. The increase of BC content to the polymer lead to more stability of refractive indices at high values, which are useful in the industry of reflectors.

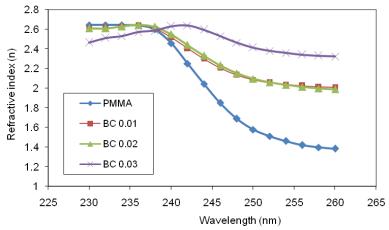


Fig. 3 Refractive index as a function of wavelength for PMMA with BC content

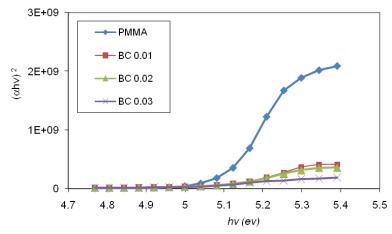


Fig. 4 Relation between the $(\propto hv)^2$ and hv for PMMA with BC content

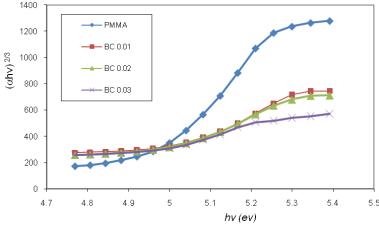


Fig. 5 Relation between the $(\varpropto hv)^{2/3}$ and hv for PMMA with BC content

Figures 4 and 5, shows the dependence of $(\propto hv)^{1/n}$ on the photon energy (hv) for direct allowed (n=1/2), direct forbidden (n=2/3) transitions, respectively. It is to be noticed that the curves are characterized by the presence of an exponentially decaying tail at low photon energy. The optical energy gap E_g^{opt} was estimated from the

extrapolation of the linear portion of the graph to the photon energy axis. It is observed that E_g^{opt} decreased with black carbon content. All plots show straight lines with some deviations from linearity at the lower of \propto , which were suggested by many researchers [27-30] as possibly due to imperfections in the material, but this region of the curve is still not fully understood. The mean vale of E_g^{opt} for direct-, allowed and forbidden transitions are tabulated in Table -1.

The Urbach plot is presented in Fig. 6 in which the natural logarithm of absorption coefficient is plotted as a function of photon energy hv. The slop of each line yields the magnitude of $1/E_t$. The measured values of width tails localized states in the band gap for all samples are tabulated in Table -1.Table1 shows that the optical energy gap E_g^{opt} decreases with increasing BC contents for all transition, on the other hand the width of the band tails of the localized state shift towards higher energies which range from (0.1234 eV) for pure PMMA to (0.3058 eV) for 0.03 filer concentration. This may be attributed to the increase in the disorder caused by the changes in the degree of crystallization, which is known to increase the width of the localized states, thus reduce the value of the optical gap. Such a change has been reported by others for different polymer compositions [11,31].

Type of transitions	Pure PMMA	PMMA-0.01 BC	PMMA-0.02 BC	PMMA-0.03 BC
Direct allowed	5.1024 eV	5.0455 eV	4.9805 eV	4.9642 eV
Direct forbidden	4.9715 eV	4.7833 eV	4.6875 eV	4.6597 eV
Width of band tail	0.1234 eV	0.2978 eV	0.2822 eV	0.3058 eV

Table -1 Optical Energy Band and Band Tail for PMMA with BC Content

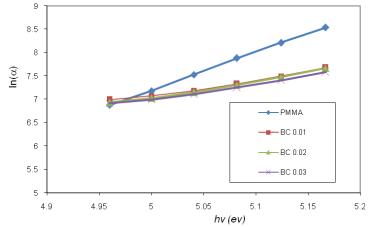


Fig. 6 Relation between the $ln(\propto)$ and hv for PMMA with BC content

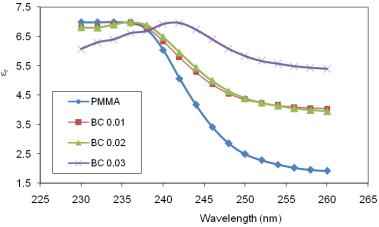


Fig. 7 Relation between the real part of dielectric constant ϵ_r and wave length for PMMA with BC content

The real and imaginary parts of dielectric constant can be calculated by using equation [32]:

$$\epsilon_r = n^2 - K^2, \quad \epsilon_i = 2nK \tag{5}$$

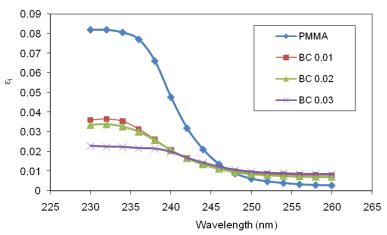


Fig. 8 Relation between the imaginary part of dielectric constant ϵ_i and wave length for PMMA with BC content

Figures 7 and 8, show the dielectric constants ϵ_r and the dielectric loss factor ϵ_i versus the photon energy for all samples. The significant decrease of loss factor can be observed in the wavelength range (235 - 245)nm and finally reaches nearly a constant value.

Both the electric nature of the polymer and the effect of BCs in the composites can be understood by correlating the dielectric constant values with the refractive index. If the sample is non-polar insulator, the dielectric constant ϵ_r for long wavelength can be expressed by Maxwell's equation $\epsilon_r = n^2$ [33]. The difference between the squared refractive index and the dielectric constant is a result of permanent dipoles and the semi-conductive character of the sample [34-36]. The difference between ϵ_r and n^2 for the pure sample is greater than that observed in the composites samples, wich indicated that the PMMA has permanent dipoles.

CONCLUSION

The optical properties of poly-methyl methacrylate (PMMA) films with difference concentrations of black carbon (BC) were investigated at room temperature. The band tail width obeys Urbach's empirical relation. The optical absorption is due to all direct and indirect transition. Generally, the optical band gap decreases with increasing BC contents for all transitions. The value of refractive indices and dielectric constant were tended to be more stability at high values by increasing the black carbon content in PMMA matrix, through the investigated range.

REFERENCES

- [1] A El-Khodary, AH Oraby and AE Youssef, The Effect of AgNO₃ Filler on the Physical Properties of Polystyrene Films, *International Journal of Material Science*, **2008**, 3, 11-24.
- [2] K Ramly, MIN Isa and ASA Khiar, Conductivity and Dielectric Behavior Studies of Starch/PEO+xwt-% NH₄NO₃ Polymer Electrolyte, *Materials Research Innovations*, **2011**, 15, S82-S85.
- [3] SF Bdewi, OG Abdullah, BK Aziz and AAR Mutar, Synthesis, Structural and Optical Characterization of MgO Nanocrystalline Embedded in PVA Matrix, *Journal of Inorganic and Organometallic Polymers and Materials*, **2016**, 26 (2), 326-334.
- [4] OG Abdullah, YAK Salman and SA Saleem, In-situ synthesis of PVA/HgS Nanocomposite Films and Tuning Optical Properties, *Physics and Materials Chemistry*, **2015**, 3(2), 18-24.
- [5] MF Shukur and MFZ Kadir, Hydrogen Ion Conducting Starch-Chitosan Blend Based Electrolyte for Application in Electrochemical Devices, *Electrochimica Acta*, **2015**, 158, 152-165.
- [6] OG Abdullah and DS Muhammad, Physical Properties of Pure and Copper Oxide Doped Polystyrene Films, *International Journal of Materials Science*, **2010**, 5 (4), 537-545.
- [7] KR Mohan, VBS Achari, VVRN Rao and AK Sharma, Electrical and Optical Properties of (PEMA/PVC) Polymer Blend Electrolyte Doped with NaClO₄, *Polymer Testing*, **2011**, 30 (8), 881-886.
- [8] OG Abdullah, Influence of Barium Salt on Optical Behavior of PVA based Solid Polymer Electrolytes, *European Scientific Journal*, **2014**, 10 (33), 406-417.
- [9] DS Vicentini, AJ Smania and MCM Laranjeira, Chitosan/Poly (Vinyl Alcohol) Films Containing ZnO Nanoparticles and Plasticizers, *Materials Science and Engineering C*, **2010**, 30 (4), 503-508.
- [10] OS Hamad and OG Abdullah, A Missing Point in The Binding Literature, *Applied Science Reports (ASR)*, **2014**, 1, 57-73.

- [11] KH Kim and WH Jo, Improvement of Tensile Properties of Poly (Methyl Methacrylate) by Dispersing Multi-
- Walled Carbon Nanotubes Functionalized with Poly(3-Hexylthiophene)-Graft-Poly (Methyl Methacrylate), *Composites Science and Technology*, **2008**, 68 (9), 2120-2124.
- [12] OG Abdullah, RR Hanna and YAK Salman, Structural, Optical and Electrical Characterization of Chitosan Methylcellulose Polymer Blends Based Film, *Journal of Materials Science: Materials in Electronics*, **2017**, 28 (14), 10283-10294.
- [13] ASA Khiar and AK Arof, Conductivity Studies of Starch-Based Polymer Electrolytes, *Ionics*, **2010**, 16 (2), 123-129.
- [14] AF Abdulameer, MH Suhail, OG Abdullah and IM Al-Essa, Fabrication and Characterization of NiPcTs Organic Semiconductors Based Surface Type Capacitive-Resistive Humidity Sensors, *Journal of Materials Science: Materials in Electronics*, **2017**, 28 (18), 13472-13477.
- [15] OG Abdullah, YAK Salman and SA Saleem, Electrical Conductivity and Dielectric Characteristics of In-Situ Prepared PVA/HgS Nanocomposite Films, *Journal of Materials Science: Materials in Electronics*, **2016**, 27 (4), 3591-3598.
- [16] MF Shukur, R Ithnin and MFZ Kadir, Ionic Conductivity and Dielectric Properties of Potato Starch-Magnesium Acetate Biopolymer Electrolytes: The Effect of Glycerol and 1-Butyl-3-Methylimidazolium Chloride, *Ionics*, **2016**, 22 (7), 1113-1123.
- [17] OS Hamad, OG Abdullah and MM Ameen, Optimum Design of Anode for Dow Electrolytic Cell, *International Journal of Applied Physics and Mathematics*, **2012**, 2 (6), 417-421.
- [18] OG Abdullah, BK Aziz and SA Hussen, Optical Characterization of Polyvinyl Alcohol Ammonium Nitrate Polymer Electrolytes Films, *Chemistry and Materials Research*, **2013**, 3 (9), 84-90.
- [19] C Umadevi, AK Sharma and VVRN Rao, Electrical and Optical Properties of Pure and Silver Nitrate-Doped Polyvinyl Alcohol Films, *Journal of Materials Letters*, **2002**, 56 (3), 167-174.
- [20] OG Abdullah, Synthesis of Single-Phase Zinc Chromite Nano-Spinel Embedded in Polyvinyl Alcohol Films and its Effects on Energy Band Gap, *Journal of Materials Science: Materials in Electronics*, **2016**, 27 (11), 12106-12111.
- [21] J Ballato, S Foulger and DW Smith, Optical Properties of Perfluorocyclobutyl Polymers, *Journal of the Optical Society of America*. *B*, **2003**, 20 (9), 1838-1843.
- [22] OG Abdullah and SA Hussen, Variation of Optical Band Gap Width of PVA Films Doped with Aluminum Iodide, *Advanced Materials Research*, **2012**, 383-390, 3257-3263.
- [23] V Krishnakumar and G Shanmugam, Electrical and Optical Properties of Pure and Pb²⁺Ion Doped PVA PEG Polymer Composite Electrolyte Films, *Ionics*, **2012**, 18 (4), 403-411.
- [24] OG Abdullah, DA Tahir and K Kadir, Optical and Structural Investigation of Synthesized PVA/PbS Nanocomposites, *Journal of Materials Science: Materials in Electronics*, **2015**, 26 (9), 6939-6944.
- [25] S Agarwal and VK Saraswat, Structural and Optical Characterization of ZnO Doped PC/PS Blend Nanocomposites, *Optical Materials*, **2015**, 42, 335-339.
- [26] OG Abdullah and SA Saleem, Effect of Copper Sulfide Nanoparticles on the Optical and Electrical Behavior of Poly (Vinyl Alcohol) Films, *Journal of Electronic Materials*, **2016**, 45 (11), 5910-5920.
- [27] PK Khare and SK Jain, Dielectric Properties of Solution-Grown-Undopedand Acrylic-Acid-Doped Ethyl Cellulose, *Bulletin of Materials Science*, **2000**, 23 (1), 17-21.
- [28] YJ Yin, KD Yao, GX Cheng and JB Ma, Properties of Polyelectrolyte Complex Films of Chitosan and Gelatin, *Polymer International*, **1999**, 48 (6), 429-432.
- [29] OG Abdullah and DR Saber, Optical Absorption of Polyvinyl Alcohol Films Doped with Nickel Chloride, *Applied Mechanics and Materials*, **2012**, 110-116, 177-182.
- [30] SZ Yusof, HJ Woo and AK Arof, Ion Dynamics in Methylcellulose–LiBOB Solid Polymer Electrolytes, *Ionics*, **2016**, 22 (11), 2113-2121.
- [31] OG Abdullah, DR Saber and LO Hamasalih, Complexion Formation in PVA/PEO/CuCl₂Solid Polymer Electrolyte, *Universal Journal of Materials Science*, **2015**, 3 (1), 1-5.
- [32] Z Osman, MIM Ghazali, L Othman and KBM Isa, AC Ionic Conductivity and DC Polarization Method of Lithium Ion Transport in PMMA–LiBF₄Gel Polymer Electrolytes, *Results in Physics*, **2012**, 2, 1-4.
- [33] OG Abdullah, DR Saber and SA Taha, The Optical Characterization of Polyvinyl Alcohol: Cobalt Nitrate Solid Polymer Electrolyte Films, *Advanced Materials Letters* **2015**, 6, 153-157.
- [34] S Rajendran, M Sivakumar and R Subadevib, Investigations on the Effect of Various Plasticizers in PVA–PMMA Solid Polymer Blend Electrolytes, *Materials Letters*, **2004**, 58 (5), 641-649.
- [35] OG Abdullah, BK Aziz and AO Saeed, Kaolin Light Concentration Effects on the Dielectric Properties of Polyvinyl Alcohol Films, *International Journal of Science and Advanced Technology*, **2012**, 2 (1), 65-70.
- [36] MH Buraidah and AK Arof, Characterization of Chitosan/PVA Blended Electrolyte Doped with NH₄I, *Journal of Non-Crystalline Solids*, **2011**, 357, 3261-3266.