

Eurasian Journal of Soil Science

Journal homepage : http://ejss.fesss.org



Soil moisture adsorption capacity and specific surface area in relation to water vapor pressure in arid and tropical soils Abdelmonem Mohamed Amer *

Department of Soil Science, Faculty of Agriculture, Menoufia University, 32511 Shebin El-Kom, Egypt

Abstract

This study is devoted to predict water vapour adsorption and hydro-physical properties of arid soils in middle Nile Delta (Farm of the Faculty of Agriculture, Shebin El-Kom, Egypt) and of tropical soils (Felix and INIAP Farms) in Quevedo zone, Los Rios, Ecuador. The vapour pressure and isothermal adsorption of water vapour is used to predict soil moisture adsorption capacity (Wa) and the specific surface area. To achieve these objectives, four soil profiles at different depths were investigated to indicate the status of hydro-physical properties of the studied area. The 1st & 2nd profiles are sandy loam (Felix Farm) and clay loam soils (Shebin El-Kom Farm), and 3rd & 4th are clay soils (INIAP Farm). Data of soilwater adsorption (W%) at different relative vapor pressures P/Po are obtained for the studied soil profiles, where the W% values increased with increasing P/Po from 1.87% to 10.01% in the 1st and 2nd sandy loam and clay loam soil profiles, and reached 27.44% in the 4^{th} clay soil profile. The highest values of water adsorption capacity (Wa) were observed in the clay depths of 60 – 90 cm and 90 – 120 cm (INIAP-soil profiles) while the lowest values were in the subsurface depth (30 – 60 cm) of soil profiles 1^{st} and 2^{nd} . The other hygrophysical properties such as adsorbed layers and maximum hygroscopic water were obtained. The specific surface area (S) in sandy loam 1st&2nd soil profiles is ranged from 113m²/g to 187m²/g and raised to 385m²/g and 553m²/g in the 3rd & 4th clay soil profiles. The corresponded values of the external specific surface area (Se) ranged from $42m^2/g$ to $98m^2/g$ and 74 m²/g to 252 m²/g respectively. Two equations were assumed (1) to predict P/Po at water adsorption capacity (Wa), and (2) to apply Wa in prediction of soil moisture retention i.e., $\psi(W)$ function at pF < 4.5.

Keywords: Water adsorption capacity, vapor pressure isotherm, soil hydro-physical properties, specific surface area, arid and tropical soils.

© 2019 Federation of Eurasian Soil Science Societies. All rights reserved

Introduction

Article Info

Received : 11.12.2018

Accepted : 18.06.2019

Agricultural and irrigation management practices, is largely depend on a timely and accurate characterization of temporal and spatial soil moisture dynamics in the root zone (Han, 2011). Consequently, measurements and detailed information about soil water sorption, water content and behavior are required. In that connection, water vapor adsorption is an important phenomenon in particular in drying periods in tropical soils as well as in arid and semi-arid regions (Amer, 2014) at which the high temperature and dry weather supports to more evapo-transpiration. Water vapor either reaches the soil from the atmosphere or is formed in the soil by the evaporation of water. The migration of water vapor in soil depends not only on the difference of vapor pressure in different sites but also on the capacity of soil particles surfaces to attract and absorb the molecules of vapor. A gain of water in the soil surface layer, not caused by rainfall or irrigation, can be caused by dew deposition or vapor adsorption. Dew deposition is a phenomenon recorded for most soil and climate types (Jacobs et al., 1999). It occurs during the night when dew point is reached, and it results in a discernible wetting of the surface.

* Corresponding author.

Department of Soil Science, Faculty of Agriculture, Menoufia University, 32511 Shebin El-Kom, Egypt

Tel.: +2050 2337480

Under the right atmospheric and soil surface conditions, water is adsorbed from the atmosphere by a thin layer of top soil, generally during the afternoon and evening. The amounts of adsorbed water can be considerable up to 70% of daily evaporation (Kosmas et al. 1998, 2001; Agam and Berliner, 2004). Therefore, quantifying adsorption is important for agricultural water management, surface energy balance studies, ecological studies (Levy and Mamedov, 2002), and remote sensing investigations (changes in surface soil moisture content will affect land surface properties such as albedo, emissivity, and thermal inertia).

The objective of this work is devoted to study the isothermal adsorption of water vapour at different vapour pressures as applied to predict soil-water adsorption capacity, specific surface area, and hygro-physical properties in semiarid soils in the Nile Delta (Egypt) and in the tropical soils of Quevedo region-Los Rios Provence, Ecuador.

Material and Methods

Four soil profiles differ in their particles size distribution, salinity, and CaCO₃% were done in arid and tropical zones. Three tropical soil profiles (I, III, IV) were selected at different distances to Quevedo city, to represent the Quevedo region of the Los Rios Province, Ecuador. The investigated soils were cultivated mainly with Cacao, Banana and Corn crops. The arid soil profile No. II was taken from the Farm of Faculty of Agriculture, Menoufia University, Shebin El-Kom, (located at the Middle of the Nile Delta), Egypt. The profiles of Quevedo area is elevated 74m (243ft) above sea level and located at coordinates 1°20'30" de latitude south and 79°28'30" de longitude occidental, dentro de una zona subtropical (Figure 1). The first profile (I) was taken from the Felix Farm, which is located as far as 30 km from Quevedo city and cultivated with Cacao trees. The third and fourth profiles were taken from the INIAP Experimental Farms, Pichilingue, which located at 6 km to Quevedo. They were cultivated with Cacao and Corn crops respectively.

Disturbed and undisturbed soil samples were taken at depths, 0 - 30 cm, 30-60 cm, 60 - 90 cm for the first and second profiles, and 0 - 30 cm, 30 - 60 cm, 60 - 90 cm, 90 - 110 cm and 110 - 130 cm for the third profile and 0 - 30 cm, 30 - 60 cm, 60-90 cm, 90 - 120 cm for the fourth one. The disturbed soil samples were air dried, gently crushed and sieved through a 2 mm sieve. Fractions below 2 mm were subjected to chemical and mechanical analysis in the laboratory. Soil texture (particle size distribution) was determined using the pipette method. The textural grade was assessed by texture triangle.



Figure1. Study areas (red color) in Los Rios province, Quevedo, Ecuador and in Nile Delta, Shebin El-Kom, Menoufia, Egypt

Physical and chemical analyses of the studied soils such as maximum hygroscopic water (*MH*), Particle size distribution %, EC, OM, $CaCO_3\%$ are determined according to Black et al. (1965) and Klute (1986). These analyses are shown in Tables 1 and 2.

Table 1. Particle size distribution of the studied soils

Profile Number & Soil location	Coarse Sand, %	Fine Sand, %	Silt, %	Clay, %	Texture Class
I- Felix Farm	38.50	20.50	24.10	16.90	Sandy Loam
II- Shebin El-Kom	15.76	18.35	27.32	38.57	Clay Loam
III-INIAP-Cacao Field	1.98	16.42	36.78	44.82	Clay
IV-INIAP-Corn Field	3.15	14.40	33.86	48.59	Clay

Chemical propertties	Felix Farm	Shebin El-Kom Farm	INIAP-Cacao Field	INIAP-Corn Field
рН	6.82	6.85	7.24	6.91
EC (dS/m)	0.73	1.24	1.10	1.96
Ca ²⁺	1.35	2.67	4.80	6.24
Mg^{2+}	1.40	2.23	1.92	2.88
Na+	1.42	3.54	3.98	9.24
K+	0.13	0.39	0.48	1.48
CO ₃ ²⁻	-	-	-	-
HCO ₃ -	3.40	3.93	2.25	2.85
Cl-	0.75	3.65	6.00	13.52
SO4 ²⁻	0.15	1.25	1.93	2.47
SAR	1.21	2.26	2.17	4.32
CaCO ₃ (%)	1.12	3.42	0.43	0.34
OM (%)	1.38	2.21	2.72	2.99

Table 2. Chemical analysis	(mean values)) of the studied soil profiles.
	()

Water vapour adsorption isotherms

The water vapour adsorption isotherm on dried soils is determined gravimetrically using saturated salt solutions such as $ZnCl_2$, $CaCl_2$, K_2CO_3 , NH_4NO_3 , KCl and K_2SO_4 whereas the corresponding P/Po values to these solutions at 20°C are 0.10, 0.35, 0.45, 0.65, 0.85 and 0.98 (Amer, 2009).

Matric suction at water vapour pressures

The soil matric suction values resulting from equilibrating the soil samples with salt solutions can be calculated using the pF formula:

$$pF = 6.502 + \log \left[2 - \log H\right]$$
(1)

where, pF = soil suction, expressed as the common logarithm of the suction (ψ) in cm of water, H is the relative humidity (H= $P/Po \ge 100$), and Relative water vapour pressure (P/Po) [P being the actual water vapour pressure on the sample particles and Po being the saturation vapour pressure of water at 20°C] were obtained by applying different appropriate salt solutions.

Moisture adsorption capacity

The property of moisture adsorption capacity (*Wa*) can be introduced as the critical limit between adsorbed and absorbed wetting films (pellicles) of soil moisture content (Amer, 2003), as well as corresponds to capillary condensation. So, the *Wa* values can be also derived from soil moisture retention curve at log (ψc) where ψc is the capillary condensation attitude.

Amer (1982, 1993) proved that the moisture adsorption capacity (Wa) is equal to three layers of adsorbed water as follows: Wa = Wm + 2Wme, where Wm is the moisture of the soil when water vapour adsorbed for monolayer, and Wme is the external mono-adsorbed layer of soil moisture content.

However, the moisture adsorption capacity (Wa), maximum hygroscopic water (MH) the specific surface area (S), and particle size composition are the most important indices characterizing the hydro-physical, physicochemical and heat properties of soil. Moreover, these parameters are inter-related, therefore the value of any of them can be obtained from the data of the other parameters.

Estimating Wm and Wme for predicting the moisture adsorption capacity and surface area

The relation between relative vapour pressure (P/Po) and moisture content (W%) is experimental obtained by maintaining a soil sample in isothermal equilibrium with an atmosphere of water vapour as mentioned above.

The BET method (Brunauer et al., 1938) as modified and described by Orchiston (1954), Quirk (1955), Farrar (1963) and Globus (1996), was applied to predict the *Wm* and *Wme*, which they in turn were used to determine moisture adsorption capacity (*Wa*) and the total, external and internal specific surface areas (*S*, *Se* and *Si*).

Results and Discussion

BET theory as applied for adsorption isotherms

Brunauer et al. (1938), derived what has come to be known as the BET equation, based on multilayer adsorption theory. In BET theory, the explanation proposed for sigmoid type isotherm is that the adsorption in multi-molecular layers on the surface rather than a monomolecular one. Farrar (1963) and Amer (1982, 1993, 2009) used the water vapour adsorption isotherm method by applying BET theory based on the

assumption that the isotherm is made up of monolayer physical adsorption combined with capillary condensation as follows:

$$P/V(P_o-P) = (1/V_m C) + (C-1) P/V_m C P_o$$
⁽²⁾

Where, *V* is the volume of gas adsorbed at pressure *P*, V_m is the volume of a single layer of adsorbed molecules over the entire surface of the adsorbent (soil particles). P_o is the gas pressure required for monolayer saturation at the temperature of the experiment, and C is a constant for the particular gas, adsorbent, and temperature,

$$C = \exp E_1 - E_L / RT \tag{3}$$

whereas, E_1 is adsorption heat of the water adsorbed layer, E_L is condensation adsorption heat.

In this work we obtained experimentally and gravimetrically the relationship between relative vapour pressure (*P*/*Po*) and adsorbed moisture content, W% (wetting films) on soil particles which depending on the thermodynamic of adsorption of water vapor through the soil. Data for water vapor adsorption at *P*/*Po*; 0.10, 0.35, 0.45, 0.65, 0.85, and 0.98 for the studied soils are presented in Table 3. It is evident that the W% at different values of *P*/*Po* was higher in both clay soil profiles (III & IV) of INIAP Farms –in particular in the deeper depths (60-90 cm and 90-120 cm) of the 4th soil profile- than in the other two sandy loam and clay loam soils (1st & 2nd profiles).

Soil location &		Soil	P/Po	0.10	0.35	0.45	0.65	0.85	0.98 (MH)
Profile number		Depth, cm	Soil water content,%			Soil v	water adso	rption	
		0 - 30	10.285	1.918	4.469	5.020	5.489	5.590	7.640
Felix Farm	Ι	30 - 60	8.359	1.874	4.241	4.790	5.031	5.372	6.955
		60 - 90	20.540	2.679	6.527	7.068	7.592	8.061	10.01
Shebin El-Kom		0 - 30	19.120	2.689	6.423	6.980	7.322	7.873	9.878
	II	30 -60	11.855	1.748	4.387	4.927	5.386	5.520	7.267
		60 - 90	15.135	1.887	4.844	5.388	5.656	6.165	8.012
		0 - 30	11.579	2.663	6.613	7.160	7.722	8.232	10.130
INIAP-Cacao		30 - 60	14.731	2.564	9.332	9.789	10.867	12.259	13.959
Field	III	60 - 90	30.252	2.974	10.470	10.920	11.458	13.604	15.521
		90 - 110	23.616	2.747	8.865	9.410	10.075	11.050	12.988
		110 - 130	17.063	3.864	10.044	10.587	11.532	12.611	14.720
		0 - 30	9.176	2.676	6.364	6.916	7.477	8.104	10.233
INIAP-Corn	IV	30 - 60	9.180	2.675	10.325	10.797	11.920	13.640	15.702
Field		60 - 90	17.198	3.879	15.504	16.050	18.131	20.005	22.647
		90 - 120	20.434	3.981	17.945	18.486	22.265	24.122	27.446

Table 3. Water adsorption (*W%*) in the studied soils at different vapor pressures (*P*/*Po*).

The W% values increased with increasing P/Po from 1.87% to 10.01% in the 1st soil profile of sandy loam soil, and from 1.75% to 9.88% in the 2nd clay loam profile. In clay soils, the increasing of W% values was from 2.56% to 15.52% in the 3rd soil profile, while it was more evident in the 4th profile where the increasing was from 2.67% to 27.44%. The clay content, mineralogical composition and salinity are the major factors that governed the absorbed water in the soils under investigation (Amer, 2009).

Solution of the BET equation for obtaining mono-adsorbed layers (Wm & Wme)

The BET equation can be applied in the following form using the gravimetric of a single layer of adsorbed molecules over the entire surface of the soil particles (Amer, 2009):

$$\frac{P}{W(Po-P)} = \frac{1}{WmC} + \frac{C-1}{WmC} \cdot \frac{P}{Po}$$
(4)

where *W*m is the moisture content when the soil surface is completely covered by a mono-molecular layer of water. *C* is a function of the state of the first adsorbed molecular layer of water and soil particles surface condition. *W* is adsorbed soil moisture content (%) equilibrated with *P*/*Po*, whereas P and *Po* are the actual and saturated water vapor pressures. By plotting *P*/*W* (*Po*–*P*) as ordinate versus *P*/*Po* at the segment 0 – 0.40 of the sorption isotherm as abscissa, a straight line would be obtained. The intercept on the y-axis is then $1/W_mC$ and the slope is $C - 1/W_mC$. Hence W_m and C can be determined.

Data for solution of BET equation are presented in Tables (4 and 5). From these data and linear equations, we can obtain W% at any P/Po values by using the next formula:

$$W = \frac{\frac{P}{Po}}{\left[1 - \frac{P}{Po}\right]\left[\frac{1}{WmC} + \frac{C - 1}{WmC} \cdot \frac{P}{Po}\right]}$$
(5)

To determine W_{me} the BET equation (4) can be developed with some assumptions to the next form:

$$W = \frac{Wme}{\left[1 - Ke\frac{P}{Po}\right]} \cdot \frac{Ce}{Ce + \frac{Po}{P} - Ke} + Wi$$
(6)

At high relative water vapor pressures, the amount Ce / (Ce + Po/P - Ke) is equal unit (Farrar, 1963), and then the equation (6) becomes (Amer, 2015):

$$W = \frac{Wme}{\left[1 - Ke\frac{P}{Po}\right]} + Wi \tag{7}$$

where the suffixes (e) and (i) refer to the external and internal surfaces respectively. The values of K_e in the indicated P/Po range were stated by Farrar (1963) as 0.9 ± 0.01 , but practically, it seems that K_e is an arbitrary coefficient ranged from 0.70 to 0.90 (Amer 1982, 2009). Equation (7) can be represented in linear equation; y = mx + c where y = W, $m = W_{me}$, $x = 1 / 1 - K_e P/Po$, and c = Wi, so W_{me} can be obtained graphically as the intercept on the y – axis.

Values of *Wm* and *Wme* are calculated for the studied soils as in Table 6. It is found that the highest values of *Wm* and *Wme* were in the deeper depths of the clay soils (INIAP farms) in particular the depths of 60 – 90 cm and 90 – 120 cm of INIAP crop field (profile IV).

Soll location	Soll Depth, cm	Р/Ро	0.10	0.20	0.30	0.35	С	Linear & adsorption equations
		W%	1.920	3.200	4.050	4.470		P/W(Po-P)=0.036+0.233P/Po
	0 - 30	W(1-P/Po)	1.728	2.560	2.835	2.905	7.472	$W\% = \frac{\frac{1}{Po}}{\frac{1}{Po}}$
		P/W(Po-P)	0.058	0.078	0.106	0.120		$\left[1 - \frac{p}{Po}\right] [0.036 + 0.233 \frac{p}{Po}]$
		W%	1.874	3.000	3.800	4.241		P/W(Po-P)=0.03+0.277P/Po
i Felix	30 -60	W(1-P/Po)	1.687	2.400	2.660	2.758	10.238	$W\% = \frac{\frac{P}{Po}}{\frac{P}{Po}}$
Farm		P/W(Po-P)	0.059	0.083	0.113	0.127		$\left[1 - \frac{P}{Po}\right] \left[0.03 + 0.277 \frac{P}{Po}\right]$
-		W%	2.679	4.500	5.800	6.527		P/W(Po-P)=0.023+0.17P/Po
	60 - 90	W(1-P/Po)	2.411	3.600	4.060	4.243	8.391	$W\% = \frac{\frac{1}{Po}}{\frac{1}{Po}}$
		P/W(Po-P)	0.042	0.056	0.074	0.083		$\left[1 - \frac{p}{Po}\right] \left[0.023 + 0.17 \frac{p}{Po}\right]$
		W%	2.689	4.480	5.750	6.423		P/W(Po-P)=0.019+0.1814P/Po
	0 - 30	W(1-P/Po)	2.420	3.584	4.025	4.175	10.548	$W\% = \frac{\frac{1}{Po}}{\frac{1}{Po}}$
		P/W(Po-P)	0.037	0.055	0.074	0.083		$\left[1 - \frac{p}{Po}\right] [0.019 + 0.1814 \frac{p}{Po}]$
		W%	1.748	2.650	3.300	4.387		P/W(Po-P)=0.024+0.2942P/Po
n Shebin	30 - 60	W(1-P/Po)	1.573	2.120	2.310	2.851	13.258	$W\% = \frac{\frac{1}{Po}}{\frac{1}{Po}}$
El-Kom -		P/W(Po-P)	0.064	0.094	0.129	0.126		$\left[1 - \frac{p}{Po}\right] [0.024 + 0.2942 \frac{p}{Po}]$
		W%	1.887	3.400	4.560	4.844		P/W(Po-P)=0.03+0.2771P/Po
	60 - 90	W(1-P/Po)	1.698	2.720	3.192	3.149	11.324	$W\% = \frac{\frac{r}{Po}}{\frac{1}{Po}}$
		P/W(Po-P)	0.059	0.074	0.940	0.111		$\left[1 - \frac{P}{Po}\right] [0.03 + 0.277 \frac{P}{Po}]$

Table 4. Using Equation (5) in solution of the BET equation for sandy loam and clay loam soil profiles (I and II)

Soil specific surface area

0.17

The specific surface of the adsorbent (soil) can be calculated by determining the number of molecules (volumetrically or gravimetrically) and multiplying this by the cross- sectional area of the molecules. Assuming that a single water molecule occupies some constant area on the sorbent surface (usually taken as 10.8 °A²), the total specific surface area (*S*) of the soil then calculated as *S* = 36.16 W_m m²/g.

Soil location	Soil Depth, cm	P/Po	0.10	0.20	0.30	0.35	С	Linear & adsorption equations
		W%	2.663	4.380	5.750	6.613		P/W(Po-P)=0.025+0.1611P/Po
	0 - 30	W(1-P/Po)	2.397	3.504	4.025	4.298	7.446	$W\% = \frac{\frac{r}{Po}}{\frac{r}{Po}}$
		P/W(Po-P)	0.042	0.057	0.075	0.081		$\frac{1}{1 - \frac{P}{Po}} [0.025 + 0.1611 \frac{P}{Po}]$
		W%	2.564	5.500	8.120	9.330		P/W(Po-P)=0.03+0.0791P/Po
INIAP-	30 -60	W(1-P/Po)	2.307	4.400	5.684	6.065	3.638	$W_{06} = \frac{P}{Po}$
Cacao Field		P/W(Po-P)	0.043	0.046	0.053	0.058		$W_{70} = \frac{P}{\left[1 - \frac{P}{Po}\right]\left[0.03 + 0.0791\frac{P}{Po}\right]}$
-		W%	2.974	6.740	9.200	10.470		P/W(Po-P)=0.02+0.0897P/Po
	60 - 90	W(1-P/Po)	2.677	5.392	6.440	6.806	5.486	$W_{06} = \frac{P}{Po}$
		P/W(Po-P)	0.037	0.037	0.047	0.051		$1 - \frac{P}{Po} \left[1 - \frac{P}{Po} \right] \left[0.02 + 0.0897 \frac{P}{Po} \right]$
-		W%	2.747	5.000	6.600	8.865		P/W(Po-P)=0.028+0.1157P/Po
	90 - 110	W(1-P/Po)	2.472	4.000	4.620	5.762	5.060	$\frac{P}{Po}$
		P/W(Po-P)	0.040	0.050	0.064	0.061		$\left[1 - \frac{P}{Po}\right] [0.028 + 0.1157 \frac{P}{Po}]$
-		W%	3.864	6.200	8.510	10.044		P/W(Po-P)=0.032+0.0617P/Po
	110 - 130	W(1-P/Po)	3.478	4.960	5.957	6.529	2.929	$W\% = \frac{\frac{1}{Po}}{\frac{1}{Po}}$
		P/W(Po-P)	0.029	0.403	0.050	0.054		$\left[1 - \frac{p}{Po}\right] [0.032 + 0.0617 \frac{p}{Po}]$
		W%	2.680	4.120	5.500	6.364		P/W(Po-P)=0.022+0.1864P/Po
	0 - 30	W(1-P/Po)	2.408	3.296	3.850	4.137	5.937	$W\% = \frac{1}{Po}$
-		P/W(Po-P)	0.042	0.061	0.078	0.085		$\left[1 - \frac{P}{Po}\right] \left[0.022 + 0.1864 \frac{P}{Po}\right]$
W		W%	2.675	5.880	8.750	10.325		P/W(Po-P)=0.029+0.0655P/Po
INIAP-	30 - 60	W(1-P/Po)	2.408	4.704	6.125	6.711	3.245	$W\% = \frac{\overline{Po}}{\overline{Po}}$
Corn Field		P/W(Po-P)	0.042	0.043	0.049	0.052		$\left[1 - \frac{P}{Po}\right] [0.029 + 0.0655 \frac{P}{Po}]$
		W%	3.879	10.000	13.800	15.504		P/W(Po-P)=0.012+0.0652P/Po
	60 - 90	W(1-P/Po)	3.491	8.000	9.660	10.078	6.481	$W\% = \frac{\frac{r}{Po}}{\frac{r}{Po}}$
		P/W(Po-P)	0.029	0.025	0.031	0.035		$\left[1 - \frac{P}{Po}\right] [0.012 + 0.0652 \frac{P}{Po}]$
-	00 120	W%	3.981	11.500	16.320	17.945		P/W(Po-P)=0.011+0.0543P/Po
	90 - 120	W(1-P/Po)	3.583	9.200	11.424	11.664	5.937	$W_{0/2} = \frac{\frac{P}{Po}}{\frac{P}{Po}}$
		P/W(Po-P)	0.028	0.022	0.026	0.030		$\left[1 - \frac{P}{Po}\right] \left[0.011 + 0.0543 \frac{P}{Po}\right]$

Table 5. Using Equation (5) in solution of the BET equation for clay tropical soil profiles (III and IV).

Data in Table 6, which based upon the water vapor adsorption isotherms show that the specific surface area (*S*) in the light textured soils ranged from 117 m²/g to 187 m²/g in the first profile and from 113 m²/g to 180 m²/g in the second profile. In clay tropical soils, the S values were from194 m²/g to $385m^2/g$ in the third profile and from 173 m²/g to $553 m^2/g$ in the fourth profile. Farrar (1963) deduced the soil external specific surface area (*Se*) by applying the relation (6) over the range of high relative vapor pressure (0.5 < P/Po > 0.8). However, the internal specific surface area (*Si*) may be calculated by the difference between *S* and *Se*.

The corresponding values of *Se* (Table 6) were 46 m²/g to 98 m²/g and 42 m²/g to 93 m²/g in the 1st and 2nd soil profiles and from 77m²/g to 165m²/g and 74 m²/g to 252 m²/g in clay soil profiles. Regarding the internal specific surface area (*Si*), it was found in loam soils that *Si* values are higher than *Se* in the surface (0 – 30 cm) and subsurface (30 – 60 cm) depths of the 1st profile (Felix farm), while the same observation was in the depths 30 – 60cm and 60 – 90cm of the 2nd profile (dry soil). In clay soils (INIAP farms) the *Si* values were higher than *Se* in all depths of the 3rd and 4th profiles. The results reflect different due to the difference of the investigated soils in their texture, clay content%, CaCO₃, salinity and mineralogical composition (El-Sharkawy, 1994; El-Fiky, 2002). However, the specific surface area is closely related to the physicochemical soil properties (Nerpin and Chudnovski, 1975), which is refer to the absence or presence of internal pores.

Whatever, the *Se/S*i ratio was less than unit in most depths of sandy loam soils and in all depths of the clay soils indicating that the *S*i values were higher than *S*e in most investigated tropical soils (Table 6). This result may refer to the mineralogical composition of these soils (Figure 2), whereas the montmorillonite, ferrous and hydrous mica are the prevailing minerals in the tropical soils as well as in the clay alluvial arid soils of the Nile Delta (El-Gabaly and Khadr, 1962).

Table 6. Water adsorption capacity (*Wa*), adsorbed layers (*Wm* & *Wme*) and specific surface area (*S*, *Se* & *Si*) of the studied soils.

Soil	Soil	Wm	Wme	Wc	Wa	[P/Po] _{wa}	<i>S,m</i> ² /g	<i>Se,</i> m ² /g	Si,	Se/Si	Wa/S
location	Depth, cm								m ² / g		
Ι	0 - 30	3.717	1.624	5.341	6.965	0.4662	134.430	58.72	75.710	0.775	0.0518
Felix	30 -60	3.256	1.274	4.530	5.804	0.4388	117.730	46.06	71.67 0	0.642	0.0493
Farm	60 - 90	5.181	2.735	7.916	10.651	0.5135	187.350	98.89	88.460	1.117	0.0568
II	0 - 30	4.989	2.581	7.570	10.151	0.5084	180.41	93.32	87.090	1.071	0.0562
Shebin	30 -60	3.143	1.188	4.331	5.519	0.4306	113.64	42.95	70.690	0.607	0.0485
El-Kom	60 - 90	3.222	1.248	4.470	5.718	0.4305	116.50	45.12	71.380	0.632	0.0491
III	0 - 30	5.372	2.148	7.520	9.668	0.4442	194.25	77.67	116.58	0.666	0.0497
INIAP-	30 -60	9.160	3.964	13.124	17.088	0.4636	331.30	143.34	187.960	0.762	0.0515
Cacao	60 - 90	9.115	4.240	13.355	17.595	0.4819	329.58	153.32	176.260	0.869	0.0533
Field	90 - 110	6.934	2.773	9.707	12.480	0.4424	250.74	100.27	150.470	0.666	0.0498
	110 - 130	10.670	4.568	15.238	19.806	0.4612	385.85	165.18	220.670	0.748	0.0513
IV	0 - 30	4.798	2.067	6.865	8.932	0.4628	173.51	74.74	98.770	0.756	0.0515
INIAP-	30 -60	10.550	4.542	15.092	19.634	0.4610	381.54	164.23	217.310	0.755	0.0514
Corn	60 - 90	12.965	5.837	18.802	24.639	0.4743	468.82	211.06	257.760	0.818	0.0525
Field	90 - 120	15.310	6.981	22.291	29.272	0.4768	553.63	252.43	301.190	0.838	0.0529





Figure 2. Clay soil samples were taken at depth >90 cm from INIAP research station- Pichilingue, Ecuador ~containing ferrous and ferric minerals with the highest values of surface area and water adsorption capacity (Amer, 2015).

Soil moisture adsorption capacity

Due to the soil – water adsorption capacity (*Wa*) is obeyed the mono-adsorbed layers (*Wm & Wme*) values (whereas, Wa = Wm + 2Wme), so it could be correlated with specific surface area. The *Wa/S* ratio was 0.0486 – 0.0568 for all the studied soils (Table 6). The values of *Wa* are the highest in the same deep depths, whereas *Wa* values in INIAP-profile IV reached 24.64% and 29.27% in the depths of 60 – 90 cm and 90 – 120 cm respectively, while reached 19.81% in the depth of 110 – 130 cm in the INIAP- profile III (Table 6). The high values of soil water adsorption capacity in the studied clay soils showed the importance of water adsorption phenomena in tropical soils.

The *Wa* values in the surface depth (0-30 cm) ranged from 6.97% to 10.15% in both sandy loam and clay loam soils under investigation. The lowest values for *Wa* were observed in the subsurface depth (30 – 60 cm) in soils profiles I and II, where, the *Wa* values were 5.80% and 5.52%, but increased to 17.09% and 19.63 in subsurface depth (30-60 cm) of the clay soil profiles III and IV respectively. This is indicated the significance of the soil texture and clay fraction content which play an important role in soil moisture content and its distribution along the soil profile depth. In general, high clay content in soil means increasing the soil moisture content & retention and water adsorption capacity. The same trend was ordinary observed with the other hygroscopic parameters such as boundary moisture films (*Wc*) and maximum hygroscopic water (*MH*). Hygroscopic water exists as a very thin film at the solid-liquid interfaces of the soil particles. At the maximum hygroscopic water (*MH*) the surface of soil particles is almost completely covered with individual molecules of water. However, It is known that the maximum hygroscopic water (*MH*) is determined practically at P/Po = 0.98. So, data in Table 3 indicate that the *MH* values (at *P/Po* = 0.98) are ranged from 6.95% –10.01% in soil profiles I& II, and from 10.13% – 27.44%, for the clay soils (profiles III &

IV). It is clear that the hygro-physical properties (*Wm, Wme, Wc, W*a and *MH*) of the investigated soils are followed the same trend that observed in turn with the specific surface areas (*S* and *Se*).

Predict of P/Po at water adsorption capacity (Wa)

At higher water vapor pressure (*P*), the equation (4) can be developed to predict the vapor pressure *P*/*Po* at water adsorption capacity *Wa* as follows:

$$\frac{\frac{P}{Po}}{w(1-\frac{P}{Po})} = A + B\frac{P}{Po}$$
(8)

Where, A = 1/WmC, and B = C-1/WmCFrom equation (8) and at $Po/P \approx 1$;

$$W(1 - P/Po) = 1/(A + B)$$
 (9)

Then at Wa;

$$[P/Po]_{wa} = 1 - [Wa (A + B)]^{-1}$$
(10)

Data in table (6) show that the relative vapor pressure $[P/Po]_{wa}$ at *Wa* is ranged between 0.43 to 0.51, indicating that at this range of P/Po the soil moisture reach water *adsorption* capacity. Above this range (P/Po > 0.51) the *absorption* process is prevailing, at which soil matric suction (ψ) values can be calculated using equation (1); pF = 6.502+log [2 – log P/Po]. The ψ values expressed in pF at adsorption and absorption processes are ranged from 4.5 to 7.0. On the other hand, it may of interest to apply the water adsorption capacity (*Wa*) to predict soil moisture retention function $\psi(W)$ at pF < 4.5 using the following suggested equation:

$$\Psi i = \psi a \left(W i - W c / W m e \right)^{-n}$$
(11)

where ψ i and ψ a are capillary–sorption potentials at soil water content (*W*i) and moisture adsorption capacity (*W*a), respectively, and *n* is a constant. The decrease in soil water suction is associated with increasing thickness of the hydration envelopes covering the soil particles surfaces (Amer, 2009).

Conclusion

Soil-water adsorption W% values increased with increasing P/Po from 1.87% to 10.01% in the 1st and 2nd sandy loam and clay loam soil profiles, and reached 27.44% in the 4th clay tropical soil profile. The values of mono adsorbed layers (*Wm & Wme*), boundary moisture films (*Wc*), maximum hygroscopic water (*MH*), water adsorption capacity (*Wa*) and specific surface area (*S*), external (*Se*), internal (*Si*) of the tropical soils in the Quevedo region area (Ecuador) and in semiarid region of the Nile Delta (Egypt) are obtained experimentally.

Water adsorption capacity (Wa) is corresponding to P/Po = 0.43 – 0.51 for all soils under investigation. The absorption process is prevailing above this range.

The highest values of water adsorption capacity (*Wa*) were observed in the clay depths of 60 - 90 cm and 90 - 120 cm (INIAP-soil profiles) while the lowest values were in the subsurface depth (30 - 60 cm) of soil profiles 1st and 2nd. The specific surface area (*S*) in sandy loam & clay loam (1st & 2nd tropical and arid soil profiles) is ranged from $113m^2/g$ to $187m^2/g$ and raised to $385m^2/g$ and $553m^2/g$ in the 3rd & 4th clay tropical soil profiles. The corresponded values of the external specific surface area (*Se*) ranged from $42m^2/g$ to $98m^2/g$ and $74m^2/g$ to $252m^2/g$ respectively.

The internal specific surface area (Si) values were higher than the external specific surface area (Se) in all depths of the clay soil profiles, while they were higher than Se in sandy loam and clay loam soils only in the subsurface (30 - 60 cm) depth.

The results of specific surface area (*S, Se, Si*), mono adsorbed layers (*Wm & Wme*), water adsorption capacity (*Wa*) reflect different due to the texture and mineralogical composition of the investigated soils.

Clay soils at depth >90 cm in the farms of INIAP research station-Pichilingue, are containing ferrous and ferric minerals with the highest surface area and water adsorption capacity.

Results of soil water adsorption capacity show the significance of water adsorption capacity for moisture plant root zone – in particular- in clay soils.

Two new equations were assumed (1) to predict P/Po at water adsorption capacity (*Wa*), and (2) to predict the soil moisture retention function $\psi(W)$ at pF < 4.5, depending on the water adsorption capacity *Wa*.

Acknowledgement

I am pleased to have the Prometheus Scholarship of the SENESCYT Prometeo Project of Ecuador during 2013/2014. The laboratorial support of SENESCYT for this work is acknowledged.

References

- Agam, N., Berliner, P.R., 2004. Diurnal water content changes in the bare soil of a coastal desert. *Journal of Hydrometeorology* 5: 922–933.
- Amer, A.M., 1982. Effect of the overburden pressure on the capillary sorption potential of water in swelling soils. PhD Thesis, Faculty of Soil Science, Lomosonov Moscow State University, Moscow, Russia. [in Russian].
- Amer, A.M., 1993. Surface area measurements as related to water vapour adsorption in arid soils of Egypt. Proceedings of the IV International Conference on Desert Development. 25-30 July, Mexico City, Mexico. pp. 619–627.
- Amer, A.M., 2003. Soil hydro-physics. First Part, Al-Dar Al-Arabia for Publishing Cairo, Egypt. 452p. [in Arabic].
- Amer, A.M., 2009. Moisture adsorption capacity and surface area as deduced from vapour pressure isotherms in relation to hygroscopic water of soils. *Biologia* 64(3): 516-521.
- Amer, A.M., 2014. Moisture dynamics and available water capacity in root zone as influenced by swelling pressure and water table in tropical soils. Final Report, submitted to SENESCYT, Prometeo Project, Ecuador.
- Amer, A.M., 2015. Vapor adsorption capacity and soil wetting. In: Wetting and wettability. Aliofkhazraei, M. (Ed.). IntechOpen, pp.1-14.
- Black, G.A., Evans, D.D., White, J.L., Ensminger, L.E., Clerk, F.E., 1965. Methods of soil analysis. Part 1&2. American Society of Agronomy Soil Science Society of America, Madison, Wisconsin, USA.
- Brunauer, S., Emmett, P.H., Teller, E., 1938. Adsorption of gases in multi-molecular layers. *Journal of the American Chemical Society* 60(2): 309-319.
- El-Fiky, Y.S., 2002. Studies on hydro-physical and physicochemical properties of new reclaimed soils in Egypt. PhD Thesis, Soil Science Department, Faculty of Agriculture, Menoufia University, Egypt.
- El-Gabaly, M.M., Khadr, M., 1962. Clay mineral studies of some Egyptian Desert and Nile alluvial soils. *European Journal* of Soil Science 13(2): 333–342.
- El-Sharkawy, A.F., 1994. Study of water imbibition in some agricultural soils. MSc Thesis, Agricultural Engineering Department, Faculty of Agriculture, Menoufia University, Egypt.
- Farrar, D.M. 1963. The use of vapour-pressure and moisture content measurements to deduce the internal and external surface area of soil particles. *Europen Journal Soil Science* 14(2): 303–321.
- Globus, A.M., 1996. On specific soil surface area computing by one point on the water vapor sorption isotherm. *Eurasian Soil Science* 28(10): 154–155.
- Han, E., 2011. Soil moisture data assimilation at multiple scales and estimation of representative field scale soil moisture characteristics. PhD. Thesis, Purdue University, West Lafayette, USA.
- Jacobs, A.F.G., Heusinkveld, B.G., Berkowicz, S.M., 1999. Dew deposition and drying in a desert system: a simple simulation model. *Journal of Arid Environments* 42(3): 211–222.
- Klute, A. 1986. Methods of soil analysis. Part I Physical and Mineralogical Methods. American Society of Agronomy -Soil Science Society of America, Madison, Wisconsin, USA.
- Kosmas, C., Danalatos, N.G., Poesen, J., van Wesemael, B., 1998. The effect of water vapour adsorption on soil moisture content under Mediterranean climatic conditions. *Agricultural Water Management* 36(2): 157–168.
- Kosmas, C., Marathianou, M., Gerontidis, St., Detsis, V., Tsara, M., Poesen, J., 2001. Parameters affecting water vapor adsorption by the soil under semi-arid climatic conditions. *Agricultural Water Management* 48(1): 61–78.
- Levy, G.J., Mamedov, A.I., 2002. High-energy-moisture-characteristic aggregate stability as a predictor for seal formation. *Science Society of America Journal* 66(5): 1603-1609.
- Nerpin, V., Chudnovski, A.F., 1975. Energy and mass-transfer in plant-soil-air system. Hydro-Meteo Izdat., Leningrad, Russia. [in Russian].
- Orchiston, H., 1954. Adsorption of water vapour: II. Clays at 25°C. Soil Science 78(6): 463–479.
- Quirk, J.P., 1955. Significance of surface areas calculated from water vapour sorption isotherms by use of the BET equation. *Soil Science* 80(6): 423–430.