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PREPARATION OF AN ALTERNATE SOLID FUEL FROM ALCOHOL DISTILLERY STILLAGE

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The article describes preparation of an alternative solid fuel from corn alcohol distillery stillage, which is the waste of the alcohol industry. We determined the calorific values of the dried alcohol distillery stillage and experimental briquettes, which were made of it by pressing at high pressure and temperature. The average value of the higher calorific value of the dried alcohol distillery stillage is 19545 kJ kg⁻¹, while this value for the produced briquettes is in the range of 22445 to 26594 kJ kg⁻¹. The established calorific values exceed those of analogues, which are widely used for the producing of alternative solid fuels (miscanthus and energy willow). Solid fuel prepared from corn alcohol distillery stillage can be used both in a solid form and in a dried one. The proposed method for preparation of the solid fuel allows providing additional utilization of alcohol distillery stillage and rational use of secondary raw materials of alcohol production.

Keywords: alcohol distillery stillage, secondary raw materials, solid fuel, briquette, calorific value.

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Introduction

The alcohol distillery stillage of grain origin is a by-product of the food industry, which is formed in large quantities in the production of ethyl alcohol. Secondary use of alcohol distillery stillage is difficult due to its short storage time [1].

To increase the storage time, the alcohol distillery stillage requires a drying stage [2], for example by the filtration method, which has previously shown its effectiveness in reducing the moisture content of dispersed raw materials [1,3,4].

An alternative and extremely promising method of using secondary raw materials with natural fibers is the production of solid fuel briquettes [5]. Fuel pellets are an environmentally friendly source of energy used in municipal energy for heating residential buildings and technical premises. The energy required to produce pellets is about 20% of all energy production consumption [6].

The main materials for the production of solid fuel briquettes are fast-growing trees and perennial grasses. The advantages of this group of plants are their high growth rate, which allows producing large volumes of raw materials and does not require large areas of cultivation. Such plants include energy willow, miscanthus and switch-grass [3]. It is also becoming common to use crop wastes (usually sunflower and corn) to produce pellets [6].

The process of converting vegetable raw materials into solid fuel briquettes is characterized by the initial compaction of the material by reducing the free volume available in the raw material. As a result of increasing the pressure of the pressing process, there are not only elastic but also plastic deformations. This leads to an increase in the forces of intermolecular interaction between the particles, and, consequently, improves the adhesion of the material and increases the strength. The pressing process is carried out at a pressure of 100–200 MN/m² for successful formation of solid fuel briquettes [7].

Briquetting of vegetable raw materials can be carried out with the use of a binder material or without it. In the case of its use, the density of the obtained samples will be higher, which requires more energy to form a briquette. In the case of briquetting without the additional use of a binder material, the binding of vegetable raw materials is due to the lignin present in the material itself. At temperatures close to 170–200°C and high pressure [5], lignin turns into a liquid state and is distributed throughout the

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volume of raw materials. After some time, the briquette leaves the briquetting machine, and as a result of a sharp decrease in temperature in contact with air, lignin hardens, which forms the mechanical properties of the formed product.

The quality of the formed briquettes is also affected by the humidity of the raw material. In the general case, this value should be in the range of 7-14% [6,8,9]. At higher humidity, the internal water pressure that occurs during the pressing of raw materials will not allow the formation of quality briquettes and will lead to the stratification of briquettes.

The main goal of the article was to determine the calorific values of dried alcohol distillery stillage of grain origin for the use as a raw material for solid fuel production.

Experimental

The object of research was corn alcohol distillery stillage after the centrifugation process, obtained on the production line of SE «Vuzlove distillery» (Vuzlove, Lviv region, Ukraine). To remove moisture from the test material, the alcohol distillery stillage was dried by filtration drying to a constant weight according to the method described elsewhere [1].

To determine the residual moisture content in the dried material, studies were performed according to the method described in the State Standard of Ukraine DSTU EN 14774-2:2013 (Solid biofuels. Determination of moisture content).

The ash content of the investigated raw materials was also determined according to the requirements of the State Standard of Ukraine DSTU ISO 18122:2017 (Solid biofuels. Method for determining the ash content).

According to the method of the State Standard of Ukraine DSTU EN 14918:2016 (EN 14918: 2009, IDT. Solid biofuels), the calorific values of dried alcohol corn distillery stillage were determined. Calorimetric studies were carried out using a precision combustion calorimeter B-08-MA with an isothermal shell (± 0.003 K) and a static calorimetric bomb. The energy equivalent of the calorimetric system (W=10348±10 J/V) was determined with an accuracy of $\pm 0.09\%$ by burning the reference benzoic acid grade K-1, the content of the main component of which reached 99.995±0.01 mol.%. The heat of combustion of benzoic acid, considering the Jessup factor, was -26434.4 J/g.

Alcohol distillery stillage samples were incinerated in terylene ampoules. Terylene also serves as auxiliary material, as the ignition of the sample occurred simultaneously over the entire surface, which encourages an increase in temperature in the combustion zone and increases the completeness of combustion of the sample. The terylene ampoule was tied with cotton thread and placed in a platinum cup. Ignition of the sample during the experiment was initiated by the discharge of capacitors through a platinum wire, which ignited the cotton thread. The initial pressure of oxygen previously purified from combustible impurities, carbon dioxide and water was 32 kPa, and the initial temperature of the main period in all experiments was 298 K. Determination of calorific value was carried out according to Eq. (1).

$$Q = (W\Delta T - q_{thread} - q_{ampoule} - q_{HNO_3} + q_{soot})/m, \quad (1)$$

where W is the energy equivalent of the calorimetric system; m is the mass of the substance that burned during the experiment; q_{thread} , $q_{ampoule}$, q_{HNO_3} , q_{soot} are the amount of energy released during the combustion of cotton thread (16704.2 J/g), terylene ampoule (22944.2 J/g), the formation of a solution of nitric acid (59 J/g) and soot formed (32800 J/g), respectively; ΔT is true temperature rises in the calorimetric experiment.

The dried alcohol distillery stillage was pressed in a P474A hydraulic press at a pressure of 100 kgf/cm² and a temperature of 150° C for 60 s, 120 s and 180 s to obtain prototypes of solid fuel briquettes.

Results and discussion

The average humidity of the dried alcohol corn distillery stillage (Fig. 1) was $\sim 3 \text{ wt.}\%$ [2].

Determination of the ash content of the studied samples was performed at a temperature of 400°C according to the method mentioned above. The results of experimental studies are presented in Table 1, the average ash content of the tested samples of dried corn distillery stillage being 2.1 wt.%.

Table 1



Fig. 1. The dried alcohol corn distillery stillage

Sample No.	Ash content, wt.%				
1	2.8				
2	2.8				
3	2.1				
4	2.0				
5	2.0				
6	1.4				
7	1.6				
average	2.1%				

The results of experimental determination of ash content

After that, it was experimentally determined the calorific value of dried alcohol corn distillery stillage according to the calorimetry of combustion given in Table 2, that was 19545 kJ/kg (4672 kcal/kg) in average.

For comparison, the values of ash and calorific value of miscanthus and energy willow are given in Table 3, as they are widely used to produce solid fuel briquettes [10].

The obtained value of calorific value of the investigated material slightly exceeds the value of analogs, which indicates a theoretically higher efficiency of solid fuel briquettes from alcohol distillery stillage and might solve the problem of utilization large volumes of its industrial waste. The results of measuring the calorific value of dried alcohol corn distillery stillage showed the prospects and efficiency of its use for the manufacture of solid fuel briquettes. In view of this, it was necessary to conduct further research on the briquettes making.

Let us note that the made pellets must possess a number of certain properties for their subsequent use and burning. For example, according to the German standard DIN 51731, pellets must meet the following parameters: certain geometric dimensions, density (<1.2 kg/dm³), moisture content (<12%), ash content (<1.5%), calorific value (3705–4661 kcal/kg) and the specified content of certain chemical elements (N, S, Cl, etc.) [11]. There are also many other standards for the pellets in European countries. Compliance with the existing requirements for solid fuel briquettes and pellets in the manufacture regulates the efficiency and safety of their use.

Piston, screw and hydraulic presses are most often used in the industry for the production of solid fuel briquettes. With the help of hydraulic presses, briquettes usually are produced in the form of parallelepipeds, the shape of which is in great demand. Pressing on these devices is carried out cyclically. The raw material is loaded into the matrix, after which the briquette is formed using a punch. After a set pressing time, the briquette leaves the mold with the help of ejectors, and the cycle is repeated again [6,12].

A P474A hydraulic press machine was used for the experiments. Solid fuel in the form of briquettes was prepared from corn alcohol distillery stillage, dried to the moisture content of $\sim 7\%$ according to the recommendations [6,8,9]. The dried alcohol

Table 2

Test No.	No mg AT V as a L a J amo J	a I	aI	Q				
Test NO.	m, g	$\Delta T, V$	q _{thread} , J	Qampoule, J	q_{HNO_3} , J	q _{soot} , J	kJ/kg	kcal/kg
1	0.320315	0.71478	1031.7	115.3	8.3	28.4	19573	4678
2	0.41836	0.92945	1364.4	109.5	11.8	32.8	19517	4665
average							19545	4672

The results of experimental determination of calorific value according to the calorimetry of combustion

Table 3

The calorific value of dried distillery stillage and widely used plants for solid fuel production

Parameter	Value					
Falanetei	dried corn distillery stillage	energy willow	miscanthus			
Ash content, wt.%	2.1	≈1	≈2.2			
Calorific value, kJ/kg	19545	≈17600	≈17500			



Fig. 2. Test sample of solid fuel from corn alcohol distillery stillage after pressing for 60 s (sample 1)

O.S. Ivashchuk, V.M. Atamanyuk, R.A. Chyzhovych, S.S. Kiiaieva, R.R. Zherebetskyi, I.B. Sobechko

Table 4

			~ I	a I		Q	
m, g	ΔΤ, V	q _{тер амп} , J	q _н , J	$q_{_{\mathrm{HNO}_3}}$, J	q _c , J	kJ/kg	kcal/kg
		Sam	ple 1 (150 ⁰ C	, 60 s, 100 kg	gf/cm ²)		
0.53400	1.25163	904.0	134.5	14.8	65.6	22405	5354
0.52670	1.26037	1122.0	76.8	15.9	44.3	22540	5387
0.55335	1.19907	-	121.9	19.8	123.0	22390	5351
		Averag				22445	5364
Sample 2 (150°C, 120 s, 100 kgf/cm ²)					-		
0.38925	0.90192	-	92.7	14.8	95.1	23945	5723
0.35283	0.82550	-	103.6	7.7	91.8	24155	5773
0.47070	1.08325	-	89.4	12.4	190.2	24002	5736
	Average value				24034	5744	
Sample 3 (150°C, 180 s, 100 kgf/cm ²)							
0.24217	0.63235	_	120.8	7.7	45.9	26679	6376
0.25820	0.66858	_	107.7	7.4	41.3	26509	6336
Average value					26594	6356	

The results of experimental determination of higher calorific value according to the calorimetry of combustion

distillery stillage was pressed at a pressure of 100 kgf/cm² and a mold temperature of 150°C for 60 s (sample 1), 120 s (sample 2) and 180 s (sample 3). The example of the obtained result is shown in Fig. 2.

The addition of binders in this method of preparing briquettes from the alcohol distillery stillage is not required. The cooled solid fuel is ready for use. It should be noted that a lower pressure was used for pressing than that given in recommendation [7].

The calorific value of the obtained briquettes was studied according to the method described above (Table 4).

In all samples, the humidity was about 1 wt.%. The ash content in the obtained briquettes is given in Table 5.

Table 5 The results of experimental determination of ash content

Sample No.	Average ash content, wt.%			
1	0.6–1.1			
2	1.2–1.6			
3	0.8–1.0			

As can be seen from Tables 2 and 4, the calorific value of briquetted fuel is slightly higher than the value of unformed dried raw materials; this is most likely due to an increase in the density of the resulting briquette during pressing at elevated temperatures and additional moisture loss.

As for the ash values of the obtained briquette samples, there are no regularities with their determined calorific value. However, it is noticeable that as the pressing time of the sample and temperature increases, the average value of calorific value increases too, which may also be associated with increasing the density and additional drying of the briquette during pressing. This issue requires additional research.

It should be noted that the sample 3 was visually fired after pressing, which indicates the critical value of the residence time in the mold 180 s at a temperature of 150° C and, accordingly, the need to find the optimal residence time of raw materials in the mold and the optimal pressing temperature.

In addition, the use of corn alcohol distillery stillage for solid fuel, which has not passed the drying stage (moisture content of the samples of $\sim 66 \text{ wt.}\%$), negatively affects the solid form of the briquette and its mechanical strength.

Conclusions

Thus, an alternative solid fuel was prepared from dried corn alcohol distillery stillage, which is characterized by high calorific value, low ash content, and environmental friendliness and can be used both in solid form (Fig. 2) and in dried form without form (Fig. 1). The average value of the highest calorific value of the dried alcohol distillery stillage is 19545 kJ/kg, and the corresponding value of solid fuel briquettes made from it is in the range from 22445 to 26594 kJ/kg. The obtained values, together with low ash content, are close to the existing standards and slightly exceed the indicators of crops that are now widely used to obtain solid fuel briquettes (miscanthus and energy willow).

The proposed method of manufacture of solid fuel is not only simple in technological implementation, but also allows ensuring the utilization of alcohol distillery stillage and the rational use of secondary raw materials of alcohol production. Alternative solid fuels from alcohol distillery stillage can further enhance the environmental situation by replacing traditional fuel resources, wood in particular, and the absence of large amounts of harmful emissions.

The obtained results testify to the prospects of further research of this alternative solid fuel, aimed at finding the optimal conditions for the formation of solid form, mechanical strength, etc.

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ВИГОТОВЛЕННЯ АЛЬТЕРНАТИВНОГО ТВЕРДОГО ПАЛИВА ІЗ ЗЕРНОВОЇ ПІСЛЯСПИРТОВОЇ БАРДИ

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Описано виготовлення альтернативного твердого палива із кукурудзяної післяспиртової барди — відходів спиртової промисловості. Охарактеризовано теплотворні характеристики осушеної післяспиртової барди та дослідних брикетів, виготовлених з неї методом пресування за підвищеного тиску та температури. Визначено, що середнє значення вищої теплотворної здатності осушеної післяспиртової барди становить 19545 кДж/кг, а виготовлених брикетів – в межах від 22445 до 26594 кДж/кг. Одержані значення теплотворної здатності перевищують показники аналогів, що широко використовуються для виготовлення альтернативного твердого палива - міскантусу та енергетичної верби. Одержане з кукурудзяної післяспиртової барди тверде паливо можна використовувати як у ствердлій формі, так і в сухому вигляді. Запропонований спосіб одержання твердого палива дозволяє забезпечити додаткову утилізацію післяспиртової барди та раціональне використання вторинної сировини спиртового виробництва.

Ключові слова: післяспиртова барда, вторинна рослинна сировина, тверде паливо, брикети, теплотворні характеристики.

PREPARATION OF AN ALTERNATE SOLID FUEL FROM ALCOHOL DISTILLERY STILLAGE

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