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Evaluation of *In Vitro* Antioxidant Properties of Methanolic and Aqueous Extracts of *Terminalia bellerica* Roxb Leaves

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Abstract: The antioxidant activity of methanolic (MTBL) and aqueous (ATBL) extracts of Terminalia bellerica leaves (TBL) was evaluated by various in vitro techniques. IC_{50} values of MTBL and ATBL were found to be 93 and 301 µg mL⁻¹ for FRSA, 42 and 252 µg mL⁻¹ for SARSA, 91 and 376 µg mL⁻¹ for LPO, 53 and 540 µg mL⁻¹ for HRSA, 142 and 544 µg mL⁻¹ for FTC, respectively. RP was detected to be 2.72 and 10.49 ASE mL⁻¹ for MTBL and ATBL. The MTBL showed better antioxidant activity than ATBL when compared with standard quercetin. TPC for MTBL and ATBL were 209.28 and 173.9 g kg⁻¹ of GAE, respectively. In conclusion, methanolic extract of TB leaf has higher antioxidant activity than the aqueous extract and it could be a good source of natural antioxidant.

Keywords: Terminalia bellerica, Antioxidant, Total Phenolic Content.

Introduction

Oxidative stress, defined as a disturbance in the balance between the production of Reactive Oxygen Species (ROS)/free radicals and antioxidant defenses. As a result, ROS/free radicals can attack on biomolecules such as lipid, protein and DNA in cells and tissues, thus inducing various cardiovascular and neurological disorders (Dhalla et al., 2000; Sayre et al., 2001). It not only causes lipid peroxidation, protein oxidation and oxidative DNA damage, but also creates interference in physiologic adaptation phenomenon and regulation of intracellular signal transduction mechanism. Two types of antioxidant (enzymatic and non-enzymatic) are present in living organisms that are usually effective in blocking harmful effects of ROS/free radicals. Many synthetic antioxidants are currently in use in various foods and pharmaceuticals industries but they may cause cellular toxicity; however, there is an increasing demand of consumers for consumption of natural antioxidants because of their good efficacy and lower side effects on health (Singh et al., 2009a; 2009b).

Terminalia bellerica (Combrataceae) is one of an important constituent of a well-known ancient polyherbal formulation, Triphala. It is a large deciduous tree found throughout Indian forests and plains. The fruits of TB are commonly used for curing cough, diarrhea, fever, piles, oral thrush, and skin diseases and have analgesic, astringent, expectorant and laxative properties (Parekh and Chanda, 2007). The fruit also showed strong antimicrobial potential against different microbes such as bacteria, virus and protist (Gupta et al., 2015, Singh et al., 2016). Plant is a richest source of various bioactive phytocompounds such as polyphenols, terpenoids, tannins, flavonoids etc (Dhingra and Valecha, 2007). Various reported studies showed antioxidant potential of fruits of T. bellerica but there is lack on information about antioxidant potential of methanolic and aqueous extract of T. bellerica leaf. Present work focuses on evaluation of antioxidant potential of T. bellerica leaf extract and their correlation with the total phenolics.

Materials and Methods

Plant Materials

Plant materials were collected from herbal garden of Narendra Dev University of Agriculture and Technology Kumarganj, Faizabad, U.P., India and identified with the help of Dr. M. N. Srivastava, Senior Scientist, Botany Division, CSIR-CDRI, Lucknow, India and the voucher specimens were submitted in CDRI herbarium.

Chemicals and Reagents

Gallic acid, DPPH, TBA was purchased from sigma, USA and nitroblue tetrazolium, phenazine methosulphate, trichloroacetic acid, NADH, FeCI₃, SDS form SRL India. Rest of all chemicals used was of analytical grade.

Extraction Procedure

Twenty grams of dried plant sample powder of *T. bellerica* leaf (TBL) was extracted with 70% methanol (MTBL) and double distilled water (ATBL) until decolouration, followed by evaporation in a vacuum rotary evaporator at 40°C and lyophilised till dryness.

Antioxidant Studies

Total Phenolics Content (TPC) was estimated as described by the method of Ragazzi and Veronese (1973), Free radical, superoxide anion (O_2^{-}) radical and hydroxyl radical scavenging activity was measured by method of Yen and Duh (1994), Nishikimi *et al.*, (1972) and Halliwell *et al.*, (1987). Lipid peroxidation, Ferric thiocyanate, and reducing power were measured by method of Ohkaowa *et al.*, (1979), Tsuda *et al.*, (1994) and (Apati *et al.*, 2003).

Statistical Analysis

Statistical analysis was done using prism software. Values of *in vitro* antioxidant activities were reported as mean ± Standard Deviation

(SD) of three determinations. The r² value and regression equation were calculated through plotting graph between TPC on x-axis and anti-oxidant deciding parameters on y axis with the help of MS office excel 2007.

Result and Discussion

TPC

The phenolics have been characterized as phenolic acid, flavonoid and stilbenoids. Due to their useful antioxidant activity, phenolics have repeatedly been implicated as natural antioxidants in fruit, vegetable and other medicinal plants. The total phenolic content of MTBL and ATBL were found to be 209.28 and 173.9 g kg⁻¹ of GAE (Figure 1) which are in agreement with the earlier findings of Ayoob et al., (2014). Plant extract containing high phenolic content showed better antioxidant activity than other. Various studies showed that the phenolic compounds are mainly responsible for antioxidant potential of medicinal plants; therefore the plant which contains high amount of phenolic compounds could be a natural source of antioxidant for various food and pharmaceutical industries (Gupta et al., 2015; Singh et al., 2009a; 2009b, Heim et al., 2002).

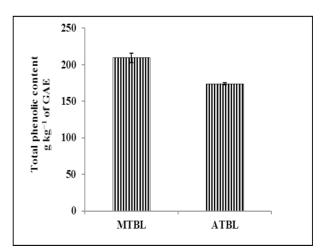


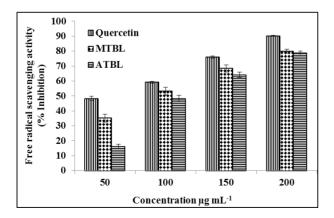
Fig. 1 Amount of phenolics in MTBL and ATBL. Values are mean \pm SD of three replications (n=3).

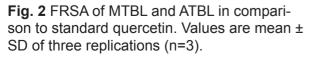
Free Radical Scavenging Activity (FRSA)

DPPH radical scavenging activity of MTBL and ATBL is presented in Figure 2. The DPPH radical (DPPH*) scavenging effect of T. bellerica leaf increases with increasing concentration of extracts from 50-200 µg mL⁻¹. MTBL exhibited higher (79.99%) inhibition potential than ATBL (51.23%) in a concentration dependent manner from 50-200 µg mL⁻¹. Kathirvel and Sujatha (2012) showed that 70% methanolic leaf extracts of T. chebula inhibits DPPH* of inhibitory concentration (IC₅₀) of 143 μ g mL⁻¹ which is lower than IC $_{\rm 50}$ value of 93 μg mL 1 reported by us. DPPH is stable nitrogen centered, lipophilic free radical widely used in evaluating antioxidant potential of plant extracts in a relatively short time with high specificity compared to some other methods. The DPPH assay use to measure the ability of the plant extracts to donate electron to free radical resulting in decolorizing of the DPPH solution. Greater the decolorizing action, higher the antioxidant activity and this is reflected by higher anti radical power and lower value of IC₅₀

Superoxide Radical Scavenging Activity (SARSA)

MTBL showed higher superoxide scavenging activity (79.70%) than ATBL (62.09%) in a concentration dependent manner (Figure 3). O_2^{-1} is





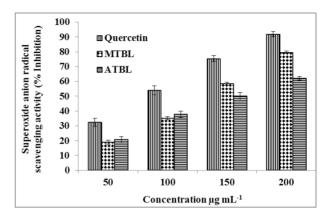
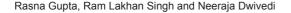


Fig. 3 Inhibitory effects of MTBL and ATBL on superoxide anion radical at varying concentrations in comparison to standard quercetin. Values are mean \pm SD of three replications (n=3).

a powerful free radical and have capability of creating auto-oxidizing chain reactions by oxidation of various biomolecules. It gives rise to generation of powerful and dangerous OH and ¹O₂, both contribute to oxidative stress. The SARSA of T. bellerica leaf was monitored by a non-enzymatic method known as PMS-NADH-NBT reduction system. In this method, O₂. derived from dissolved oxygen by PMS-NADH coupling reaction reduces the yellow dye (NBT²⁺) to produce the blue formazan, which is measured spectrophotometrically at 560 nm. The decrease in color intensity showed that antioxidants present in the plant extracts scavenges the O⁻, by inhibiting the formation of blue formazan complex. According to Alam et al., (2011) the inhibitory potential of methanolic extract was 47.32% and aqueous extract did not show any O2- scavenging activity which is different from our value 79.70 and 62.09% for MTBL and ATBL, respectively.

Lipid Peroxidation (LPO)

The study was further carried out to determine the anti-LPO activity of plant extracts in presence of ferrous ion under *in vitro* conditions and was expressed in terms of percentage inhibition (Figure 4). MTBL showed higher anti-LPO activity (75.08%) than ATBL (54.82%) at 400 µg mL⁻¹



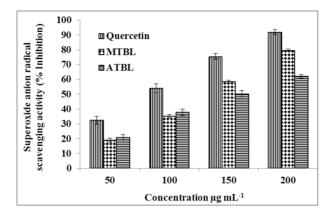


Fig. 4 Inhibitory effects of MTBL and ATBL on LPO in comparison to standard quercetin by using egg homogenate as a lipid-rich source at varying concentrations. Values are mean \pm SD of three replications (n = 3).

of plant sample was added to reaction mixture, respectively, and the result was in agreement with earlier finding (Sherin et al., 2015). LPO is a successive process of oxidative degradation of polyunsaturated fatty acids present in the biological membranes and production of a variety of secondary products including several aldehydes such as MDA. MDA reacts with TBA and produce pink colored products. MDA is the major reactive aldehyde resulting from the peroxidation of polyunsaturated fatty acid present in biological membranes (Vaca et al., 1988). So, the plant parts having better protection against free radical induced LPO may be used as anti-LPO as well as anticarcinogenic/antimutagenic substances. The high LPO scavenging effects in T. bellerica leaf observed in our experiment may be due to the high contents of phenolic compounds or other radical scavengers present in the extracts which can terminate the peroxidation chain reaction easily and quench free radicals, thereby inhibiting the oxidation of lipid and other biological molecules.

Hydroxyl Radical Scavenging Activity (HRSA)

In this experiment, protection of DNA by plant extracts against OH induced damage was

determined in terms of the damage to its deoxyribose sugar moiety. The effect of T. bellerica leaf extract on hydroxyl radicals generated by Fe³⁺ ion was measured by determining the degree of deoxyribose degradation, as indicated bythiobarbituric acid-malondialdehyde (TBA-MDA) adduct formation. HRSA of MTBL and ATBL are presented in Figure 5. MTBL showed high inhibition potential than ATBL in a concentration dependent manner. MTBL and ATBL extracts showed 42.06, 55.63, 65.22, 78.15% and 30.20, 37.72, 43.90, 53.86% inhibition at 50, 100, 150 and 200 µg mL⁻¹ concentrations, respectively. Hazra et al., (2010) showed that 70% methanolic fruit extract of T. bellerica inhibits OH at IC₅₀ of 203 µg mL⁻¹ which is higher than IC_{50} value of 53 µg mL⁻¹ reported by us.

Ferric Thiocynate Assay (FTC)

Membrane lipids are the main targets of LPO by free radicals. The FTC method is used to assess the amount of peroxide produced during the initial stages of LPO, whereas TBA method is used to measure a later stage of LPO. FTC assay is based on the ability of antioxidants to reduce Fe³⁺ to Fe²⁺ in the presence of ammonium thiocyanate, forming an intense red Fe²⁺ thiocyanate complex with absorption maxima at 500 nm. Antioxidant/radical scavenge converts ferric iron back to ferrous iron, itself

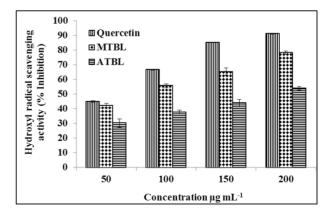


Fig. 5 HRSA activity of MTBL and ATBL in comparison to standard quercetin. Values are mean \pm SD of three replications (n = 3).

becoming oxidized, thus allowing another cycle of OH[•] generation from renewed ferrous iron. MTBL and ATBL showed 36.19, 60.13, 65.96, 82.55% and 20.72, 26.34, 35.45, 51.91% ferric thiocynate activity at 50, 100, 150 and 200 μ g mL⁻¹ concentrations, respectively (Figure 6) showing methanolic extract being more potent of peroxide generation. Result showed that the *T. bellerica* leaf extract is an active scavenger of Fe³⁺ ion which is in agreement with the work done on known Fe³⁺ scavengers (Singh *et al.*, 2013, 2014).

Reducing Power (RP)

RP of extracts was presented in terms of ascorbic acid equivalent per milliliter (ASE mL⁻¹). In tested extracts, MTBL showed high RP of 2.72 ASE mL⁻¹ than ATBL (10.49 ASE mL⁻¹), which is comparable to standard querce-tin value 1.12 ASE mL⁻¹ (Figure 7). Refahy and Saad (2014) reported similar RP (2.66) of MTBL as showed by us (2.72 ASE mL⁻¹). With regard to RP, a higher reducing capacity might be attributed to higher amount of total phenolics (Singh *et al*, 2013).

Correlation between TPC and Antioxidant Activities

Correlation between TPC and antioxidant studies showed strong correlation coefficient of $R^2 = 1$ (Figure 8a to 8e). This indicates that the phenolic contents of *T. bellerica* leaf are responsible for its antioxidant activity. Many plants exhibit efficient antioxidant properties because of their phenolic constituents (Cook and Samman, 1996; Larson, 1988). These findings suggest that the phenolic contents of *T. bellerica* leaf are highly attributed to their antioxidant activity.

In conclusion, this study asserts the *in vitro* antioxidant potential of methanolic and aqueous extracts of leaves of *T. bellerica*. Results are compared with values of standard natural antioxidant quercetin. Out of the two extracts, methanolic extract have more potential anti-

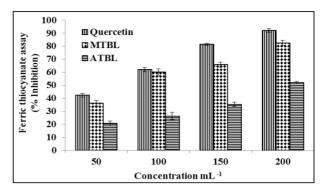
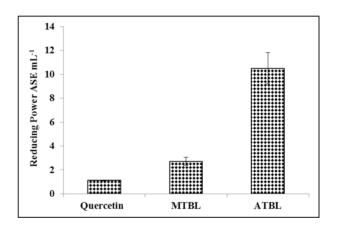
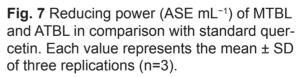
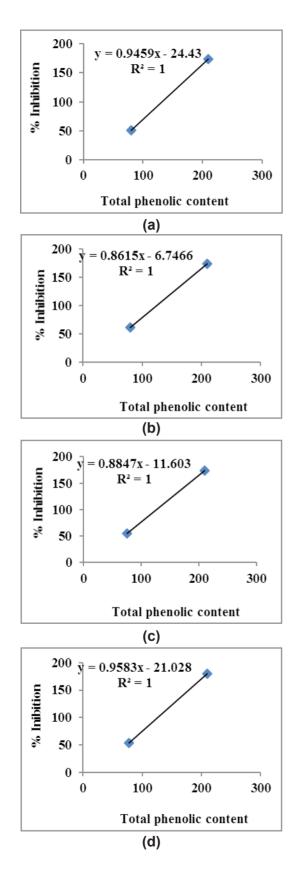


Fig. 6 Inhibitory effects of MTBL and ATBL on ferric ion chelation by ferric thiocyanate assay method at varying concentrations in comparison with standard quercetin. Each value represents mean \pm SD of three replications (n = 3).





oxidant activity than aqueous extract. It is very clear that methanol can extract the antioxidant molecules better than water from *T. bellerica* leaf. In a nutshell, methanolic extract of *T. bellerica* leaf could be a potential source of natural antioxidants for food and pharmaceutical applications. Further studies are in progress to elucidate the mechanistic analysis of antioxidant, and anti-inflammatory potential of *T. bellerica* leaf.



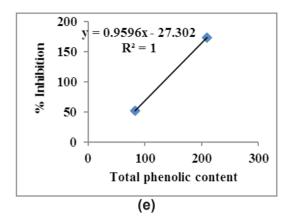


Fig. 8 Linear Correlation between TPC (x axis) in the plant extracts in relation to their antioxidant activity (y axis). TPC versus FRSA (8a), TPC versus SARSA (8b), TPC versus LPO (8c), TPC versus HRSA (8d), TPC versus FTC assay (8e).

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