Hygeia.J.D.Med. May 2016 - Octoberer 2016 May 2016 Cesswww.hygeiajournal.com Research article section: Pharmaceutical Analysis A Half Yearly Scientific, International, Open Access Journal for Drugs and Medicines DOI:10.15254/H.J.D.Med.8.2016.152



Research Article

A COST EFFECTIVE HPLC METHOD FOR THE ANALYSIS OF CURCUMINOIDS

Radha A., P. Ragavendran, Alex Thomas, D. Suresh Kumar*

Confederation for Ayurvedic Renaissance-Keralam Ltd, KINFRA Small Industries Park, Nalukettu Road, KINFRA Park, P.O. - 680 309, Koratty, Thrissur District, Kerala, India.

Kev words Curcuminoids, HPLC validation, turmeric, Cost-effective method

Correspondence D.Suresh Kumar, R&D Lab, CARe Keralam Ltd, KINFRA Small industries Park, KINFRA Park P.O., Koratty- 680 309, Thrissur District, Kerala.

Received: 15 October 2015, Revised: 20 November 2015 Accepted: 25January 2016, Available online: 15 June 2016

ABSTRACT

Plan: Validation of a method for the HPLC estimation of curcuminoids Preface: Many methods are available for the assay of curcumin, the major pigment in turmeric rhizomes. These include direct fluorimetric, spectroscopic and HPLC methods. HPLC analysis of compounds is expensive, as HPLC grade water and solvents are used. Considering the expensive nature of solvents used in HPLC analysis, there is a need to develop cost-effective methods for the estimation of compounds using HPLC.

Methodology: We modified a recently-reported HPLC method for the estimation of curcuminoids. The modified method was validated and found to be accurate, precise, specific, reproducible, and rugged.

Outcome: This cost-effective method can be utilized for the speedy and routine HPLC estimation of curcuminoids.

1. INTRODUCTION

Curcumin is the major pigment and biologically active constituent of turmeric rhizomes. All oriental medical traditions use this herb for the treatment of a variety of ailments. The scientific rationale behind these uses is well-known ¹⁻⁴. Many methods are available for the assay of curcumin. They include direct fluorimetric, spectroscopic (IR, NMR, MS) and HPLC methods ⁵. However, HPLC methods are widely used considering ease and sensitivity.

Corresponding author email: dvenu21@yahoo.com Phone: +91 9449348897

Hygeia.J.D.Med. Vol.8 (1), May 2016 © All rights reserved Hygeia journal for drugs and medicines, 2229 3590 Rid: D-2044-2014

HPLC analysis of curcumin was first attempted by Asakawa *et al* (1981), using Nucelosil C₁₈ column as stationary phase, a mixture of Acetonitrile: H₂O: Acetic acid (51: 49: 5) as mobile phase and benzyl benzoate as an internal standard⁶. Many improved methods have been reported since then ⁷⁻¹⁰. HPLC analysis of compounds is expensive, as HPLC grade water and solvents are used. Therefore, there is a need to develop cost-effective methods for the estimation of compounds using HPLC.

During literature survey we came across a HPLC method for the estimation of curcumin in rat plasma ¹¹. This method employed a mobile phase consisting of acetonitrile-5% acetic acid. While working with this method we noticed that this method could be modified to yield a better and less expensive analytical method. Development of this method is reported in this communication.

2. MATERIALS AND METHODS

2.1. Modification of the method

Li *et al* (2009) performed the chromatographic separation with a mobile phase consisting of acetonitrile-5% acetic acid (75:25 v/v) at a flow rate of 1.0 ml/ mi. the wavelength of detection was 420 nm. Injection volume was 50μ l and a sample was analyzed in 3 minutes¹¹.

In the present study we performed chromatographic separation with a mobile phase composed of acetonitrile-2% acetic acid (55:45 v/v) at a flow rate of 0.5 ml /min. wavelength of 425 nm was used for detection. Injection volume was 10µl and running time was 10 minutes.

Transfer of method is best achieved by a systematic method validation process. This is carried out by challenging the method and determining the limits of allowed variability for the conditions required to run the method. The present method was validated with reference to parameters like accuracy, precision, specificity, linearity, limit of quantification (LOQ), limit of detection (LOD) and ruggedness ^{12, 13.}

2.2. Solvents

HPLC grade acetonitrile, methanol and water were procured from Merck India, Mumbai, HPLC grade acetic acid from Hi Media Laboratories Pvt Ltd., Mumbai and reference standard of curcuminoids from Merck KGaA, Darmstadt, Germany (Batch No. S6351554).

2.3. HPLC instrumentation and conditions

The HPLC system consisted of Agilent quaternary system with 1260 quat pump, injector, variable wavelength detector 1260 VWD VL and auto sampler 1260 ALS. A column (4.6 x 150 mm) packed with 5 μ m particle size C₁₈ material was used for the separations. Agilent Chem Station software was used for the control of the equipment and for data evaluation. Quantification of the compound was carried out using the peak areas method.

Chromatographic separation was achieved with a mobile phase composed of acetonitrile-2% acetic acid (55:45 v/v) at a flow rate of 0.5 ml/min. the mobile phase was filtered through a 0.45 µm membrane filter and ultrasonically degassed prior to use. Wavelength of 425 nm was used for detection. Injection volume was 10µl and running time was 10 minutes. All statistical analyses were carried out using *Graphpad Prism* Version 5.

3. RESULTS

3.1 Validation of the method

The method was validated according to the guidelines of International Conference on Harmonization of Technical Requirements for Registration of Pharmaceuticals for Human Use ¹⁴.

3.2. Accuracy

A range of four concentrations of curcuminoids standard (0.5 ppm, 1 ppm, 2 ppm, 5 ppm) was prepared and analyzed four times by the same analyst under same conditions. Area was recorded and the mean, standard deviation and % relative standard deviation (% RSD) was calculated (Table 1). The % RSD was below 1% and therefore, this method is validated for accuracy. (Table.1)

3.3. Precision: System precision

To check system precision, the same concentration of curcuminoids standard was injected six times. The area recorded and concentration of the standard was calculated from regression equation. The standard deviation and % RSD are given in Table 2. The percentage RSD of < 1 confirms that the method has system precision.

3.4. Method repeatability

A sample of turmeric extract (CKL/SP/F/07-13/0080) was analyzed six times by the same analyst. The concentration and % purity were calculated from area and then the standard deviation and % RSD were calculated. Table 3: Repeatability of the method.

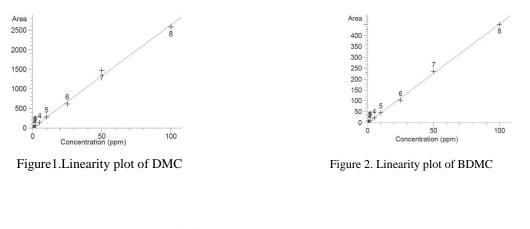
3.5. Reproducibility

Reproducibility of the methods was evaluated by analyzing the same sample on same day and different days by different analysts. The % purity values, standard deviation and % RSD were calculated from the data (Tables 4-6).

3.6. Linearity

A range of eight concentrations of curcuminoids standard was analyzed, regression equations calculated and correlated with calibration graph. A linear relationship was obtained between the peak areas and concentrations of DMC, BDMC and curcumin. The correlation coefficients for DMC, BDMC and curcumin were 0.99744, 0.99755 and 0.99729 respectively (Table 7).

Regression analysis demonstrated an excellent relationship between the peak areas and concentrations (Figures 1-3).



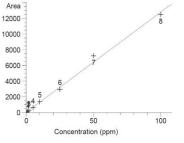


Figure 3. Linearity plot of curcumin

3.7. Recovery

Test sample was fortified by transferring 3 ml of sample solution to a 50 ml volumetric flask and spiking 0.85 ml of 296 ppm concentration of curcuminoids standard into it. The volume was made up to the mark with HPLC grade methanol. A concentration of 5.33 ppm was injected and analyzed seven times by the same analyst (Table 8).

3.8. LOD and LOQ

LOD and LOQ were determined from the specific calibration curve obtained using eight standard solutions (0.5, 1, 2, 5, 10, 25, 50 and 100 ppm). The following equations recommended in ICH (2005) were used for calculating LOD and LOQ:

$$LOD = \frac{3.3 \text{ x } \sigma}{\text{S}} \qquad \qquad LOQ = \frac{10 \text{ x } \sigma}{\text{S}}$$

Where σ is the standard deviation of the response and S is the slope of the calibration curve. The data are provided in Table 9.

3.9. Specificity: Acid degradation

Specificity of the method was established by studying the degradation of the standard. After subjecting the standard to acid degradation by the addition of 1 M HCl and heating in a water bath for 2 hrs at 40° C, the standard was analyzed six times by the same analyst. Percentage of degradation was calculated using curcuminoids standard (Table 10).

3.10. Base degradation

Base degradation was caused by the addition of 1M NaOH into the standard and keeping in a boiling water bath for 2 hrs at 40^oC. After that the solution was cooled and made up to the mark using HPLC grade methanol. This solution was analyzed by HPLC six times by the same analyst. Percentage of degradation was calculated using curcuminoids standard (Table 11).

3.11. Thermal degradation

Thermal degradation was caused by the addition of a small quantity of methanol into the standard and by keeping in a boiling water bath for 2 hrs at 80^oC. Thereafter, the solution was cooled and made up to the mark using HPLC grade methanol. This solution was analyzed six times by the same analyst. Percentage of degradation was calculated using curcuminoids standard (Table 12).

3.12. Ruggedness

Ruggedness was evaluated by small deliberate variations in experimental conditions, like changing mobile phase composition by \pm 5 ml of acetonitrile, λ max by \pm 5 nm and flow rate by \pm 0.1 ml. The optimum conditions selected for the analysis are mobile phase composition of acetonitrile- 2% aqueous acetic acid 55:45, flow rate 0.5 ml/min and λ max 425 nm.

For alteration-I the conditions selected were mobile phase composition of acetonitrile- 2% aqueous acetic acid 60:40, flow rate 0.6 ml/min and λ max 430 nm.

For alteration-II the conditions selected were mobile phase composition of acetonitrile- 2% aqueous acetic acid 50:50, flow rate 0.4 ml/min and λ max 420 nm.

The alterations caused significant changes in resolution of peak area and retention time (Table 13) confirming the robustness of the method.

3.13. System suitability

3.14. Retention time

Single concentration of curcuminoids standard was injected six times and the % RSD calculated. % RSD was 0.176.

3.15. Tailing factor

Single concentration of curcuminoids standard was injected six times and tailing factor (TF) was calculated with the following formula: The data are presented in Table 14.

$$TF = \frac{Peak \ width}{Half \ of \ peak \ width} x 2$$

3.16. Retention factor

A single concentration of curcuminoids standard was injected six times and retention factor calculate using the formula $k = (t_r - t_o)/t_o$ (Table 15).

3.17. Theoretical plates

Single concentration of curcuminoids standard was injected six times and the number of theoretical plates was calculated using the formula N = $5.545 \text{ x} (t_r/W_{b1/2})^2$ (Table 16).

3.18. Resolution

Single concentration of curcuminoids standard was injected six times and resolution calculated by the following formula:

 $\frac{R_{\rm S}}{Wb1} = 2 \frac{(tR2 - tR1)}{Wb1 + Wb2}$

Where tR2 = retention time of first peak, tR1 = retention time of second peak, wb1 = width of base peak 1 and wb2 is width of base peak 2¹⁵ (Table 17).

3.19. Measurement uncertainty

The sources of uncertainty for a HPLC method can be identified as repeatability, bias, measurement of the peak area, concentrations of standards and the mass and volume of the sample ¹⁶. Uncertainty of these factors was calculated and a measurement uncertainty budget prepared (Table 18). Measurement uncertainty limit of curcuminoids was calculated to be 96.23 ± 7.93 at 95% confidence level.

4. DISCUSSION

Methanol, acetonitrile and tetrahydrofuran are the solvents commonly employed in reversed phase HPLC analysis. Among them methanol does not provide the required resolution/selectivity for the separation of curcuminoids ¹⁰. The use of tetrahydrofuran instead of acetonitrile reverses the order elution of the curcuminoids ^{5,8,17}. Moreover, acetonitrile is the solvent of choice because of its low wavelength transparency, polarity, and intermediate position between methanol and tetrahydrofuran ¹⁸. Therefore, we developed a method based on acetonitrile ^{10, 11}.

The modified method consumes less HPLC solvents, and sample for injection. Li *et al* (2009) used a mobile phase of acetonitrile: 5% acetic acid ¹¹. However, we reduced the strength of acetic acid to 2%. The ratio of acetonitrile: acetic acid used in the original method was 75: 25. We could obtain good separation of curcuminoids with a ratio of 55:45. There was reduction in the injection volume as well. We injected 10µl of sample instead of 50µl used by Li *et al* (2009)¹¹. Injection volumes ranging from 25µl - 50 µl have been used by others ¹⁹⁻²². The analysis time was 3 minutes in the case of the original method. The peaks obtained were mixed in some cases. But we could resolve with the present method in 10 minutes, distinct peak at low concentration as 5 ppm (Figure 4).

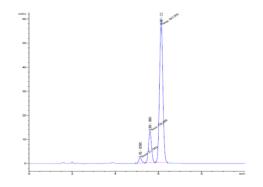


Figure 4. HPLC chromatogram of 5 ppm curcuminoids standard.

Though many methods are available for the estimation of curcumin using HPLC, several of them are not validated ^{19,21-24}. Among validated methods, many are validated for only a few parameters like linearity, precision, LOD and LOQ ^{11, 20, 25-27}. Therefore, we carried out full validation of the modified method based on Li *et al* (2009) ¹¹, confirming its practical utility.

The % RSD of area of a range of four concentrations of curcuminoids was below 1%, indicating the accuracy of the method.

Radha A et al

The % RSD related to system precision and method repeatability were also < 1 confirming the successful validation of these parameters. The measurement of the peak areas showed low values of % RSD (< 2) which suggested excellent accuracy and precision of the method.

The limit of detection (LOD) is defined as the lowest concentration of an analyte in a sample that can be detected, but not quantified. It is expressed as a concentration at a specified signal: noise ratio, usually 3:1 ²⁸. The limit of quantitation (LOQ) is defined as the lowest concentration of an analyte in a sample that can be determined with acceptable precision and accuracy under the stated operational conditions of the method. The ICH has recommended a signal: noise ratio 10:1 ¹⁴. The LOD and LOQ were calculated in the present study, based on the standard deviation of the response (SD) and the slope of the calibration curve using eight standard solutions. The LOD for DMC, BDMC and curcumin were 0.3557, 1.796 and 0.0738 ppm respectively. Similarly, the LOQ for DMC, BDMC and curcumin were 1.0781, 5.445 and 0.2236 ppm respectively.

Analysis of the same sample on same day and different days by different analysts showed that the method is reproducible. The % RSD of these analyses were below1%, confirming the reproducibility of the method.

Small deliberate changes in mobile phase composition by ± 5 ml of acetonitrile, λ max by ± 5 nm and flow rate by ± 0.1 ml caused significant changes in resolution of peak area and retention time, indicating the ruggedness of the method. Thus it is evident that the modified HPLC method for estimation of curcuminoids reported in this communication is found to be economical, precise, specific, reproducible, and rugged for routine analysis.

SI. No	Concentration	Detector	Detector Response				Standard	% RSD
	(ppm)	Ι	II III		IV	Mean	Deviation	% KSD
1.	0.5	54.8	54.5	54.4	54.1	54.45	0.2886	0.5301
2.	1.0	135.6	134.7	134.0	134.1	134.60	0.7348	0.5459
3.	2.0	267.6	266.6	268.1	267.5	267.45	0.6245	0.2335
4.	5.0	629.9	629.9	631.2	631.5	630.63	0.8461	0.1341

Table 1 .Accuracy of the method

Replicate	Sys	tem precision
	Area of curcumin *	Concentration from graph (ppm)
1	659.6	5.01
2	656.2	4.98
3	660.3	5.01
4	659.3	5.00
5	658.1	4.99
6	656.6	4.98
	Mean	5.00
	Standard deviation	0.014
	% RSD	0.280

*Atomic units

Replicate	Repeatabili	Repeatability of samples (area)*			% Purity			
	BDMC	DMC	Curcumin	BDMC	DMC	Curcumin	curcuminoids	
1	95.4	741.5	4277.7	1.690	13.136	75.785	90.61	
2	91.6	734.0	4273.0	1.622	13.003	75.702	90.32	
3	92.3	765.9	4271.0	1.635	13.569	75.666	90.87	
4	93.7	743.1	4287.8	1.660	13.165	75.964	90.78	
5	89.3	739.5	4251.4	1.582	13.101	75.319	90.00	
6	85.1	754.3	4249.3	1.507	13.363	75.282	90.15	
Mean							90.46	
Standard de	eviation						0.352	
% RSD							0.389	

Table 3. Repeatability of the method

Table 4. Inter day analysis-I

Domligata	Reproducib	Reproducibility of samples (area)*				% purity of	
Replicate	BDMC	DMC	Curcumin	BDMC	DMC	Curcumin	curcuminoids
1.	95.4	741.5	4277.7	1.690	13.136	75.785	90.61
2.	91.6	734.0	4273.0	1.622	13.003	75.702	90.32
3.	92.3	765.9	4271.0	1.635	13.569	75.666	90.87
4.	93.7	743.1	4287.8	1.660	13.165	75.964	90.78
5.	89.3	739.5	4251.4	1.582	13.101	75.319	90.00
б.	85.1	754.3	4249.3	1.507	13.363	75.282	90.15
Mean							90.46
Standard de	viation						0.352
% RSD							0.389

*Atomic units

Table 5.Inter day analysis-II	
-------------------------------	--

Replicate	Reproducil	Reproducibility of samples (area)*			% Purity			
	BDMC	DMC	Curcumin	BDMC	DMC	Curcumin	curcuminoids	
1.	94.7	743.7	4316.1	1.67	13.17	76.46	91.39	
2.	87.4	734.9	4282.2	1.54	13.01	75.86	90.43	
3.	97.5	753.7	4307.0	1.72	13.35	76.30	91.38	
4.	95.7	763.3	4300.1	1.69	13.52	76.18	91.40	
5.	90.7	743.4	4279.9	1.61	13.17	75.82	90.60	
6.	86.7	738.6	4261.2	1.53	13.08	75.49	90.11	
Mean							90.89	
Standard de	viation						0.575	
% RSD							0.630	

Radha A et al

Table 6.Intraday analysis

Domlianta	Reproducit	oility of sample	es (area*)	% Purity			% purity of
Replicate	BDMC	DMC	Curcumin	BDMC	DMC	Curcumin	curcuminoids
1.	86.3	737.5	4273.3	1.52	13.06	75.70	90.30
2.	86.6	751.6	4272.6	1.53	13.31	75.69	90.54
3.	89.9	763.9	4280.6	1.59	13.53	75.83	90.96
4.	93	743.1	4287.8	1.64	13.16	75.96	90.77
5.	91.5	764.5	4299.6	1.62	13.54	76.17	91.33
6.	89	737.6	4275.4	1.57	13.06	75.74	90.38
Mean							90.71
Standard de	viation						0.389
% RSD							0.429

*Atomic units

Table 7.Linearity of the method

Concentration		Area*	
(ppm)	DMC	BDMC	Curcumin
0.5	10.21	**	54.98
1	26.07	4.37	135.66
2	51.71	7.55	270.73
5	124.87	23.37	643.17
10	271.05	50.75	1356.13
25	585.83	112.21	2953.37
50	1448.36	278.78	7231.31
100	2527.91	489.16	12529.90

*Atomic units, **Not detected

Table 8. Recovery analysis of fortified samples

SI. No	Spiked	Obtained Conc	D 0/	
	Concentration (ppm)	Area*	Concentration	— Recovery %
1	5.33	730.0	5.35	100.48
2	5.33	733.8	5.38	101.04
3	5.33	741.0	5.44	102.09
4	5.33	738.0	5.42	101.65
5	5.33	735.2	5.39	101.24
6	5.33	733.3	5.38	100.96
7	5.33	721.8	5.29	99.28
Mean				100.96
Standard	deviation			0.903
% RSD				0.894

Sl. No.	Concentration of		Area of curcum	inoids*
	standard (ppm)	DMC	BDMC	Curcumin
1	0.5	10.21	1.54	54.98
2	1	26.06	4.36	135.66
3	2	51.71	7.55	270.73
4	5	124.87	23.37	643.17
5	10	271.05	50.75	1356.13
6	25	585.83	112.21	2953.37
7	50	1448.37	278.78	7231.31
8	100	2527.91	489.16	12529.90
Residua	al standard deviation	2.783	2.72	2.86
Slope (m)		25.820	5.00	128.05
LOD (ppm)		0.356	1.80	0.074
	LOQ (ppm)	1.078	5.45	0.224

Table 9.Derivation of LOD and LOQ

*Atomic units

Table 10. Effect of acid degradation on curcuminoids standard

Replicate	Before acid degradation		After acid degrad		
	Area of curcuminoids standard*	Purity of curcuminoids standard	Area of curcuminoids standard*	Purity of curcuminoids standard	% Degradation
1	1356.1	95.9	1284.8	90.41	5.48
2	1356.1	95.9	1286.3	90.51	5.38
3	1356.1	95.9	1285.8	90.48	5.41
4	1356.1	95.9	1289.1	90.71	5.18
5	1356.1	95.9	1286.4	90.52	5.37
6	1356.1	95.9	1282.5	90.24	5.65
Mean					5.41
Standard de	viation				0.154
% RSD					2.84

*Atomic units

Table 11. Effect of base degradation on curcuminoids standard

Replicate	Before base degradation		After base degrad	After base degradation	
	Area of curcuminoids standard*	Purity of curcuminoids standard	Area of curcuminoids standard*	Purity of curcuminoids standard	% Degradation
1	1356.1	95.9	858.1	60.38	35.52
2	1356.1	95.9	853.1	60.03	35.87
3	1356.1	95.9	851.2	59.89	36.00
4	1356.1	95.9	793.6	55.84	40.06
5	1356.1	95.9	789.0	55.52	40.38
б	1356.1	95.9	783.6	55.14	40.76
Mean					38.10
Standard de	viation				2.54
% RSD					6.66

*Atomic units

	Before thermal degradat	ion	After thermal degradation		
Replicate	Area of curcuminoids standard*	Purity of curcuminoids standard	Area of curcuminoids standard*	Purity of curcuminoids standard	% Degradation
1	1356.1	95.9	1158.4	81.515	14.68
2	1356.1	95.9	1157.3	81.438	14.46
3	1356.1	95.9	1156.0	81.346	14.55
4	1356.1	95.9	1155.3	81.297	14.60
5	1356.1	95.9	1152.5	81.100	14.80
6	1356.1	95.9	1155.9	81.339	14.56
Mean					14.61
Standard de	viation				0.118
% RSD					0.808

Table 12. Effect of thermal degradation on curcuminoids standard

Table 13.Confirmation of the ruggedness of the method

Experimental condition	Retention time & Response*	DMC*	BDMC*	Curcumin*
At optimum conditions	Retention time	5.638 ± 0.44	5.183 ± 0.50	6.15 ± 0.35
	Response**	$131.\ 175 \pm 0.67$	20.6 ± 0.76	629.725 ± 0.20
Alteration-I	Retention time	3.53 ± 0.62	3.274 ± 0.69	3.83 ± 0.57
	Response	90. 85 ± 1.99	12.7 ± 1.80	3.70 ± 0.74
Alteration-II	Retention time	9.83 ± 0.24	8.94 ± 0.28	10.86 ± 0.18
	Response	161.875 ± 0.40	25.55 ± 1.56	749.6 ± 0.39

*Values are expressed as mean ± % RSD, **Atomic units

Table 14.Derivation of tailing factor

Replicate	Peak width	¹ / ₂ of peak width	Tailing factor
1	0.177	0.088	0.999
2	0.180	0.089	1.008
3	0.178	0.090	0.991
4	0.177	0.086	1.001
5	0.177	0.089	0.992
6	0.179	0.090	0.999
Mean			0.998
Standard deviation			0.006
% RSD			0.630

Replicates	Curcumin				
	tr	to	tr - to	k	
1	6.006	2.007	3.999	1.992	
2	6.037	2.007	4.030	2.001	
3	6.031	2.012	4.019	1.997	
4	6.026	2.007	4.019	2.002	
5	6.030	2.012	4.018	1.997	
6	6.028	2.012	4.016	1.996	
Mean				1.998	
Standard deviation				0.004	
% RSD				0.180	

Table 15. Derivation of retention factor

Table 16. Derivation of number of theoretical plates

Replicates	Peak RT	¹ /2 height of peak width	Number of theoretical plates
1	6.00	0.088	25388.02
2	6.04	0.089	25192.74
3	6.03	0.090	24863.17
4	6.03	0.088	26085.84
5	6.03	0.089	25078.09
6	6.03	0.090	24949.57
Mean			25259.57
Standard deviation			444.94
% RSD			1.76

Table 17. Derivation of resolution

Replicates	tR2	tR1	wb1	wb2	Resolution factor	
1	6.01	5.49	0.177	0.164	3.01	
2	6.04	5.51	0.180	0.167	3.02	
3	6.03	5.51	0.177	0.164	3.03	
4	6.03	5.51	0.177	0.166	3.03	
5	6.03	5.51	0.177	0.165	3.05	
6	6.03	5.51	0.179	0.166	3.02	
Mean					3.03	
Standard deviation					0.014	
% RSD					0.462	

Uncertainty - Budget Standard RSU² RSU Parameters Value Uncertainty Sample Weight 0.0148 0.00337837 0.0000114134 0.00005 Made up volume 50 0.0303169 0.000003676 0.00060633 Calibration Standard (1 ppm) 1 0.025022 0.02502251 0.0006261261 Calibration Standard (2 ppm) 2 0.025022 0.01251107 0.0001565271 Calibration Standard (5 ppm) 5 0.024942 0.00498842 0.0000248844 10 Calibration Standard (10 ppm) 0.024942 0.00249421 0.0000062211 Calibration Standard (25 ppm) 25 0.000009954 0.024947 0.00099769 Calibration Standard (50 ppm) 50 0.027568 0.00049894 0.000002489 Calibration Standard (100 ppm) 100 0.00027568 0.00000076 0.02756 Standard deviation from graph 19.66 0.02939471 0.000864049 0.5779 Recovery (%) 100.78 0.00419 0.00004158 0.00000017 Repeatability (%) 96.2298 0.24 0.00249403 0.000006220 Combined uncertainty 0.041196238 Uncertainty in curcuminoids estimation 3.964305753 Effective degrees of freedom 372214.775 Coverage factor at 95% confidence level 2 7.9286 Expanded uncertainty

Table 18.Uncertainty measurement budget of the method

REFERENCES

- 1. Sharma RA, Gescher AJ, Steward WP. Curcumin: The story so far. Eur J., Cancer 2005; 41: 1955-68. CrossRef
- 2. Aggarwal BB, Shishodia S. Molecular targets of dietary agents for prevention and therapy of cancer. *Biochem Pharmacol* **2006**; 71: 1397-421. CrossRef
- Bengmark S. Curcumin, an atoxic antioxidant and natural NFkappaB, cyclooxygenase-2, lipooxygenase, and inducible nitric oxide synthase inhibitor: A shield against acute and chronic diseases. J Parenter Enter Nutr 2006; 30: 45-51.CrossRef
- 4. Maheshwari RK, Singh AK, Gaddipati J, Srimal RC. Multiple biological effects of curcumin: A short review. *Life Sciences* **2006**; 78: 2081-2087.CrossRef
- 5. Jayaprakasha GK, Rao LJM, Sakariah KK. Chemistry and biological activities of *C. longa. Trends Food Sci Technol* **2005**; 16: 533-548.CrossRef
- 6. Asakawa N, Tsuno M, Hattori T, Ueyama M, Shinoda A, Miyake Y, Kagei K. Determination of curcumin content of turmeric by high-performance liquid chromatography. *Yakugaku Zasshi* **1981**; 101: 374-377.
- 7. Tonnesen HH, Karlsen J. High performance liquid chromatography of curcumin and related compounds. *Journal of Chromatography* **1983**; 259: 367-371.CrossRef
- 8. Smith R, Witowska B. Comparison of detectors for the determination of curcumin in turmeric by high performance liquid chromatography. *Analyst* **1984**; 109: 259-261.CrossRef
- 9. Tonnesen HH, Karlsen J. Studies on curcumin and curcuminoids. VII Chromatographic separation and quantitative analysis of curcumin and related compounds. *Zeit Lebens Untersuch Forsch* **1986**; 182, 215-218. CrossRef
- 10. Jayaprakasha GK, Rao LJM, Sakariah KK. Improved HPLC method for the determination of curcumin, demethoxycurcumin and bisdemethoxycurcumin. J Agric Food Chem 2002; 50: 3668-3672. CrossRef
- 11. Li J, Jiang Y, Wen J, Fan G, Wu Y, Zhang C. A rapid and simple HPLC method for the determination of curcumin in rat plasma: assay development, validation and application to a pharmacokinetic study of curcumin liposome. *Biomed Chromatogr* **2009**; 23: 1201-1207.CrossRef
- 12. Snyder LR, Kirkland JK, Glajch JL. Practical HPLC Method Development. John Wiley & Sons Inc: New York; **1997**; 685-713.CrossRef
- Fajgelj A, Ambrus A. Principles and Practices of Method Validation. The Royal Society of Chemistry: Cambridge, 2000; 100-107.CrossRef

- 14. ICH. Validation of analytical procedures: Text and methodology (Q2 (R1). http://www.ich.org/fileadmin/Public_Web_Site/ICH_Products/Guidelines/Quality/Q2_R1/Step4/Q2_R1__Guideline.pd f 2005; Accessed on 9.12.2014.
- 15. Anonymous. Compendium of Chemical Terminology Gold Book, International Union of Pure and Applied Chemistry. http://goldbook.iupac.org/PDF/goldbook.pdf 2014; p.1074, Accessed on 10.12.2014.
- 16. Masson P. Quantitative analysis method validation quality control. In *High Performance Liquid Chromatography in Phytochemical Analysis* (edited by M. Waksmundzka-Hajnos, J. Sherma). CRC Press: Boca Raton, 2011; 351-372.
- 17. Rouseff RL. High-performance liquid chromatographic separation of spectral characterization of the pigments in turmeric and annatto. *J Food Sci* **1988**; 53: 1823-1826.CrossRef
- 18. McMaster MC. HPLC- A Practical User's Guide. Wiley-Interscience: Canada, 2007; 39. CrossRef
- 19. Wang YJ, Pan MH, Cheng AL, Lin LI, Ho YS, Hsieh CY, Lin J.K. Stability of curcumin in buffer solutions and characterization of its degradation products. *J Pharmac Biomed Anal* **1997**; 15: 1867-1876.CrossRef
- 20. Ramshankar YV, Suresh S, Pandit V, Puranik SB. Reverse phase high performance liquid chromatographic method for the estimation of curcumin. *Oriental J Chem* **2009**; 25: 105-110.
- 21. Ramshankar YV, Suresh S. A sensitive reversed phase HPLC method for the determination of curcumin. *Pharmacognosy Mag* **2009**; 5: 71-74.
- 22. Revathy S, Elumalai S, Benny M, Antony B. Isolation, purification and identification of curcuminoids from turmeric (*Curcuma longa* L.) by column chromatography. *J Exper Sci* **2011**; 2: 21-25.
- Sandur SK, Pandey MK, Sung B, Ahn KS, Murakami A, Sethi G, Limtrakul P, Badmaev V, Aggarwal B.B. Curcumin, demethoxycurcumin, bisdemethoxycurcumin, tertrahydrocurcumin and turmerones differently regulate antiinflammatory and anti-proliferative responses through a ROS-independent mechanism. *Carcinogenesis* 2007; 28: 1765-1773.CrossRef
- 24. Vareed SK, Kakarala M, Ruffin MT, Crowell JA, Normolle D, Djuric Z, Brenner D.E. Pharmacokinetics of curcumin conjugate metabolites in healthy human subjects. *Cancer Epidemiol Biomark Prev* **2008**; 17: 1411-1417.CrossRef.
- Jadhav BK, Mahadik KR, Paradkar AR. Development and validation of improved reversed phase-HPLC method for simultaneous determination of curcumin, demethoxycurcumin and bis-demethoxycurcumin. *Chromatographia* 2007; 65: 483-488.CrossRef.
- 26. Zhang J, Jinnai S, Ikeda R, Wada M, Hayashida S, Nakashima K. A simple HPLC-fluorescence method for quantification of curcuminoids and its application to turmeric products. *Anal Scie* **2009**; 25: 385-388.CrossRef.
- 27. Boominathan U, Sivakumar PK. A liquid chromatographic method for the determination of curcumin in PGR inoculated *Curcuma longa* L. plant. *Int J Pharmac Sci Res* **2012**; **3**: 4438-4441.
- 28. Anonymous. Q2B: Validation of Analytical Procedures: Methodology. Federal Register 1997; 62: 27463–27467.



Radha A., P. Ragavendran, Alex Thomas, D. Suresh Kumar. A Cost Effective HPLC Method for the Analysis of Curcuminoids *.Hygeia.J.D.Med* **2016**; 8(1):1-15. Available from http://www.hygeiajournal.com/Article ID-Hygeia.J.D.Med/152/16. DOI: 10.15254/H.J.D.Med.8.2016.152.

This is an Open Access article distributed in accordance with the Creative Commons Attribution Non Commercial (CC BY-NC 3.0) license, which permits others to share ,distribute, remix, transform, build upon this work non-commercially, and license their derivative works on different terms, provided the original work is properly cited and the use is non-commercial