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Evaluation of anti-apoptotic activity of different dietary antioxidants in renal cell carcinoma against hydrogen peroxide

Garg Neeraj K¹, Mangal Sharad¹, Sahu Tejram², Mehta Abhinav¹, Vyas Suresh P¹, Tyagi Rajeev K^{3*}

¹Department of Pharmaceutical Sciences, Dr. H. S. Gour University, Sagar, (M.P.), India

²Biologie et génétique du paludisme, Institut Pasteur, 28 Rue Du Docteur Roux, 75724 Paris Cedex 15, France

³Biomedical Parasitology Unit, Pasteur Institute, 25–28 Rue Du Dr Roux, 75724 Paris Cedex 15, France

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ABSTRACT

Objective: To evaluate the anti-apoptotic and radical scavenging activities of dietary phenolics, namely ascorbic acid, α -tocopherol acetate, citric acid, salicylic acid, and estimate H_2O_2 -induced apoptosis in renal cell carcinoma cells. **Methods:** The intracellular antioxidant potency of antioxidants was investigated. H_2O_2 -induced apoptosis in RCC-26 was assayed with the following parameters: cell viability (% apoptosis), nucleosomal damage and DNA fragmentation, bcl-2 levels and flow cytometry analysis (ROS production evaluation). **Results:** The anticancer properties of antioxidants such as ascorbic acid, α -tocopherol acetate, citric acid, salicylic acid with perdurable responses were investigated. It was observed that these antioxidants had protective effect (anti-apoptotic activity) against hydrogen peroxide (H_2O_2) in renal cell carcinoma (RCC-26) cell line. **Conclusions:** This study reveals and proves the anticancer properties. However, in cancer cell lines anti-apoptotic activity can indirectly reflect the cancer promoter activity through radicals scavenging, and significantly protect nucleus and bcl-2.

1. Introduction

Reactive oxygen species (ROS) such as hydrogen peroxide (H_2O_2), superoxide anion radical (O_2^-), and the hydroxyl radical (OH \cdot) are an entire class of highly reactive molecules derived from the metabolism of oxygen. This could be elicited by ionizing radiations and some chemical substances[1], and has been implicated in many human degenerative diseases, including aging, cancer, Huntington's disease, cardiovascular disease, mutations and neurodegenerative disorders such as Alzheimer's disease and Parkinson's disease[2]. H_2O_2 belongs to the ROS, and is known to modulate diverse cellular functions *in vivo* by producing hydroxyl radicals through the interaction with metal ions near DNA. Also, H_2O_2 induces DNA damage and cell damage by lipid peroxidation[3]. Although apoptosis and necrosis have different impacts on cellular physiology, the cellular response to H_2O_2 is continuing from apoptosis to necrosis[4] *i.e.* high concentration induces necrosis converse is applicable for apoptosis[3–5].

Many antioxidants such as resveratrol found in grapes and other food products have anticancer or anti-carcinogenic

properties along with the anti-apoptotic activity[6]. Furthermore, they can protect cells from oxidative damage and programmed cell death[7]. Antioxidants prevent carcinogenesis in murine model with anti-apoptotic activity[8].

Apoptosis, in association with oxidative stress, is an active cell death mechanism, which occurs during several pathological situations for cancer suppression[9–11]. It can be induced by two different pathways, the death-receptor pathway and mitochondrial pathway (mediated by caspase-8 and caspase-9, respectively). Eventually, both pathways lead to the activation of effector caspases such as caspase-3[12–15]. The differential expression *i.e.* pro-apoptotic versus anti-apoptotic bcl-family proteins determines the inherent susceptibility of a given cell that responds to apoptotic signals. The pro-apoptotic Bax and anti-apoptotic bcl-2 are important proteins of bcl-2 family and the ratio of bcl-2/Bax play a critical role in the regulation of apoptosis[16].

Renal cell carcinoma is a common impairment, constituting 56% of all secondary malignancies. After nephrectomy, renal cell carcinoma tends to metastasize as a solitary lesion without local tumour recurrence or distant metastasis to other organs unlike other cancers[17]. Current study is designed to evaluate radical scavenging, mechanism of anticancer and the anti-apoptotic activities of antioxidants (dose dependent) with special emphasis to renal cell carcinoma (RCC-26) cell lines.

*Corresponding author: Rajeev K Tyagi, M.Sc. (Bio-Technology), Biomedical Parasitology Unit, Institut Pasteur 5–28 Rue Du Dr Roux, 75724 Paris Cedex 15, France.

Tel: +33-1-45688000 ext.7249

Fax: +33-1-45 68 86 40

E-mail: tyagi@pasteur.fr; bhardwaj82@rediffmail.com

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2. Materials and methods

2.1. Chemicals

Ascorbic acid (AA), α -tocopheryl acetate (α -TA), citric acid (CA), salicylic acid (SA), H₂O₂, RPMI-1640, were purchased from Sigma Chemicals (St. Louis, MO, USA). Rabbit anti-bcl-2 and anti-rabbit-IgG-AP antibodies kit were purchased from Santa Cruz, CA, USA. Apoptotic DNA ladder kit and protease inhibitors were purchased from Roche Diagnostics, GmbH, Germany. RCC-26 cell line was purchased from NCCS Pune, India. Dichlorofluorescein diacetate (DCFH-DA) stock solution (10 mM in methanol) was stored at -80 °C. All other chemicals were of the highest analytical grade and purchased from common sources.

2.2. Determination of H₂O₂ scavenging activity

The H₂O₂ scavenging activity of antioxidants was determined according to the method of Ruch *et al* with some modification, based on UV spectrophotometer. H₂O₂ has optimum absorbance at 230 nm, depending on its concentration. Solution of H₂O₂ (3.5 mM) and antioxidants were prepared in phosphate buffer (pH =7.4) followed by the determination of concentration by spectrophotometer at 230 nm^[18].

Thereafter, antioxidants at different concentrations were added to H₂O₂ solution, separately. The absorbance of H₂O₂ at 230 nm was determined at appropriate time gap against a blank solution consisting of phosphate buffer as a negative control. The percentage of scavenging of H₂O₂ against antioxidant was calculated as per the formula:

$$\text{The \% of scavenged H}_2\text{O}_2 = [(A_0 - A_1)/A_0] \times 100$$

Where A₀ is the absorbance of control, A₁ is the absorbance in the presence of the antioxidants.

The ability of the antioxidants to scavenge H₂O₂ was determined at different concentration of antioxidants against different concentration of H₂O₂. Here, we optimized the antioxidant property of antioxidants against 3.5 mM H₂O₂ at 20 and 50 mg/mL concentrations.

2.3. Cell viability assessment

Cell viability was estimated by 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT) method, which is based on the cleavage of a tetrazolium salt by mitochondrial dehydrogenases in viable cells^[19]. Cultured RCC-26 cells (2 × 10⁶) were seeded and allowed to adhere in a 96-well plate using RPMI-1640 medium. Then cells were treated with various antioxidants (20 and 50 μ g/mL) followed by 2 h incubation and 3.5 mM H₂O₂ was added for another 3 h. After the incubation 25 μ L of 5 mg/mL MTT (dissolved in PBS) was added to each well followed by incubation for 4 h at 37 °C. Thereafter, formazan crystals were dissolved in 150 μ L DMSO followed by the quantitative measurement of formazan purple by the absorbance at 540 nm. Survival rate of cells (30%) treated with H₂O₂ (without antioxidant) was taken as control and percent viability for treated cells (with antioxidant) was then expressed with reference to % viability of control cells.

2.4. Nuclear morphology analysis

In this methodology, the observation of nuclear morphology has been measured through propidium iodide (PI) test. Cell lines were cultured and seeded on sterilized

cover glasses and treated with different antioxidants, at both concentrations as mentioned earlier, in different methods for 24 h followed by the PI staining (10 μ g/mL). Morphology of cell nucleus was observed by fluorescence microscope at the magnitude of 200 × consisting of the sample of 2 × 10⁶ cells^[20]. Nucleosomal damage studies have been further carried out for more precision.

2.5. Nucleosomal damage detection

Photometric immunoassay of cytoplasmic histone associated DNA fragments was used for cell death detection of quantitative measurement of apoptosis. In brief, cells were cultured for 18 h either in presence or absence of various antioxidants (AA, CA, SA, α -TA) followed by exposure of 3.5 mM H₂O₂ for 3 h. Thereafter, cells were pelleted and incubated at room temperature in the lysis buffer (provided with the kit) for 30 min and centrifuged at 200 × g for 10 min. 25 mL of supernatant containing mono and oligonucleosomes (released by treated cells) was used for enzyme-linked immunosorbent assay (ELISA). The results were expressed in terms of an enrichment factor, *i.e.*, ratio of absorbance (A₄₀₅ nm/A₄₉₀ nm) of treated cells/absorbance (A₄₀₅ nm/A₄₉₀ nm) of control cells.

2.6. DNA fragmentation

DNA fragmentation was determined by using conventional agarose gel electrophoresis with the help of apoptotic DNA ladder kit, as per the manufacturer instructions. In short, RCC-26 cells (2 × 10⁶) were lysed with the help of lysis buffer (6 M guanidine HCl, 100 mM Urea, and 10 mM Tris-HCl, 20% Triton-X-100, pH 4.4) and DNA was purified and separated by glass entrapment method in glass fiber fleece and eluted with the help of elution buffer (10 mM Tris, pH 8.5). DNA aliquots (1–3 μ g/well) were prepared and loaded onto agarose gel (1%), containing 50 μ g of ethidium bromide and were electrophoresed at 75 V for 1.5 h. DNA was visualized by placing the gel over UV-transilluminator.

2.7. Bcl-2 detection

2.7.1. Protein extraction

Firstly, cells were treated with antioxidants and then were lysed by sonication method. Cell lysate and ice-chilled methanol were mixed in the ratio of 1:4, respectively and left on ice for 1 h. Pellet of protein was collected after centrifugation at 10000 rpm for 15 min and suspended in 0.25 M Tris-HCl buffer (pH 6.8) containing 10 μ g/mL leupeptin, 1 μ g/mL aprotinin and 10 μ g/mL pepstatin. Total protein content was estimated by the method of Lowry *et al*^[21].

2.7.2. Immunoblotting

Sodium dodecyl sulphate polyacrylamide gel electrophoresis (SDS-PAGE) method was used for separation of detergent soluble proteins (40 μ g protein/lane) on 10% gel under reducing conditions. Protein solution was mixed with the sample buffer and protein's concentration was adjusted to 2 μ g/ μ L. Then after proteins were transferred on nitrocellulose membrane at 100 V for 2 h. Nitrocellulose membrane was blocked overnight at 4 °C with 5% skimmed milk powder in PBS, and subsequently incubated for 1 h with rabbit-anti-bcl-2 monoclonal antibody. Membrane was washed by PBS, which contains 0.05% Tween-20 (PBS-T), and then probed with anti-rabbit-IgG conjugated with alkaline phosphates in 2.5% skimmed milk in PBS for 1 h at 37 °C. Finally, membrane was rinsed twice with PBS-T and then with PBS. Protein bands were visualized by membrane

incubation with 5-bromo-4-chloro-3-indolyl phosphate/nitro blue tetrazolium (BCIP-NBT), a substrate for alkaline phosphatase. For comparative study, band intensity of protein was estimated by imaging densitometer software (Bio-Rad, Model GS-670) with respect to the band intensity of (H_2O_2 -untreated) control cells.

2.8. Flow cytometry analysis

Intracellular ROS was determined by measuring changes in fluorescence resulting from intracellular probe oxidation. Renal cell carcinoma cells were cultured in RPMI-1640 medium supplemented with 10% fetal bovine serum. These viable cells ($2 \times 10^6/mL$) were dispersed in 96-well micro plates for 24 h at 37 °C with 5% CO_2 . The medium was removed and cells were washed subsequently three times by PBS for the complete removal of medium, then different antioxidants were added and the microplates were incubated for 2 h. Thereafter, 20 μM 2', 7'-dichlorofluorescein diacetate (DCFH-DA) was loaded after another 45 min. The cells were washed with PBS for removing the DCFH-DA. 3.5 mM H_2O_2 was then added into the cells for another 45 min, and the change was observed by fluorescent spectrophotometer at $\lambda_{ex}=475$ nm.

2.9. Statistical analysis

The results were expressed as mean \pm standard deviation. Data obtained were subjected to student's *t*-test for statistical analysis and statistical significance was designated as ($P < 0.05$). Multiple comparisons were made using one way analysis of variance (ANOVA) followed by post hoc analysis using Tukey test to determine the significance of the results between different groups obtained using GraphPad Instat software.

3. Results

3.1. ROS scavenging activity of antioxidants

Naturally occurring antioxidants have a broad range of biochemical activities, including inhibition of ROS generation, direct or indirect scavenging of free radicals, and alteration of intracellular redox potential[2]. In this study, we have evaluated radical (generated by H_2O_2) scavenging activity of selected antioxidants by spectrophotometer. H_2O_2 scavenging (Hs) ability of the different antioxidants (at 20 and 50 $\mu g/mL$) against H_2O_2 was depicted in Figure 1, 2. H_2O_2 scavenging ability of antioxidants is concentration-dependent since it increases significantly with the increase in the concentration of antioxidants. The percentage of the H_2O_2 scavenging activity (%Hs) was 89.94% for α -TA, 71.78% for CA, 67.78% for AA and 9.12% for SA at 50 $\mu g/mL$. However, at 20 $\mu g/mL$ concentration %Hs were 69.22, 59.05, 57.76, and 5.34 for α -TA, CA, AA, and SA, respectively. As per the results obtained, α -TA had stronger H_2O_2 scavenging activity among the studied antioxidants.

3.2. Cytoprotectivity (cell viability)

Cell viability action of antioxidants was against direct oxidative damage. It is previously accepted that high levels of free radicals or active oxygen species create oxidative stress leading to a variety of biochemical and physiological lesions which in turn result in metabolic impairment and cell death[22]. We determined cytoprotective activity of

antioxidants against H_2O_2 . It was observed that antioxidants exhibited no cytotoxic effects on RCC-26 cells instead stimulated the growth of renal cells, especially the α -TA (Figure 2). α -TA increased the cell viability and protected cells against H_2O_2 significantly ($P < 0.005$) though, SA responded conversely. Moreover, AA and CA intermediated cell protection ability. These results reflected that cell survival ability was improved by antioxidants in oxidative stress conditions. Also, the enhanced cell protection ability made them plausible for functional ingredients or drugs without any significant adverse effects.

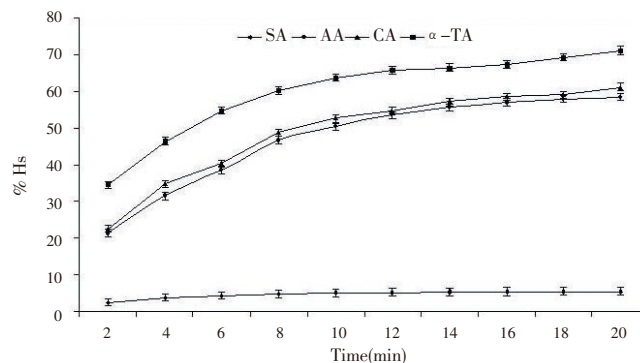


Figure 1. Percent H_2O_2 scavenging activity of different antioxidants (20 $\mu g/mL$).

Results are expressed as mean \pm SD, ($n=5$).

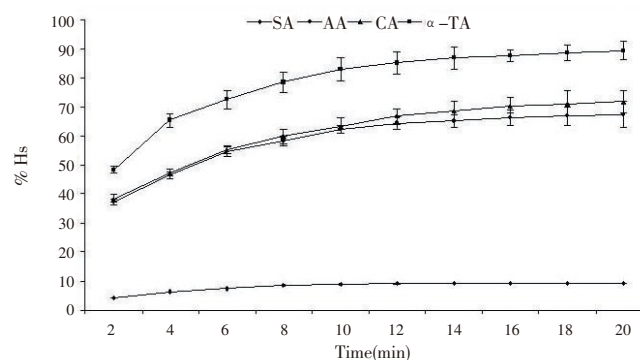


Figure 2. Percent H_2O_2 scavenging activity of different antioxidants (50 $\mu g/mL$).

Results are expressed as mean \pm SD, ($n=5$).

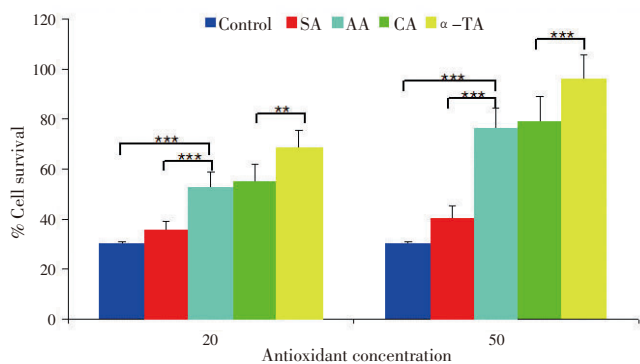


Figure 3. Effect of antioxidants on cell survival after H_2O_2 exposure.

Results are expressed as mean \pm SD, ($n=6$).

As is evident, oxidative stress is an important factor to induce cell death. In the presence of only H_2O_2 (control), cell death ratio reached up to 70% after 45 min of the treatment (Figure 3). As compared with the control, the reduction

of cell death was observed after the treatment of each antioxidant at 20 and 50 μ g/mL concentration.

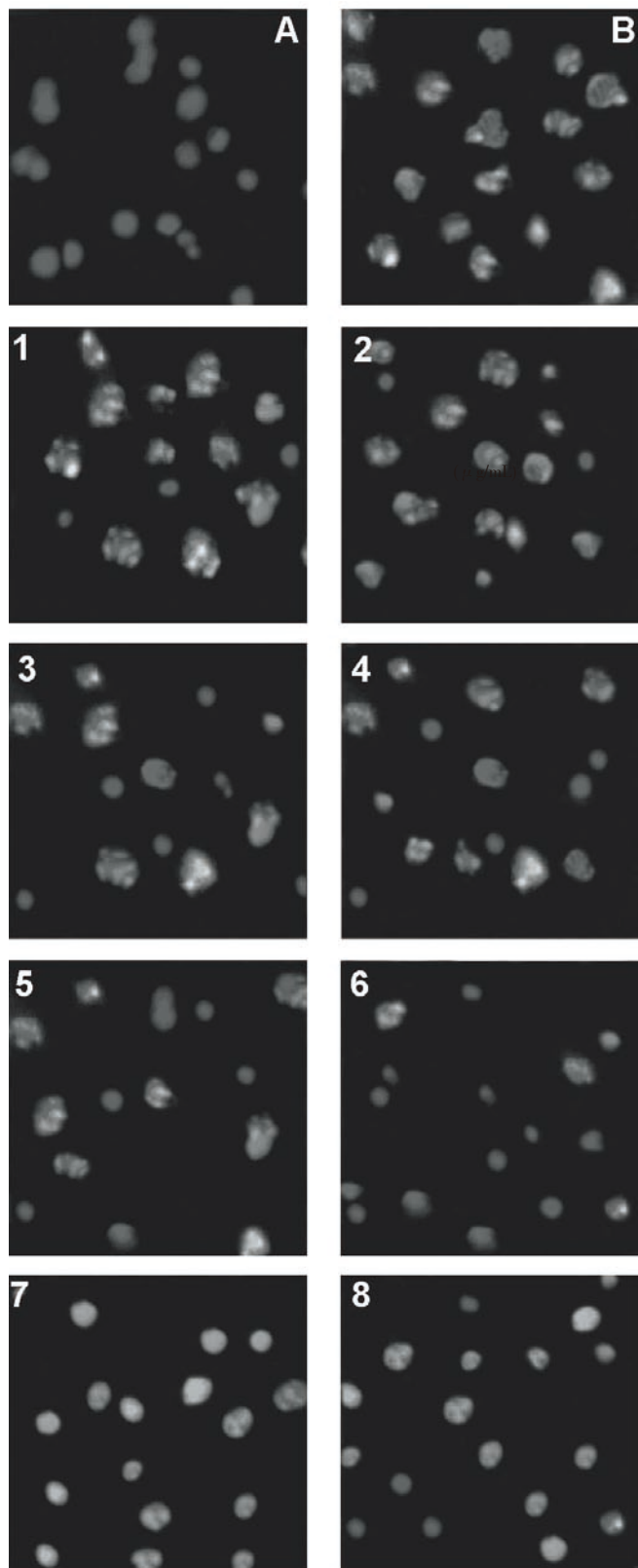


Figure 4. Nuclear fragmentation due to 3.5 mM H₂O₂. Fluorescence at the magnitude of 200X; A: Vehicle only (positive control); B: 3.5 mM H₂O₂ (negative control); 1, 3, 5, 7 at 20 μ g/mL, 2, 4, 6, 8 at 50 μ g/mL of antioxidants for 1 prior to addition of 3.5 mM H₂O₂ for salicylic acid, ascorbic acid, citric acid, alfa-tocopheryl acetate, respectively.

A-TA showed the highest protective effect among the

various antioxidants. The cell survival significantly increased with antioxidants treated cells at the concentration of 50 μ g/mL from 20 μ g/mL progressively. These results reflected the protective effect of antioxidants against intracellular oxidative damage directly induced by H₂O₂.

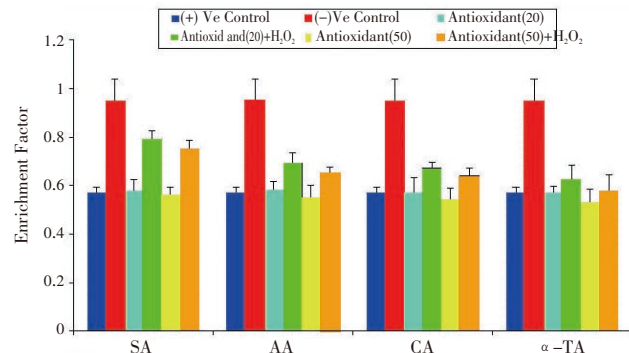


Figure 5. Effect of antioxidants on H₂O₂ induced nucleosomal damage (in terms of enrichment factor) in RCC-26. Results are expressed as mean \pm SD, (n=6).

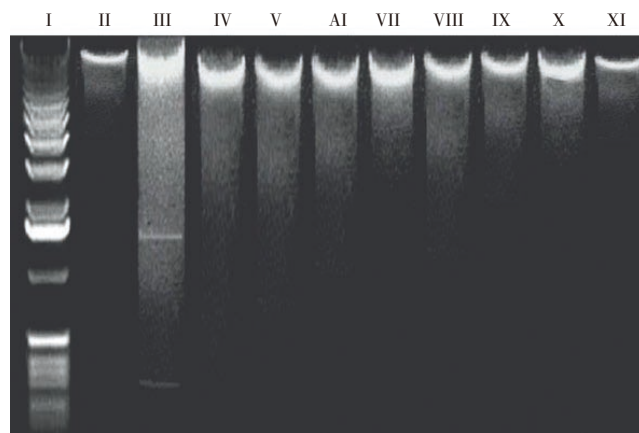


Figure 6. Agarose gel electrophoresis for DNA fragmentation. Lane I: Marker; Lane II: Normal RCC-26 (positive control); Lane III: H₂O₂ (negative control); Lane IV, V, VI, VII: for 20 μ g/mL and Lane VIII, IX, X, XI for 50 μ g/mL of antioxidant prior to addition of H₂O₂, salicylic acid, ascorbic acid, citric acid, alfa-tocopheryl acetate, respectively.

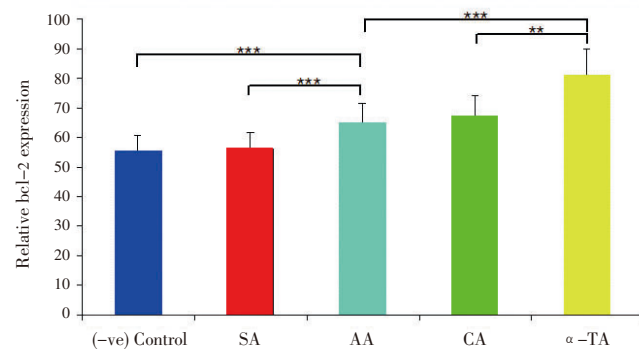


Figure 7. Effect of antioxidants on bcl-2 level (20 μ g/mL). Relative bcl-2 levels in which cell without treatment with H₂O₂ (positive control) was taken as 100%, where antioxidants were taken at 20 μ g/mL. Results are expressed as mean \pm SD, (n=6).

3.3. Protection of nucleosomal damage by antioxidants

H₂O₂ is known as a source of ROS (intra as well as extra

cellular), which can cause chromosomal and nucleosomal aberrations through oxidative damage of DNA^[3,23]. We have analyzed the protective effect of antioxidants on H₂O₂-induced apoptosis. PI was used for staining nuclei of RCC-26 cells treated either with H₂O₂ alone or with both antioxidants and H₂O₂. Cells treated with 3.5 mM H₂O₂ (Figure 4) showed significant nuclear fragmentation. However, when cells were treated with antioxidants for 1 h prior to H₂O₂ treatment, a marked reduction in nuclear fragmentation was observed depending on type (activity or potential) of antioxidants and their concentration.

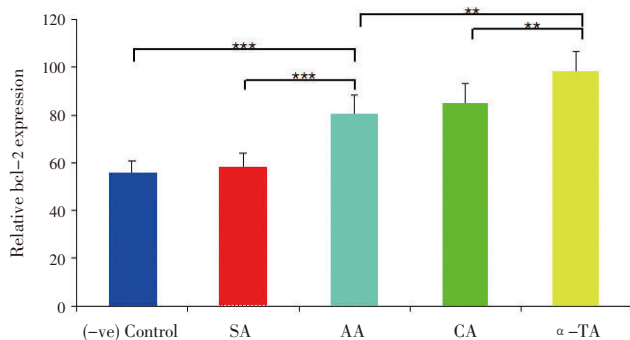


Figure 8. Effect of antioxidants on bcl-2 level (50 μ g/mL). Relative bcl-2 levels in which cell without treatment with H₂O₂ (positive control) was taken as 100%, where antioxidants were taken at 50 μ g/mL. Results are expressed as mean \pm SD, (n=6).

We estimated enrichment factor to detect nucleosomal damage, a quantitative measurement of apoptosis (cell death detection), which was carried out by ELISA. This method is based on the photometric immunoassay of cytoplasmic histone associated DNA fragments. Treatment of cell lines with antioxidants at two different concentrations (20 μ g/mL and 50 μ g/mL) with subsequent exposure to 3.5 mM H₂O₂, showed drastic change in enrichment factor as compared with positive and negative control. Comparing to positive (Figure 5), negative control cell showed 1.6 fold enrichment.

In addition, we have evaluated that antioxidants protect DNA against H₂O₂ by conventional agarose gel electrophoresis. DNA gel electrophoresis (Figure 6) pattern obtained after the treatment of different antioxidants with cell lines. Oxidative stress in cell lines by H₂O₂ for 2 h was able to produce DNA fragments (lane III) which were absent in control (lane II) and significantly visualized in cells treated with the antioxidants (lane IV, V, VI, VII at 50 μ g/mL and lane VIII, IX, X, XI at 20 μ g/mL). Pre-treatment of cells with antioxidants for 18 h prior to H₂O₂ exposure inhibited the fragmentation of DNA. However, protection of DNA against H₂O₂ depends upon the concentration and potential of antioxidants. Antioxidants at a concentration of 50 μ g/mL exhibited good DNA protective activity (Figure 6), and α -TA mainly showed the best DNA protective activity amongst all antioxidants. Other antioxidants AA and CA can intermediate, and SA had very low DNA protective activity against H₂O₂. The same order was followed at 20 μ g/mL, however, protection was directly correlated with the concentration of antioxidants.

3.4. Effect of antioxidants on bcl-2 level

Antioxidants raised bcl-2 level against the pro-oxidant in the cell mediated reaction. Thereafter, cells become resistant to oxidant and oxidant-induced apoptosis^[24]. Results (Figure 7, 8) indicated that antioxidants (SA, AA, CA and α -TA) treated at both concentrations 20 and 50 μ g/mL to cell lines for 18 h significantly protected the bcl-2 levels. Additionally,

only H₂O₂ treatment (negative control) for 2 h significantly reduced the bcl-2 levels as compared with positive control RCC-26 cell lines. The protection of bcl-2 by antioxidants was in the following order α -TA > CA > AA \gg SA at both (20 and 50 μ g/mL) concentration.

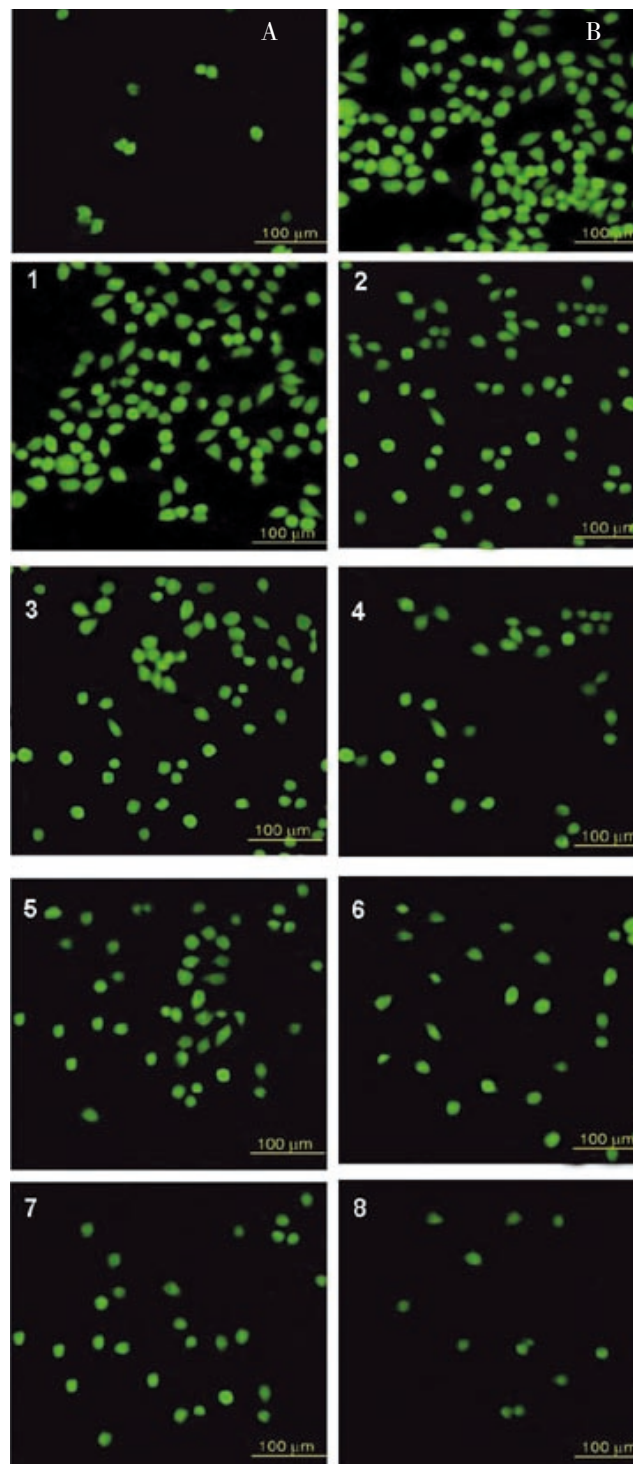


Figure 9. Fluorocytometric assay (ROS production evaluation with DCFH-DA assay/DCF fluorescence in RCC-26 cells.

A: Vehicle only (positive control); B: 3.5 mM H₂O₂ only (negative control); 1, 3, 5, 7 at 20 μ g/mL; 2, 4, 6, 8 at 50 μ g/mL, of antioxidants for 1 h prior to addition of 3.5 mM H₂O₂ for salicylic acid, ascorbic acid, citric acid, alfa-tocopherol acetate, respectively.

3.5. Protective action of antioxidants against direct oxidative stress

It is documented that antioxidants have been used to inhibit apoptosis since apoptosis is initially thought to be oxidative stress mediated phenomenon[25]. We attempted to observe the changes in DCF fluorescence with fluorescence microscopy (Figure 9). ROS and oxidative stress agents (H_2O_2) cause cell damage (Figure 9B). Enzymes were released from cell membrane like esterase cleaved ester bond of DCF-DA and DCF was produced which reflects fluorescence. Photomicrograph is a typical microscopic presentation of the DCF fluorescence in antioxidants treated RCC-26. Cells treated with 3.5 mM H_2O_2 for 45 min showed elevated levels of fluorescence as compared with the positive control samples (Figure 9A) clearly indicating that antioxidants at concentrations of 20 and 50 μ g/mL for 2 h significantly inhibited the oxidation of DCFH into DCF. The intensity of fluorescence and number of fluorescent cells was showed in microphotograph for antioxidants in following order SA »AA> CA> α -TA indicating the protective activity of cell against ROS (H_2O_2 as source) in following order α -TA >CA>AA>SA.

4. Discussion

There are sufficient evidences revealing the vital range of biological activities of antioxidants such as direct or indirect scavenging of free radicals, anticancer, antimutagenesis[2] and anti-apoptotic activity[25]. Although the role of apoptosis in pathological situation for cancer suppression has been already evaluated[9–11]. This study demonstrates that antioxidants have the potential scavenging effect on intracellular oxidative damage directly induced by H_2O_2 which, in turn, lead to an improvement in the percentage cell viability. Also, the anti-apoptotic effect of antioxidants on cancerous cells (RCC-26) is quite evident. The protective effect of antioxidants over cells against ROS has been studied. Antioxidants prevent radical (especially hydroxyl radical generated by H_2O_2) mediated cell damage and indirectly suppress apoptosis. Thus, poly-phenolic antioxidants show significant cell protecting activity and inhibit apoptosis. Again, our findings are in agreement with those obtained for H_2O_2 scavenging activity, indicating major structural features responsible for the anti-apoptotic activity against reactive radicals.

During various physiological and pathological processes, ROS (like OH^- , O_2^- etc.) is generated from H_2O_2 by mitochondria at both extra and intracellular levels[26]. This induces oxidative chromosomal damage as well as direct breakage of DNA[23]. H_2O_2 itself is not reactive enough, however, in association with super oxide anions can damage many cellular components[27]. Thus, these antioxidants which possess the property of scavenging of free radicals or ROS can protect oxidative damage followed by the decrease in the level of pro-oxidants such as H_2O_2 [28]. Therefore, H_2O_2 (ROS generating agent) has been used to evaluate the radical scavenging activity of different antioxidants. In parallel, we have studied radical scavenging activity of different oxidants at different concentration. Our results, in agreement with earlier reports, have reflected the inhibition of carcinogenesis and mutagenesis by antioxidants via the formation of adducts (with DNA), masking the binding sites which are occupied by mutagens or carcinogens[23]. Furthermore, antioxidants have large number of hydrogen atoms showing the importance to be available for abstraction and subsequent reduction and neutralization of the oxidative species. Moreover, they remain stable through electron delocalization across both side chains and conjugated ring.

Therefore, antioxidants could reduce the bioavailability and the exposure of noxious radicals towards DNA. The antioxidants (novel radical scavengers) could scavenge ROS generated by electron leakage and protect cells against apoptosis induced by ROS.

As is cited in the literature, H_2O_2 induces oxidative nucleosomal and chromosomal damage through ROS[3,23], and these ROS are scavenged by polyphenolic antioxidants[2]. In the current study the protective effect of different antioxidants against H_2O_2 mediated nucleosomal damage in cancer cell lines especially RCC-26 was investigated. Here, we observed that these antioxidants protect nucleosome against ROS produced by H_2O_2 . However, we hypothesized the dose dependent protection of nucleosome by antioxidants and their anti-apoptotic activity[25]. This dose dependency could be explained in terms of the ROS level and extent of oxidative stress.

Anti-apoptotic property of bcl-2 is quite evident either by controlling the membrane potential of mitochondria and/or inhibiting calcium ion depletion of endoplasmic reticulum[29,30]. The mechanism of anti-apoptosis through bcl-2 is explained by its pro-oxidant, though indirect, nature. The rise in the level of pro-oxidant is directly related with the increase in intrinsic cellular anti-oxidant. Here the increase of bcl-2 expression is in coherence with above mentioned observation and could be correlated like, bcl-2 mediated reaction leads the cells to make them resistant to oxidant and oxidant-induced apoptosis[24]. The anti-apoptotic function and its efficiency are controlled by the bcl-2 family members, which function partly by interacting physically with one another at mitochondrial level[31]. Here, our findings have shown the importance and role of bcl-2 over the H_2O_2 -induced apoptosis and its prevention by antioxidants in renal cell carcinoma cells. Interestingly, antioxidants used in the present study significantly reduced the H_2O_2 mediated bcl-2 degradation and/or reduction in the expression. Moreover, cell survival is prolonged by the bcl-2 gene product by inhibiting apoptosis induced by H_2O_2 mediated stimuli. Here, we observed significant changes in bcl-2 protein levels in RCC-26 cells treated with antioxidants. Presumably, there are some unknown mechanisms accounting for preventing RCC-26 from H_2O_2 -induced apoptosis. However, this change in bcl-2 level is dose dependent and also depends on antioxidant potency of the compounds.

The bond dissociation enthalpy (BDE) plays an important role in molecular properties of the compounds which make the OH^- weaker and its reaction with the free radicals become faster. As per the hypothesis already proposed, more the exothermic reactions of antioxidant with free radicals, lesser is the barrier which in turn leads the antioxidants to react faster with the radicals thereby preventing its reaction with the biological substrate[32]. We argue that α -TA is an effective antioxidant due to its larger BDE. Also, the physical and chemical environment of antioxidants seems to be an important factor for its reactivity. The position, number of hydroxyl groups, BDE, IP, and presence of bulky groups near OH^- group have considerable effect on antioxidant activity of these phenolics[33]. Apart from that, enhanced anti-apoptotic effect of α -TA can be partly attributed to its role in activation of ROS metabolizing enzymes like SOD[34].

ROS are mainly generated at the complex III site (bcl complex) in mitochondria. During respiratory chain the complex III, shuttles transfer electrons from Q10 to cytochrome c and release two protons to their inter-membrane space which in turn reduces cytochrome c. The occurred electron leakage and the these electron transports are associated with a redox shift which creates interference in the mitochondria and build up of ROS playing a major role in apoptotic cascade[35–38]. Considering these findings, we evaluated the protective role of antioxidants in apoptotic cell death induced by H_2O_2 , which is a source of ROS (complex III inhibitor).

In conclusion, current study shows that the antioxidants, preferentially α -TA, significantly inhibit cell damage and exhibit DNA protective effect in RCC-26 cell lines against

H₂O₂ induced oxidative stress. The anti-apoptotic activity of the antioxidants is mediated through scavenging ROS and can be partly attributed to bcl-2 recruitment. The anti-apoptotic activity of bcl-2 dependent mechanism is also an important finding giving insight into the molecular mechanism of reported antioxidant. Present findings bring new insight in the direction of therapeutic intervention for cancer.

Conflict of interest statement

We declare that we have no conflict of interest.

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