



Qualitative and Quantitative Determination of Various Extracts of *Ocimum basilicum* L. Leaves

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Abstract

Ocimum basilicum L. (Lamiaceae) commonly known as basil possesses various medicinal activities to treat headache, diarrhoea, kidney malfunctions, diabetes, cancer and several pharmacological activities. These are mainly due to the various chemically diversified constituents present in it. In order to evaluate the chemical composition of basil leaves, the dried leaves were extracted with petroleum ether, chloroform, ethyl acetate, methanol and water by Soxhlet's extraction. The extracts were screened for the qualitative determination by preliminary phytochemical screening. Quantitative determination was also performed for the total alkaloids, flavonoids, phenols, saponins and tannins. Amongst the extracts, methanol (MeOH) was identified to contain various classes of secondary metabolites, so GCMS analysis was performed on the MeOH extract. The results revealed that chemical constituents claimed for various biological activities were present.

Keywords: Chemical Constituents, GCMS, *Ocimum basilicum* L., Qualitative, Quantitative

1. Introduction

Ocimum basilicum L. (also called basil) belongs to *Lamiaceae*, is a culinary herb which is highly spread across Southeast Asia and worldwide. From ancient times, it is widely used to treat various ailments due to its versatile medicinal properties^{1,2}. Traditionally, it is used as folk medicine and Uyghur medicine in Turkey; treating pimples, headaches and kidney malfunctions in India; treating aches and pains in Bulgarian folk medicine, Sedative in Spain³. It is also used in the treatment of insect stings, snake bites and skin infections externally. It is also used as ornamental and kitchen herb for the aroma present in it. Essential oils extracted from basil leaves have high aroma and used as herbal flavours in most of the food products⁴. It possesses various pharmacological activities such as antimicrobial, anti-insecticidal, antioxidant, anti-inflammatory, anti-hyperlipidemic, anti-diabetic, anticonvulsant, antiplatelet, anti-thrombotic,

immune-modulatory, cytotoxicity and anti-cancer. These activities are mainly due to the various chemically diversified constituents present in it⁵⁻⁹. Alkaloids, flavonoids, phenols, terpenoids are present in basil leaves which have potent medicinal use^{10,11}.

This study was designed to evaluate the chemical composition of basil leaves with different solvents such as petroleum ether, chloroform, ethyl acetate, methanol and water. Qualitative and quantitative determination of chemical constituents were evaluated through standard procedures for all the five extracts.

Total alkaloids, flavonoids, phenols, tannins and saponin contents were determined^{12,13}. In which, Methanol (MeOH) extract showed the high percentage yield of different chemical constituents. In order to quantify, MeOH extract was subject to Gas chromatography coupled with Mass spectrometer with suitable working conditions¹⁴.

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2. Materials and Methods

2.1 Plant Collection

Matured healthy leaves of *Ocimum basilicum* L. were collected from a botanical garden in Thanjavur, Tamil Nadu, India and authenticated.

2.2 Chemicals and Glassware

The solvents used were of analytical grade (Nice chemicals). Glassware used were sterilized and graduations were accurate.

2.3 Extraction

The collected plant leaves were washed, shade dried and ground in a blender with 2 mm dia mesh size. The powdered samples of weight, each 100 g were extracted continuously with 500 ml of petroleum ether, chloroform, ethyl acetate, methanol and aqueous (1:5 w/v) in Soxhlet's apparatus. The process was continued for about 24 hours with a temperature below the boiling point of the solvent. The extracts were collected and filtered through Whatmann no.1 filter paper. Then the filtrate was concentrated in a rotary vacuum evaporator and stored in dark bottles and kept at -20° C till further use¹⁵.

2.4 Qualitative Determination of Chemical Constituents of *Ocimum basilicum* Leaf Extracts^{16,17}

2.4.1 Test for Alkaloids

Wagner's Test

To a 2-3 ml of plant extract, few drops of Wagner's reagent was added and kept undisturbed for 3-4 minutes. The reddish brown precipitate indicates the presence of alkaloid.

Mayer's Test

To a 2-3 ml of plant extracts, few drops of Mayer's reagent was added and kept undisturbed for 3-4 minutes. The formation of precipitate indicates the presence of alkaloid.

2.4.2 Test for Flavonoids

Sodium Hydroxide Test

To a 2-3 ml of plant extract, 2 ml of 10% sodium hydroxide solution was added to form an intense yellow color which would be turned to colorless in addition of diluted hydrochloric acid, this marks the presence of flavonoids.

2.4.3 Test for Terpenes

Copper Acetate Test

To a 3 ml of plant extract, 8-10 drops of copper acetate solution was added. The emerald green color indicates the presence of terpenes.

2.4.4. Tests for Carbohydrates

Molish's Test

To a 2-3 ml of plant extract, a few drops of α -naphthol solution was added and shaken well to which a few drops of concentrated sulphuric acid were added along the sides of the test tube. The formation of a violet ring at the junction indicates the presence of carbohydrates.

2.4.5. Tests for Protein

Biuret's Test

To a 2-3 ml of plant extract, a few drops of 4 % sodium hydroxide and 1% Copper sulphate solution was added. Appearance of Violet or pink color indicates the presence of proteins.

Millon's Test

To a 3 ml of plant extract, a few drops of millon's reagent was added and heated gently. The appearance of reddish brown coloration indicates the presence of proteins.

2.4.6 Tests for Amino Acid

Ninhydrin Test

To 1 ml of plant extract, a few drops of 5% Ninhydrin solution was added and the mixture was heated for about 8-10 min. Appearance of purple or blue color indicates the presence of amino acid.

2.4.7 Test for Fats and Oils (Fixed)

Spot Test

A small quantity of extract was pressed between two Whatmann No.1 filter papers for about 2 min. The oil stain on the paper indicates the presence of fixed oils.

Saponification Test

To a 2-3 ml of plant extract, 0.5 N alcoholic potassium hydroxide solution along with 2 drops of phenolphthalein was added and heated for about 2 hrs. The formation of soap or partial neutralization of alkali indicates the presence of fixed oils and fats.

2.4.8 Test for Steroids

Salkowski Tests

To a 2-3 ml of plant extract, 2 ml of chloroform & 2 ml of concentrated sulphuric acid was added and shaken well. The chloroform layer appears red and acid layer shows greenish yellow fluorescence which shows the presence of steroids.

2.4.9 Test for Cardiac Glycosides

KellarKillani's test

A few ml of plant extracts was dissolved in water with Glacial acetic acid and ferric chloride and concentrated sulphuric acid. The formation of brown ring at the junction indicates the presence of cardiac glycosides.

2.4.10 Test for Phenols and Tannins

Ferric Chloride Test

To a 0.5 ml of plant extract, 5 ml of D. H₂O was added and boiled for 10 min. To the 2 ml of collected filtrate, a few drops of 10% ferric chloride solution were added. Appearance of greenish blue or violet color indicates the presence of a phenolic hydroxyl group.

Lead Acetate Test

To a 2-3 ml of plant extract, 3 ml of lead acetate solution was added. The occurrence of white precipitate indicates the presence of tannins and phenols.

2.4.11 Test for Saponins

Foam Test

To a few quantity of the plant sample, respective amount of water was added and shaken vigorously for about 10 min. The observation of persistent stable foam indicates the presence of saponins.

2.5 Quantitative Determination of Chemical Constituents of *Ocimum basilicum* Leaf Extracts

2.5.1 Determination of Total Alkaloids

1 mg/ml equivalent of plant sample was dissolved in a dimethyl sulfoxide and 1 ml of 2 N hydrochloric acid was added and the mixture was filtered. The filtrate was transferred to a separating funnel, 5 ml of bromocresol green solution and 5 ml of phosphate buffer were added. The mixture was shaken vigorously with 1, 2, 3 and 4 ml chloroform and collected in a 10-ml volumetric flask and diluted to the volume with chloroform. The absorbance for test and standard solutions were determined against the reagent blank at 470 nm with an UV-Vis spectrophotometer. A set of reference standard solutions of atropine (20, 40, 60, 80 and 100 µg/ml) were prepared in the same manner as described earlier¹⁶.

2.5.2 Determination of Total flavonoids

Total flavonoids were determined by the aluminium chloride colorimetric assay. The reaction mixture consists of 1 ml of extract and 4 ml of distilled water was subjected to the 0.3 ml of 5 % sodium nitrite solution. After 5 minutes, 0.3 ml of 10 % aluminium chloride was mixed. After 5 minutes, 2 ml of 1M Sodium hydroxide was treated and diluted to 10 ml with distilled water. The absorbance for test and standard solutions were determined against the reagent blank at 510 nm with an UV-Vis spectrophotometer. A set of reference standard solutions of quercetin (20, 40, 60, 80 and 100 µg/ml) were prepared in the same manner as described earlier¹⁸.

2.5.3 Determination of Total Saponins

Add 20 g of powdered sample in a conical flask containing 100 ml of 20 % aqueous ethanol. The

solution was heated for 4 h with constant stirring at 55° C. Then the solution was filtered and extracted with 200 ml of 20 % ethanol. After that, both extracts were mixed and solvent was evaporated till it reached 40 ml volume of extract. The concentrated filtrate was further extracted with 20 ml of diethyl, aqueous layer was recovered while the ether layer was discarded. The aqueous extracts were purified by adding 60 ml n-butanol. Then, it was washed with twice 10 ml of 5 % aqueous sodium chloride¹⁹.

2.5.4 Determination of Total Tannins

The total tannins were determined by Folin-Ciocalteu method. About 0.1 ml of the sample was subjected to a 7.5 ml of distilled water and 0.5 ml of Folin-Ciocalteu phenol reagent and 1 ml of 35 % Na₂CO₃ solution and diluted to 10 ml with distilled water. The mixture was shaken well and kept at room temperature for 30 min. Absorbance for test and standard solutions were measured against the blank at 725 nm with an UV-Vis spectrophotometer. A set of reference standard solutions of Gallic acid (20, 40, 60, 80 and 100 µg/ml) were prepared in the same manner as described earlier²⁰.

2.5.5 Determination of Total Phenolics

The total phenolics was determined by Folin-Ciocalteu assay method. About 1 ml of extract was added with 9 ml of distilled water & 1 ml of Folin-Ciocalteu phenol

reagent and shaken well. After 5 minutes, 10 ml of 7 % Na₂CO₃ solution was treated to the mixture and the volume were made up to 25 ml. Incubated for 90 min at room temperature and the absorbance for test and standard solutions were determined against the reagent blank at 550 nm with an UV-Vis spectrophotometer. A set of standard solutions of Gallic acid (20, 40, 60, 80 and 100 µg/ml) were prepared in the same manner as described earlier²¹.

2.5.6 GC-MS Analysis

The identification and quantification of basil chemical constituents were evaluated by Gas chromatography coupled with Mass spectrometry QP2010 plus, Shimadzu, Japan equipped with RTX-5 MS GC capillary column (5% diphenyl/ 95% dimethyl polysiloxane) of 0.5 µm dia and 30 m length. GC working conditions: The temperature was kept between 40- 290° C with a gradual increase of 8° C/min. Column oven and injection temperatures were set at 100° C and 270° C respectively. Injection mode was set as split with a ratio of 20; Helium was used as carrier gas (mobile phase) with a flow rate of 1ml/min. MS working conditions: ion source and interface temperatures were set at 200 and 260° C. Solvent cut-off time was set as 4 min and detector voltage was set at 0.1 kV. Injection conditions: 1 µL injection volume; 10 µL injection syringe; injection temperature at 240° C; mass range at 20–300 m/z. The analytes were matched with the NIST and

Table 1. Qualitative determination of chemical constituents of *Ocimum basilicum* Leaf extracts

S.No	Chemical constituents	Test name	PE extract	CF extract	EA extract	MeOH extract	Aq extract
1.	Alkaloids	Wagner's test	+	+	+	++	+
		Mayer's test	+	+	+	++	++
2.	Flavonoids	Sodium hydroxide test	+	+	+	+	+
3.	Terpenoids	Copper acetate test	-	+	+	+	-
4.	Carbohydrates	Molisch's test	-	+	-	+	+
5.	Proteins	Biuret's test	-	+	-	++	+
		Millon's test	+	-	+	+	-
6.	Amino acids	Ninhydrin test	+	+	+	+	+

7.	Fats and oils (Fixed)	Spot test	++	+	-	+	-
		Saponification	-	-	-	-	+
8.	Steroids	Salkowski Test	+	+	+	+	+
9.	Cardiac glycosides	KellarKillani's test	+	+	+	+	+
10.	Tannins and Phenolics	Ferric chloride test	+	-	+	+	+
		Lead acetate test	+	+	++	++	+
	Saponins	Foam test	+	-	+	++	++

Note: PE- Petroleum ether; CF- Chloroform; EA- Ethyl acetate; MeOH- Methanol; Aq- Aqueous; ++ denotes relatively high; + denotes present; - denotes not detected.

Wiley library for the similar hits of the basil chemical compositions²²⁻²⁴.

3. Results and Discussion

3.1 Qualitative Determination

Table 1 represents the presence of basil chemical constituents in various extracts such as petroleum ether, chloroform; ethyl acetate; methanol; aqueous. PE and EA extracts showed various class of secondary metabolites than CF extract but lesser than MeOH and aq extracts. Alkaloids, flavonoids, steroids, cardiac glycosides and phenolics were present in almost all the extracts. Comparatively, MeOH extract showed various chemical constituents of basil that the other extracts.

3.2 Quantitative Determination

Table 2 represents the quantitative determination of chemical constituents of *Ocimum basilicum* L leaf extracts in respect to total alkaloids, flavonoids, phenols, tannins and saponins content. Alkaloids, flavonoids, phenols and saponins were highly present in MeOH but saponin content was low. Comparatively, MeOH extracts showed more quantity of chemical constituents amongst all the extracts, so it was taken to the GCMS quantification. The values here are mentioned in percentage by converting mg/standard gram equivalent.

3.3 GC-MS Analysis

Figure 1 and Table 3 represents the chromatogram and percentage composition of chemical constituents

Table 2. Quantitative determination of chemical constituents of *Ocimum basilicum* Leaf extracts

S.No	Extracts	Quantitative determination of chemical constituents (%)				
		Alkaloids	Flavanoids	Saponins	Tannins	Phenols
1.	PE extract	1.85±0.5	10.98±0.2	7.8±0.9	5.25±0.25	18.25±0.35
2.	CF extract	0.1±0.05	11.58±1.8	7.0±1.65	4.1±0.65	17.5±0.85
3.	EA extract	0.8±0.2	8.39±0.6	6.15±0.25	3.0±0.45	20.2±0.25
4.	MeOH extract	2.5±0.25	16.35±1.5	8.5±0.8	2.5±0.35	22.5±0.85
5.	Aq extract	1.25±0.8	15.6±0.5	7.65±0.95	3.2±0.85	14.02±0.95

n= 3, values are in percentage, SEM

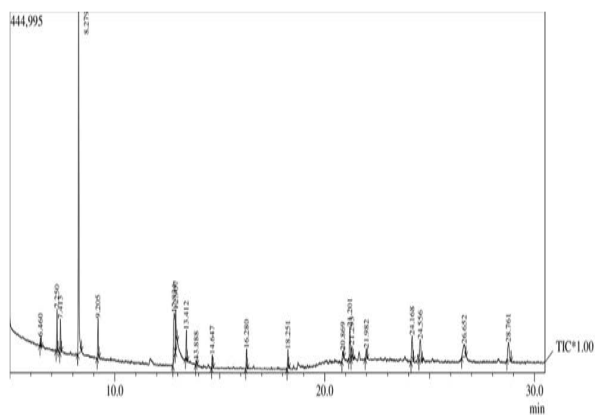


Figure 1. GCMS chromatogram of methanolic extract of *Ocimum basilicum* leaves.

of methanol leaf extract of basil. The peaks in the chromatogram were compared with the Wiley and NIST database of spectrum installed with GCMS library. The results revealed the presence of chemical composition with area % and m/z ratio with respect to their retention times.

Each component of the composition possesses unique retention times which vary in peak areas. The most abundant components in the methanol leaf extract were found to be eugenol, germacrene, β -elemene, gurjunene and menthol with higher peak area percentage. Mostly terpene alcohols and hydrocarbons were detected, and sesquiterpene also with considerable concentrations.

Table 3. Quantitative determination of chemical constituents of *Ocimum basilicum* Leaf MeOH extract by Gas chromatography Mass spectrometry

Peak	Retention time	Area %	Name
1	6.460	0.59	MYCRENE
2	7.250	3.79	EUCALYPTOL (1,8-CINEOLE)
3	7.413	3.16	1,3,6-OCTATRIENE, 3,7-DIMETHYL-, (E)-
4	8.279	38.14	EUGENOL
5	9.205	3.88	BICYCLO[2.2.1]HEPTAN-2-ONE, 1,7,7-TR
6	12.834	10.70	GERMACRENE
7	12.907	5.01	CYCLOHEXANEETHANOL,4-ETHENYL-
8	13.412	2.96	BICYCLO[7.2.0]UNDEC-4-ENE,4,11,11-TR
9	13.888	0.44	1-PROPENE, 3-BROMO-
10	14.647	1.29	2,3,5,8,10,11-HEXAMETHYLENEDODECA
11	16.280	1.91	BETA ELEMENE
12	18.251	1.82	9-DODECEN-1-OL, (Z)-
13	20.869	1.28	BIS-(3,5,5-TRIMETHYLHEXYL) ETHER
14	21.201	4.16	GURJUNENE
15	21.293	2.04	1,5-Dimethyl-1-vinyl-4-hexenyl butyrate
16	21.282	1.26	LINALOOL FORMATE
17	24.168	4.39	MENTHOL
18	24.556	4.31	METHYL(Z) CINNAMATE
19	26.652	4.29	1,5-Pentanediol,O,O'-di(propargyloxycarbony
20	28.761	4.57	1-PENTANONE, 3-[4-(DIPHENYLMETHY

4. Conclusion

Ocimum basilicum L. leaf extracts were tested for the presence of medicinally important chemical constituents. The qualitative and quantitative determination of chemical constituents of basil leaf extracts were evaluated. GCMS analysis reveals the exact composition with m/z ratio in respect to retention time. The results revealed that the methanol (MeOH) is the most suitable solvent to extract the basil leaf essential oil with high yield of chemical constituents. Further investigation will be helpful in the isolation and characterization of these bio-active compounds and by exploring their pharmacological activities will lead to the development of new therapeutic drugs.

5. Acknowledgement

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6. Conflicts of Interests

The authors declare there is no any conflicts of interests.

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