



## Newly Developed Highly Sensitive Method for the Determination of Capecitabine by Using UV-Spectroscopy

Manas Ranjan Mishra\*, Punam Agrawal, Surya Narayan Das

Gayatri College of Pharmacy, Sambalpur, Odisha-768200, India

Copyright © 2019 Manas Ranjan Mishra *et al.* This is an open access article distributed under the terms of the Creative Commons Attribution-NonCommercial-ShareAlike 4.0 International License which allows others to remix, tweak, and build upon the work non-commercially, as long as the author is credited and the new creations are licensed under the identical terms.

### ABSTRACT

Capecitabine is a 'pro-drug' to the cytotoxic agent 5-fluorouracil (5-FU) intended to administered orally. Capecitabine is generally used as first line monotherapy for advanced colon cancer. Simple, rapid, accurate UV spectrophotometric methods were developed in the present study and validated for the estimation of Capecitabine in bulk and its formulations as per ICH guidelines. Three solvent systems viz., 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) was tried. The results suggest that the developed method shows linearity over the range of concentration 2-24 $\mu$ g/ml and a correlation coefficient of 0.9999. Accuracy, precision, linearity, robustness, and ruggedness were statistically validated as per ICH guidelines for all the developed methods. The % RSD values for validated methods were found to be less than 1.5 and methods will find application in routine analysis of drug formulations containing Capecitabine.

**Keywords:** Capecitabine, UV-spectrophotometric, 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3).

DOI: 10.25004/IJPSDR.2019.110304

Int. J. Pharm. Sci. Drug Res. 2019; 11(3): 91-97

\*Corresponding author: Dr. Manas Ranjan Mishra

Address: Gayatri College of Pharmacy, Sambalpur, Odisha-768200, India

Tel.: +91-9438643438

E-mail ✉: mishra053@gmail.com

Relevant conflicts of interest/financial disclosures: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

Received: 21 February, 2019; Revised: 09 April, 2019; Accepted: 19 May, 2019; Published: 25 May, 2019

### INTRODUCTION

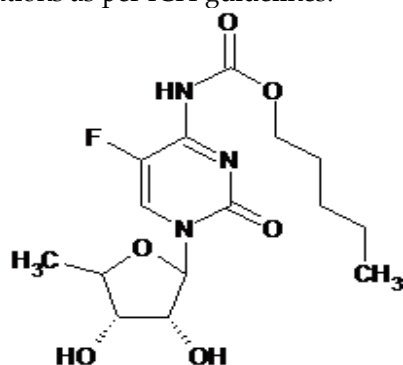
The drug is official in the Indian Pharmacopoeia (IP). [1-2] Capecitabine ( $C_{15}H_{22}FN_3O_6$ ) is 5-deoxy-5-fluoro-N-(pentyloxy) carbonyl]-cytidine with molecular weight 359.35. [3-4] The synthesis of thymidine monophosphate inhibited by 5-FU, thymidine monophosphate is an active form of thymidine which is required for *de novo* synthesis of DeoxyRibo Nucleic acid (DNA).

Capecitabine is an orally administered chemotherapeutic agent used in the treatment of metastatic breast and colorectal cancers and is a prodrug of 5'-deoxy-5- fluorouridine (5'-DFUR), which

is enzymatically converted to 5-fluorouracil in the tumor, where it inhibits DNA synthesis and slows growth of tumor tissue. The activation of Capecitabine follows a pathway with three enzymatic steps and two intermediary metabolites, 5'- deoxy-5-fluorocytidine (5'-DFCR) and 5'-deoxy-5- fluorouridine (5'-DFUR), to form 5-fluorouracil. Chemically it is 5'-deoxy-5-fluoro-N-[(pentyloxy) carbonyl] - cytidine (figure 1) with empirical formula  $C_{15}H_{22}FN_3O_6$  and the molecular weight of 359.35 g/mol. [5]

A very few methods appeared in the literature for the assay of Capecitabine in biological fluids and

pharmaceutical formulations viz., HPLC [6-9], LC-UV [10-11], LC-MS [12], and LMS/MS [13] methods. Hence in the present work it was aimed to develop and validate accurate, precise, simple and rapid UV spectroscopic methods for the estimation of Capecitabine in bulk and its formulations as per ICH guidelines.



pentyl [1-[(3R,4S)-3,4-dihydroxy-5-methyltetrahydrofuran-2-yl]-5-fluoro-2-oxo-1,2-dihydropyrimidin-4-yl]carbamate

Fig. 1: Chemical structure of Capecitabine

## MATERIALS AND METHODS

Capecitabine gift sample was obtained. Procured methanol, hydrochloric acid and sodium hydroxide from S.D Fine chemicals, Mumbai and double distilled water were used throughout the experiments. The other chemicals used were of analytical grade.

### METHODS

#### Determination of absorption maxima ( $\lambda_{max}$ )

#### Preparation of Capecitabine standard stock solution (1000 $\mu$ g/ml)

About 50 mg of Capecitabine working standard weighed accurately and transferred to a 50 ml volumetric flask. To it 40 ml of 0.1N NaOH was added and was shaken for 5 minutes to dissolve and the volume was made up to 50 ml with 0.1N NaOH. Similarly a standard stock solution was prepared in 0.1N HCl and Methanol: Water (1:3) solvent systems.

#### Preparation of Capecitabine sample solution

Transferred aliquots of standard stock solution into a series of 10 ml volumetric flask and diluted with 0.1N NaOH to get desired concentrations. Similarly sample solutions were prepared from above stock solution. UV scanning was done for the sample solutions in the range of 200-380 nm using double beam UV Spectrophotometer and absorption maxima of Capecitabine was determined.

#### Determination of linearity range

Standard solutions of Capecitabine in the concentration range of 4-40 $\mu$ g/ml were prepared in 0.1N NaOH and absorbance was measured at 292.8nm taking the 0.1N NaOH as the blank. Similarly absorbance of Capecitabine in the concentration range of 4-40 $\mu$ g/ml in 0.1N HCl, and Methanol: Water (1:3) were measured at 304 nm and 300.8 nm respectively using 0.1N HCl, and Methanol: Water (1:3) solvent systems as blank.

#### Calibration curve

Appropriate aliquots from standard stock solutions of Capecitabine were transferred to series of 10 ml

volumetric flasks. The volume was adjusted to the mark with 0.1N NaOH to obtain concentrations of 4, 8, 12, 16, 20 and 24 $\mu$ g/ml and the absorbance was measured at 292.8 nm. Similarly a set of same concentrations were prepared in 0.1N HCl and Methanol: Water (1:3). The absorbance was Measure at 304 nm and 300.8 nm respectively against respective solvent systems as blank. The concentration vs absorbance values were plotted and interpreted.

#### Accuracy Validation

The accuracy was evaluated by applying the proposed methods to the analysis formulations with known amounts of drug. The accuracy was calculated with respect to percentage of the drug recovered from the formulations.

#### Capecitabine standard solution preparation (for bulk)

Accurately weighed 50 mg of Capecitabine working standard was transferred to a 50 ml volumetric flask, about 40 ml of 0.1N NaOH was added to dissolve. Diluted up to the mark with 0.1N NaOH and mixed. To get desired concentrations, aliquots of stock solution were further diluted with 0.1N NaOH. Similarly standard solutions were prepared in other solvent systems viz., 0.1N HCl, Methanol: Water (1:3).

#### Capecitabine sample preparation (for tablets)

Accurately weighed 5 tablets were grounded in a mortar and transferred the equivalent to 50 mg of Capecitabine into a 50 ml volumetric flask. Added 40 ml of 0.1N NaOH and was shaken it for 1 h. Diluted to volume with 0.1N NaOH and the contents was mixed and filtered through 0.45 $\mu$ m membrane filter. Aliquots of the filtrate were transferred to 25 ml volumetric flask and diluted to volume with the 0.1N NaOH to get desired concentration. Similarly sample solutions were prepared in other solvent systems viz., 0.1N HCl, Methanol: Water (1:3). Recovery studies were carried out by adding known amount of standard drug (40% and 20%) to the sample solution. The absorbance measured and the amount was calculated from the calibration curve. The recovery percentage was calculated in terms of RSD percentage which was less than 2%.

#### Precision

The precision was determined by intraday and inter day observations. Repeatability was evaluated assaying 3 determinations at the same concentration (10 $\mu$ g/ml), during the same day, under the same experimental conditions. Intermediate precision was analyzed comparing the assays in 3 determinations at the same concentration (10 $\mu$ g/ml) during 3 different days. Precision (repeatability and intermediate precision) was expressed as relative standard deviation (RSD).

#### Sample preparation (for tablets)

Accurately weighed 5 tablets were grounded in a mortar and transferred equivalent to 50 mg of Capecitabine into a 50 ml volumetric flask, 40 ml of 0.1N NaOH was added and shaken for 1 h. Diluted to volume with 0.1N NaOH. The contents mixed and were filtered through 0.45 $\mu$ m membrane filter. Aliquots of

the filtrate was transferred to 25 ml volumetric flask and diluted to volume with the 0.1N NaOH to get desired concentration. Similarly sample preparations were prepared in other solvent systems viz., 0.1N HCl, Methanol: Water (1:3).

Intraday precision was determined by for three times in the same day analyzing Capecitabine (morning, afternoon, evening) at respective absorption maxima using respective solvent systems. Interday precision was determined by analyzing daily once (morning) for three days at respective absorption maxima using respective solvent systems. The percentage RSD values were calculated and it should be less than 2%.

#### LOD and LOQ

LOD/LOQ parameters are always useful to demonstrate that the analysis is being conducted in a region which is above the LOQ value. The LOD and LOQ were calculated based on the standard deviation of the response (y intercepts of regression lines) and the slope using three independent analytical curves, as denoted by ICH.

$$\text{LOD } (\mu\text{g/ml}) = 3.3 \times \frac{\sigma}{s}$$

$$\text{LOQ } (\mu\text{g/ml}) = 10 \times \frac{\sigma}{s}$$

Where  $\sigma$  - Standard deviation of the response;  $s$  - Slope ratio curve.

The lowest possible concentration where the drug Capecitabine shows response was determined in all the solvent systems. The absorbance at this concentration was measured in triplicate in respective solvent systems at respective absorption maxima. The LOD/LOQ was calculated by using formulae from the data obtained.

#### Robustness

Robustness of the proposed methods was determined by the analysis of samples and standard solutions (10 $\mu\text{g/ml}$ ) at different wavelengths ( $\pm 5$  nm) and at different solution temperatures (refrigeration condition 2-8°C and 37°C). The stability study was performed maintaining the drug working solution in respective solvent systems for 48 h protected from light, looking for the decrease of absorbance compared with those of freshly prepared solutions to assess the stability of drug.

Appropriate concentrations of Capecitabine from bulk and formulations were prepared in respective solvent systems. Analysis was carried out at three different wavelengths (actual and  $\pm 5$  nm). Amount found was calculated at three different wavelengths in terms of percentage RSD and values were less than 2%.

#### Ruggedness

Ruggedness is not addressed in the ICH documents; it is a measure of reproducibility of test results under normal, expected operational conditions from analyst to analyst and instrument to instrument.

Appropriate concentrations of Capecitabine from bulk and formulations were prepared in respective solvent systems. Analysis was carried out by two different

analysts and also two instruments. Amount found was calculated at three different wavelengths in terms of percentage RSD and values were less than 2%.

**Table 1: Linearity range curve of Capecitabine in 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) solvent systems.**

Conc. ( $\mu\text{g/ml}$ )	0.1N NaOH	0.1N HCl	Methanol: Water (1:3)
	Absorbance* $\pm$ SD	Absorbance* $\pm$ SD	Absorbance* $\pm$ SD
4	0.0672 $\pm$ 0.0004	0.0677 $\pm$ 0.0003	0.0670 $\pm$ 0.003
8	0.118 $\pm$ 0.0021	0.123 $\pm$ 0.0021	0.124 $\pm$ 0.0022
12	0.178 $\pm$ 0.0016	0.183 $\pm$ 0.0016	0.188 $\pm$ 0.0016
16	0.237 $\pm$ 0.0010	0.242 $\pm$ 0.0010	0.243 $\pm$ 0.0011
20	0.302 $\pm$ 0.0049	0.307 $\pm$ 0.0049	0.308 $\pm$ 0.0048
24	0.362 $\pm$ 0.0020	0.368 $\pm$ 0.0020	0.369 $\pm$ 0.0021
28	0.382 $\pm$ 0.0011	0.421 $\pm$ 0.0011	0.387 $\pm$ 0.0011
32	0.412 $\pm$ 0.0032	0.441 $\pm$ 0.0032	0.412 $\pm$ 0.0032
36	0.472 $\pm$ 0.0023	0.473 $\pm$ 0.0023	0.462 $\pm$ 0.0023
40	0.512 $\pm$ 0.0033	0.511 $\pm$ 0.0033	0.489 $\pm$ 0.0023

**Table 2: Calibration curve data of Capecitabine in 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) solvent systems served absorption maxima.**

Conc. ( $\mu\text{g/ml}$ )	0.1 N NaOH	0.1 N HCl	Methanol: Water (1:3)
	Absorbance* $\pm$ SD	Absorbance* $\pm$ SD	Absorbance* $\pm$ SD
0	0.0000 $\pm$ 0.0000	0.0000 $\pm$ 0.0000	0.0000 $\pm$ 0.0000
4	0.0622 $\pm$ 0.0012	0.0658 $\pm$ 0.0015	0.0637 $\pm$ 0.0018
8	0.1180 $\pm$ 0.0015	0.1240 $\pm$ 0.0019	0.1230 $\pm$ 0.0026
12	0.1780 $\pm$ 0.0019	0.1840 $\pm$ 0.0021	0.1830 $\pm$ 0.0029
16	0.2370 $\pm$ 0.0021	0.2470 $\pm$ 0.0024	0.2460 $\pm$ 0.0032
20	0.2970 $\pm$ 0.0026	0.3080 $\pm$ 0.0026	0.3070 $\pm$ 0.0023
24	0.3560 $\pm$ 0.0018	0.3690 $\pm$ 0.0021	0.3680 $\pm$ 0.0019

**Table 3: Statistical data of calibration curve for Capecitabine in 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) solvent systems.**

Parameters	0.1 N NaOH	0.1 N HCl	Methanol: Water (1:3)
$\lambda_{\text{max}}$ (nm)	292.8	304.0	300.8
Beer's range ( $\mu\text{g/ml}$ )	2-24 $\mu\text{g/ml}$	2-24 $\mu\text{g/ml}$	2-24 $\mu\text{g/ml}$
Molar absorptivity (mol <sup>-1</sup> cm <sup>-1</sup> )	1.714 $\times$ 10 <sup>4</sup>	1.716 $\times$ 10 <sup>4</sup>	1.711 $\times$ 10 <sup>4</sup>
<b>Best fit values</b>			
Slope	0.0147 $\pm$ 0.000057	0.0153 $\pm$ 0.000050	0.01531 $\pm$ 0.000076
Y-intercept when X=0.0	0.00082 $\pm$ 0.00082	0.00078 $\pm$ 0.00072	0.001714 $\pm$ 0.0011
X-intercept when Y=0.0	-0.05554	-0.05135	-0.1120
1/Slope	67.61	65.36	65.33
<b>95% CI</b>			
Slope	0.01464 to 0.01494	0.01517 to 0.01543	0.01511 to 0.01550
r <sup>2</sup>	0.9999	0.9999	0.9999
P value	< 0.0001	< 0.0001	< 0.0001

## RESULTS AND DISCUSSION

Simple, rapid, economic, accurate, precise and sensitive UV spectrophotometric methods were developed and validated as per ICH guideline and USP 2000 for the estimation of Capecitabine in bulk and formulations. Three different solvent systems viz., 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) was selected. The developed methods were further validated for

accuracy, precision, LOD, LOQ, specificity, robustness and ruggedness with statistical data.

**Table 4:** Data showing accuracy of Capecitabine (bulk) in all solvent systems.

Method I 0.1 N NaOH			
Sample No	Concentration of Capecitabine (µg/ml)		% of recovery
	Theoretical	Experimentally	
1	2	2	100
2	4	3.9	98.7
3	6	6	100
4	8	8.04	100.5
5	10	9.97	99.7
Method II 0.1 N HCl			
Sample No	Concentration of Capecitabine (µg/ml)		% of recovery
	Theoretical	Experimentally	
1	2	2.01	100.5
2	4	4.02	100.5
3	6	5.9	99.2
4	8	8.0	100
5	10	10.2	100.2
Method III Methanol : Water (1:3)			
Sample No	Concentration of Capecitabine (µg/ml)		% of recovery
	Theoretical	Experimentally	
1	2	2.02	101.1
2	4	3.9	99.4
3	6	6.03	100.5
4	8	8.04	100.5
5	10	10.06	100.6

**Table 5:** Data showing recovery studies of capecitabine (formulations) in all solvent systems.

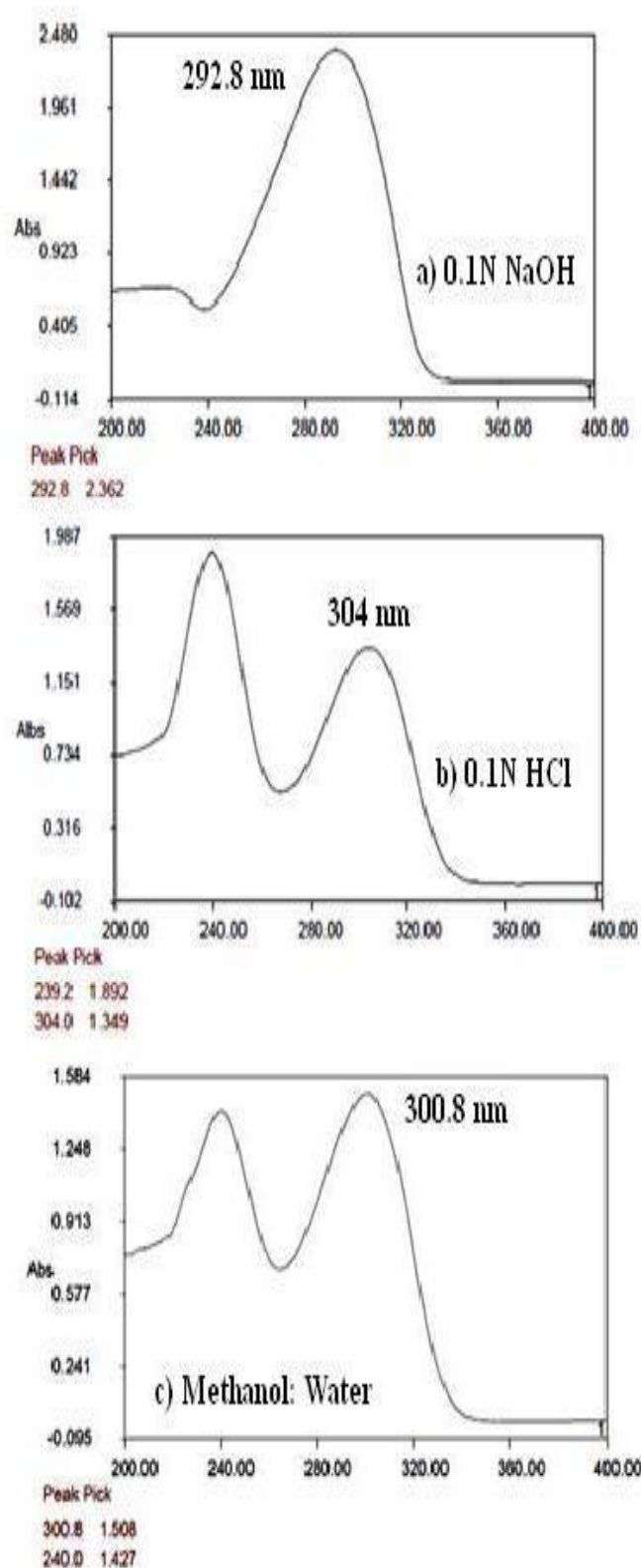
Amount present in formulation (µg/ml)	Amount added		Amount recovered (µg/ml)	Mean % Recovery	RSD
	µg	%			
10	4	40	13.9	99.2 ± 1.909	1.924
	2	20	11.8	98.7 ± 1.779	1.802
Amount present in formulation (µg/ml)	Amount added		Amount recovered (µg/ml)	Mean % Recovery	RSD
	µg	%			
10	4	40	14.03	100.2 ± 1.808	1.824
	2	20	12.03	100.3 ± 2.219	2.102
Amount present in formulation (µg/ml)	Amount added		Amount recovered (µg/ml)	Mean % Recovery	RSD
	µg	%			
10	4	40	14.1	100.9 ± 2.707	1.691
	2	20	12.1	101.1 ± 3.143	1.130

**Table 6:** Data showing LOD/LOQ of Capecitabine in all solvent systems.

Method I 0.1 N NaOH		
	Mean ± SD	SEM
Limit of detection	0.191 ± 0.046	0.026
Limit of quantitation	0.584 ± 0.145	0.084
Method II 0.1 N HCl		
	Mean ± SD	SEM
Limit of detection	0.48 ± 0.047	0.027
Limit of quantitation	1.46 ± 0.148	0.084
Method III Methanol : Water (1:3)		
	Mean ± SD	SEM
Limit of detection	0.51 ± 0.134	0.072
Limit of quantitation	1.54 ± 0.384	0.224

The absorption maxima (λ<sub>max</sub>) with characteristic peak for capecitabine were found at 292.8 nm, 304.0 nm and 300.8 nm for 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) respectively. These absorption maxima

were used to determine the linearity and it was shown linear relationship with correlation coefficient of 0.9999; 0.9999 and 0.9999 for 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) respectively in the concentration range of 2-24µg/ml. The spectra and data were shown in figure 2, 3 and table 1.



**Fig. 2:** Absorption maxima of Capecitabine in 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) solvent systems.

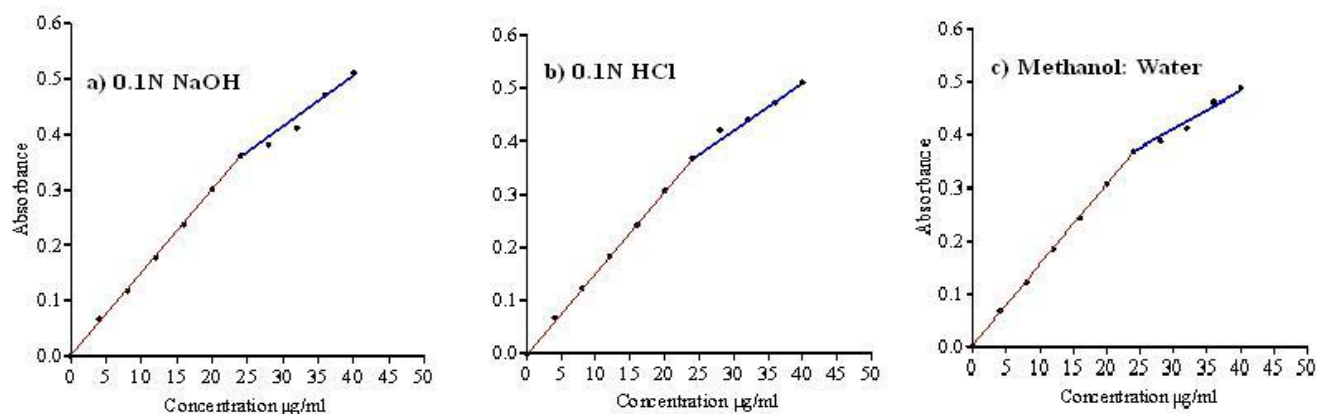


Fig. 3: Linearity range of Capecitabine in 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) solvent systems

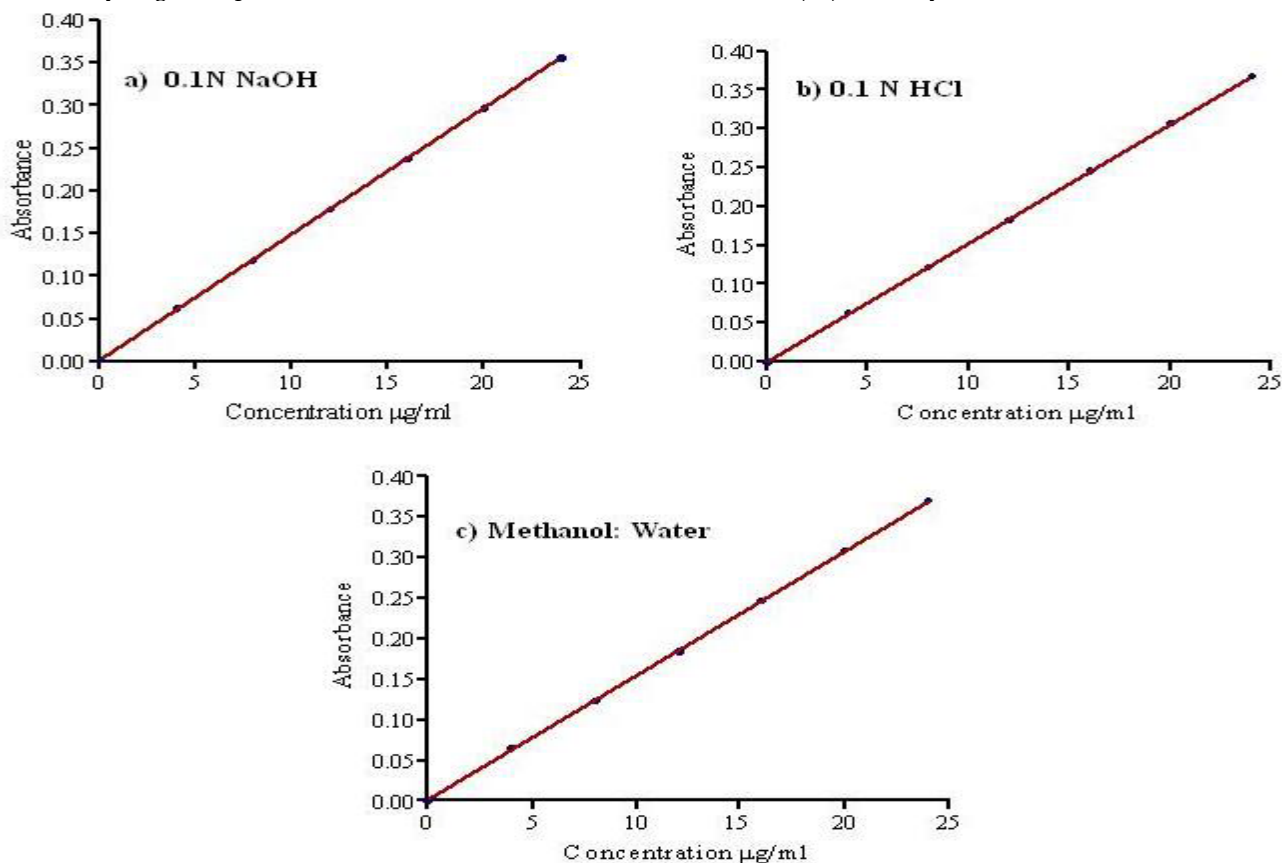


Fig. 4: Calibration curve of Capecitabine in 0.1N NaOH, 0.1N HCl and Methanol: Water (1:3) solvent systems.

Table 7: Data showing precision Intraday and Inter day trials with RSD values for Capecitabine in all solvent systems.

Method I 0.1 N NaOH						
Trials	Label claimed (mg/tab)	Amount found (mg/tab)	% Label claimed (Mean ± SD)	SEM	RSD	
Day-1	50	50.3	100.6 ± 1.246	0.729	1.239	
		50.2	100.4 ± 0.853	0.429	0.843	
		49.5	99.0 ± 0.642	0.373	0.654	
Day-2	50	50.1	100.2 ± 1.474	0.851	1.473	Intra day trials
		49.2	98.4 ± 1.060	0.611	1.062	
		49.6	99.2 ± 1.385	0.860	1.388	
Day-3	50	50.5	101.0 ± 0.692	0.400	0.686	
		50.01	100.02 ± 0.961	0.200	0.959	
		49.7	99.4 ± 0.473	0.272	0.473	
Method II 0.1 N HCl						
Day-1	50	50.4	100.8 ± 1.381	0.800	1.378	
		50.3	100.6 ± 0.765	0.441	0.758	
		50.2	100.4 ± 1.024	0.592	1.016	Intra day trials
50.4	100.8 ± 0.662	0.384	0.658			
Day-2	50	50.2	100.4 ± 1.204	0.693	1.196	
		50.3	100.6 ± 0.670	0.393	0.678	

Day-3	50	50.2		100.4 ± 0.201		0.155	0.199
		49.9		99.8 ± 1.386		0.860	1.388
		49.6		99.2 ± 0.871		0.504	0.875
<b>Method III Methanol : Water (1:3)</b>							
Day-1	50	49.9		99.8 ± 0.722		0.416	0.718
		49.7		99.4 ± 0.791		0.458	0.796
		49.9		99.8 ± 0.792		0.458	0.803
Day-2	50	50.6	Intra day trials	101.2 ± 1.155	Intra day trials	0.665	1.156
		50.2		100.4 ± 1.363		0.913	1.575
		49.8		99.6 ± 0.757		0.437	0.760
Day-3	50	50.2		100.4 ± 1.22		0.705	1.213
		49.8		99.6 ± 0.756		0.437	0.754
		49.7		99.4 ± 1.151		0.664	1.156

**Table 8: Data showing robustness of Capecitabine at different wavelengths in all solvent systems.**

Method	Conc. (µg/ml)	Wave length	Amount found	Mean (% ± SD)	SEM	RSD
Method I	10	292	9.95	99.5 ± 0.962	0.552	0.905
		297	8.67	86.7 ± 1.735	1.012	1.817
		287	8.7	87.0 ± 1.013	0.601	1.163
Method II	10	304	9.93	99.3 ± 0.712	0.407	0.732
		309	8.7	87.0 ± 1.507	0.928	1.720
		299	8.8	88.0 ± 1.059	0.665	1.241
Method III	10	300	9.98	99.8 ± 1.241	0.721	1.256
		305	8.34	83.4 ± 1.122	0.702	1.332
		295	8.41	84.1 ± 0.941	0.562	1.076

**Table 9: Data showing ruggedness of Capecitabine by different Analysts in all solvent systems..**

Method	Conc. (µg/ml)	Analyst	Amount found	Recovery ± SD	SEM	RSD
Method I	10	Analyst 1	10.01	100.1 ± 0.54	0.223	0.539
		Analyst 2	9.94	99.4 ± 0.3	0.122	0.301
Method II	10	Analyst 1	9.99	99.9 ± 0.75	0.307	0.756
		Analyst 2	10.05	100.5 ± 0.565	0.231	0.565
Method III	10	Analyst 1	10.01	100.1 ± 0.41	0.169	0.413
		Analyst 2	9.99	99.9 ± 0.393	0.161	0.612

**Table 10: Data showing ruggedness of Capecitabine by using different Instruments in all solvent systems**

Method	Conc. µg/ml)	Instrument	Amount found	Recovery (± SD)	SEM	RSD
Method I	10	Instrument 1	9.92	99.2 ± 0.963	0.554	0.964
		Instrument 2	10.01	100.1 ± 0.493	0.284	0.492
Method II	10	Instrument 1	9.7	97.0 ± 0.709	0.409	0.733
		Instrument 2	9.71	97.1 ± 0.642	0.371	0.658
Method III	10	Instrument 1	9.92	99.2 ± 1.249	0.721	1.251
		Instrument 2	9.89	98.9 ± 0.600	0.346	0.610

**Table 11: Robustness of capecitabine at refrigerated condition and room temperature**

Trial	Label Claim (mg/tab)	Refrigerated Condition					Room Temperature				
		Amount will be found (mg/tab)	% of Label Claim Mean ± SD	SEM	RSD	Amount will be found (mg/tab)	% of Label Claim Mean ± SD	SEM	RSD		
Method-I	Day-1	49	98 ± 0.341	0.201	0.344	50.3	100.6 ± 1.249	0.727	1.233		
		49.1	98.2 ± 0.350	0.208	0.358	50.4	100.8 ± 0.850	.424	.42		
		48.8	97.6 ± 0.352	0.202	0.350	49.8	99.6 ± 0.645	.373	.654		
	Day-2	48.8	97.6 ± 0.351	0.201	0.350	50.02	100.04 ± 1.474	.852	1.474		
		48.4	96.8 ± 0.251	0.145	0.251	49.9	99.8 ± 1.061	.611	1.062		
		48.5	99.0 ± 0.253	0.143	0.253	49.9	99.8 ± 1.383	.850	1.382		
Day-3	48.6	97.2 ± 0.513	0.288	0.522	50.4	100.8 ± 0.692	.400	.684			
	49	98.0 ± 0.251	0.145	0.253	50.03	100.06 ± 0.96	.201	0.959			
	49.1	98.2 ± 0.288	0.161	0.292	49.8	99.6 ± 0.473	.273	.473			
Method-II	Day-1	49	98 ± 0.603	0.348	0.600	50.3	100.6 ± 1.387	.800	1.378		
		49	98 ± 0.452	0.260	0.449	50.2	100.4 ± 0.763	.411	.758		
		49.1	98.2 ± 0.360	0.208	0.361	50.1	100.2 ± 1.026	.592	1.016		
	Day-2	49.1	98.2 ± 0.351	0.202	0.350	50.3	100.6 ± 0.665	.384	.658		
		49.2	98.4 ± 0.451	0.260	0.451	50.1	100.2 ± 1.201	.693	1.196		
		49.1	98.2 ± 0.360	0.206	0.362	50.1	100.2 ± 0.681	.393	.678		
Day-3	49.4	98.8 ± 0.450	0.260	0.453	50.1	100.2 ± 0.200	.155	.199			
	49	98 ± 0.556	0.321	0.560	49.9	99.8 ± 1.386	.860	1.388			
	49.1	98.2 ± 0.360	0.20	0.363	49.8	99.6 ± 0.875	.504	0.873			
Method-III	Day-1	49	98 ± 0.341	0.272	0.344	50.1	100.2 ± 0.722	.416	.718		
		49.1	98.2 ± 0.350	0.230	0.358	49.7	99.4 ± 0.794	.458	.796		
		49.2	97.6 ± 0.352	0.264	0.352	49.8	99.6 ± 0.793	.458	.804		
	Day-2	49	97.6 ± 0.351	0.176	0.350	50.4	100.8 ± 1.153	.665	1.156		
		49	96.8 ± 0.251	0.378	0.402	50.2	100.4 ± 1.365	.913	1.575		
		49.1	99.0 ± 0.253	0.324	0.462	49.8	99.6 ± 0.471	.788	1.369		
Day-3	49	97.2 ± 0.513	0.201	0.307	50.2	100.4 ± 1.22	.705	1.212			
	49	98.0 ± 0.251	0.208	0.253	48.8	99.6 ± 0.757	.437	.754			
	49.1	98.2 ± 0.288	0.202	0.288	49.8	99.6 ± 1.150	.664	1.156			

The proposed UV spectrophotometric methods were found to be simple, rapid, accurate, precise and economic. From the above data it was observed that all validation parameters met the predetermined acceptance criteria and validated in terms of linearity, accuracy, precision, reproducibility, robustness, and ruggedness as per the ICH guidelines. Thus it has been concluded that the proposed methods were validated for the analysis of capecitabine in bulk and its formulations.

## REFERENCES

1. Brunton LL, Lazo JS, Parker KL. Goodman and Gilman: The pharmacological basis of therapeutics. Edn. 11, McGraw Hill Medical Publishing Division, 2006, pp. 1404-1405.
2. Hirsch BR, Zafar SY. Capecitabine in the management of colorectal cancer. *Cancer Manag Res.* 2011; 3: 79-89.
3. Walko CM, Lindley C. Capecitabine: a review. *Clin Ther.* 2005; 27: 23-44.
4. Milano G, Schellens JHM. Pyrimidine antimetabolites. In: Schellens JHM, McLeod HL, Newell DR. eds. *Cancer Clinical Pharmacology.* Oxford: University Press, 2005; pp. 51-62.
5. Indian Pharmacopoeia, 2010, 2, pp. 972-973.
6. Guichard SM, Mayer I, Jodrell DI. Simultaneous determination of capecitabine and its metabolites by HPLC and mass spectrometry for preclinical and clinical studies. *J Chromatogr B.* 2005; 826: 232-237.
7. Dhananjeyan MR, Liu J, Bykowski C, Trendel JA, Sarver JG, Andob H, *et al.* Rapid and simultaneous determination of capecitabine and its metabolites in mouse plasma, mouse serum, and in rabbit bile by high-performance liquid chromatography. *J Chromatogr A.* 2007; 1138(1-2): 101-108.
8. Farkouh A, Ettlinger D, Schueller J, Georgopoulos A, Scheithauer W, Czejka M. A rapid and simple HPLC assay for quantification of capecitabine for drug monitoring purposes. *Anticancer Res.* 2010; 30: 5207-5212.
9. Piorkowska E, Kaza M, Fitatiuk J, Szlaska I, Pawinski T, Rudzki PJ. Rapid and simplified HPLCUV method with on-line wavelengths switching for determination of capecitabine in human plasma. *Pharmazie.* 2014; 69: 500-505.
10. Reigner B, Blesch K, Weidekamm E. Clinical pharmacokinetics of capecitabine. Springer Science+, *Clin Pharmacokinet.* 2001; 40: 85-104.
11. Zufia L, Aldaz A, Giraldez J. Simple determination of capecitabine and its metabolites by liquid chromatography with ultraviolet detection in a single injection. *J Chromatogr B.* 2004; 809: 51-58.
12. Xu Y, Grem JL. Liquid chromatography-mass spectrometry method for the analysis of the anticancer agent capecitabine and its nucleoside metabolites in human plasma. *J Chromatogr B.* 2003; 783: 273-285.
13. Siethoff C, Orth M, Ortling A, Brendel E, Redeker WW. Simultaneous determination of capecitabine and its metabolite 5-fluorouracil by column switching and liquid chromatographic/tandem mass spectrometry. *J Mass Spect.* 2004; 39: 884-889.

**HOW TO CITE THIS ARTICLE:** Mishra MR, Agrawal P, Das SN. Newly Developed Highly Sensitive Method for the Determination of Capecitabine by Using UV-Spectroscopy. *Int. J. Pharm. Sci. Drug Res.* 2019; 11(3): 91-97. DOI: 10.25004/IJPSDR.2019.110304