

## Synthesis and evaluation of some 1,2,3,4-tetra-hydro pyrimidine derivatives as antitumor agents.

### Hole MB1\*, Pattan SR<sup>2</sup>, and Vijayalakshmi P<sup>3</sup>

<sup>1</sup>Research and Development Cell, Jawaharlal Nehru Technological University, Kukatpally, Hyderabad-500085, Telangana, India., <sup>2</sup>Department of Pharmaceutical Chemistry, Abasaheb Kakade B. Pharmacy College, Bodhegaon, Tal-Shevgaon, Dist- Ahmednagar - 414503, Maharashtra, India. <sup>3</sup>Department of Pharmaceutics, Siddhartha Institute of Pharmacy, Ghatkesar, Hyderabad-501301, Telangana, India.

\*Corresponding author E-mail: mangeshhole4u@rediffmail.com | Mob No-9890837978

### **Manuscript Details**

Available online on <u>http://www.irjse.in</u> ISSN: 2322-0015

### Editor: Dr. Arvind Chavhan

### Cite this article as:

Hole MB, Pattan SR, Vijayalakshmi P. Synthesis and evaluation of some 1,2,3,4-tetra-hydro pyrimidine derivatives as antitumor agents., *Int. Res. Journal of Science & Engineering*, January 2018, Special Issue A3 : 37-42.

© The Author(s). 2018 Open Access This article is distributed under the terms of the Creative Commons Attribution 4.0 International License (<u>http://creativecommons.org/licenses/by/4.0/</u>), which permits unrestricted use, distribution, and

which permits unrestricted use, distribution, and reproduction in any medium, provided you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made.

### ABSTRACT

Sixteen [3, 6 (4- substituted aryl)1-(1,2 hydrazino-2-oxoethyl)-20x0-1, 2, 3, 4-tetrahydropyrimidine-5-yl] propionic acid derivatives have been synthesized in a four-step reaction. In the first step benzene or substituted benzene react with glutaric anhydride in presence of aluminum trichloride (Friedel Craft reaction) to obtain 5-(substituted phenyl)-5-oxo pentanoic acid. Second step involves synthesis of [4, 6 (4-substituted aryl)-2-oxo-1, 2, 3, 4- tetrahydro-pyrimine-5 yl propionic acid by reaction between 5-(substituted phenyl)-5-oxo pentanoic acid, urea and substituted aldehydes (Biginelli reaction). Third step involves reaction of [4, 6 (4-substituted aryl)-2-oxo-1, 2, 3, 4- tetrahydropyrimine-5 yl propionic acid was dissolved in a solution prepared by reacting Na (0.1mol) with 200ml of absolute ethanol and then solution is refluxed with ethyl chloroacetate to produce. 3-[4-chloro-6-(4-chlorophenyl)-1-(2-ethoxy-2-oxoethyl)-2-oxo-1,2-dihydropyrimidin-5-yl] propanoic acid. In forth step Compound (3) was treated with hydrazine hydrate to form[3, 6 (4- substituted aryl)1-(1,2hydrazino-2-oxoethyl)- 2oxo- 1,2,3, 4tetrahydro pyrimidine -5-yl] propionic acid derivatives. Their structures are confirmed by IR, 1H NMR. TLC of synthesized compounds performed in chloroform: ethanol (3:1) solvent system. All compounds were screened for their antitumor activity by trypan blue assay at. (1  $\mu$ g /mL, 10  $\mu$ g/mL, 100  $\mu$ g /mL, 1000  $\mu$ g /mL). And their results were compared with standard drug Cyclophosphamide.

**Key words:** Pyrimidine, Antitumor activity, aryl alkanoic acid.

### INTRODUCTION

In medicinal chemistry pyrimidine derivatives have been very well known for their therapeutic applications. [1] The presence of a pyrimidine base in thymine, cytosine and uracil, which are the essential binding blocks of nucleic acids, DNA and RNA is one possible reason for their activity. [2,3,4] The literature indicated that compounds having pyrimidine nucleus possess broad range of biological activities, like 5fluorouracil as anticancer; [5] idoxuridine and trifluoridine as antiviral; [6] zidovudine and stavudine as antiHIV; [7] trimethoprim, sulphamethiazine and sulphadiazine as antibacterial; [8] sulphadoxin as antimalarial and antibacterial; [9] minoxidil and prazosin as antihypertensive; [10] barbiturates e.g. phenobarbitone as sedative, [11] hypnotics and anticonvulsant; [12] propylthiouracil as antithyroid; [13] thionzylamine as H1antihistamine; [14] and toxoflavin as antibiotics.[15] Pyrimidines derivatives are undertaken with the aim of improving safety profile and reduction of GI side effects by increasing aryl chain length of acid and addition of amide group. [16,17,18]

### METHODOLOGY

The chemicals used are benzene, glutaric anhydride, sulphuric acid, aluminium chloride, aluminium hydroxide, methanol, urea, chlorobenzaldehyde, ferrous chloride, ethanol, potassium dichromate, hydrazine hydrate, ethyl chloroacetate were of LR grade and purchased from Sigma Aldrich, Mumbai, India.IR Spectra taken by using Perkin Elmer 65 FT-IR Spectrophotometer using KBr disc.1H NMR Spectra is taken by using Brucker Spectrophotometer (400 MHz) in DMSO from University of Pune Maharashtra, India. All melting points were determined in open capillaries and are uncorrected. Purity of compounds are checked by Thin layer chromatography using prepared silica gel G slides as a stationary phase and chloroform: ethanol (3:1) solvent system as mobile phase. The spots resolved were checked by UVchamber and iodine chamber.

The chemicals used are benzene, glutaric anhydride, sulphuric acid, aluminium chloride, aluminium hydroxide, methanol, urea, chlorobenzaldehyde, ferrous chloride, ethanol, potassium dichromate, hydrazine hydrate, ethyl chloroacetate were of LR grade and purchased from Sigma Aldrich, Mumbai, India.IR Spectra taken by using Perkin Elmer 65 FT-IR Spectrophotometer using KBr disc.1H NMR Spectra is taken by using Brucker Spectrophotometer (400 MHz) in DMSO from University of Pune. Maharashtra, India.

All melting points were determined in open capillaries and are uncorrected. Purity of compounds are checked by Thin layer chromatography using prepared silica gel G slides as a stationary phase and chloroform: ethanol (3:1) solvent system as mobile phase. The spots resolved were checked by UVchamber and iodine chamber.

### Step 1: Synthesis of 5-(substituted phenyl)-5-oxopentanoic acid [19]

Placed (0.00225 mol) of substituted benzene and (0.0034 mol) of glutaric anhydride in a 1 liter threenecked flask, Stirred the mixture and added (0.0075 mol) of powdered aluminium chloride all at once. The reaction usually start immediately, HCl is generated, evolved and reaction mixture become hot. If there is no apparent reaction warm gently. Heated in an oil bath to gentle refluxing with continued stirring for a half hour. Allowed to cool, immerse flask in a bath of cold water and slowly added 15 ml of water from a separating funnel. Introduced 5 ml of conc. HCl and separated benzene layer. Transferred the hot mixture to a 60 ml beaker. The 5-(substituted phenyl) -5-oxopentanoic acid separated as colourless oil, which soon solidified. Cooled in ice, filtered off acid at the pump and washed with 10 ml of cold water. Dissolved the crude acid in solution of 4g of anhydrous sodium carbonate in 25 ml water by boiling for 10-15 min. filtered the solution, suctioned to remove the small amount of aluminium hydroxide and washed with 2.5 ml portion of hot water. Treated the hot filtrate with 2g of decolourising carbon. Stirred for 5 min and filtered at the pump through a preheated buchner funnel. Transfer the filtrate to 1-liter beaker, cooled to about 50 0C and cautiously acidified with 6-7 ml of conc. HCl. Cooled to 0 0C in a freezing mixture of ice and salt. Filtered, washed thoroughly with cold water, dried for 12 hours upon filter paper and then weighed the compound.

### Step-2: Synthesis of [4, 6-(4-substituted aryl)-2thioxo- 1, 2, 3, 4-tetrahydro-pyrimidin-5-yl]propanoic acid [20]

An equimolar reaction mixture of 5-(substituted phenyl) -5-oxo- pentanoic acid (0.0022 mol) urea (0.002 mol), substituted aldehyde (0.0022 mol) and K2CO3 (0.0022 mol) in 7 ml ethanol were refluxed in oil bath for 7 hrs. The reaction mixture was cooled and the solid obtained by filtered was dissolved in hot water and filtered. The filtrate was neutralized with acetic acid. The product thus obtained was recrystallized from methanol.

### Step-3 : Synthesis of 3-[4-chloro-6-(4-chlorophenyl)-1-(2-ethoxy-2-oxoethyl)-2-oxo-1,2-dihydropyrimidin-5-yl]propanoic acid [21]

The compound 2 (0.1mol) was dissolved in a solution prepared by reacting Na (0.1mol) with 200 ml of absolute ethanol. The solution was refluxed with stirring and ethyl chloroacetate (0.1mol) was added in three portions over a period of 0.5 hr. After heating under reflux for 16 hr, the reaction mixture was filtered while hot to remove precipitated sodium chloride, the solvent was removed on a rotary vaccum evaporator. The crude product was collected and recrystallised from ethanol.

# Step-4 : Synthesis of compounds [3, 6 (4- substituted aryl)1-(1,2 hydrazino-2-oxoethyl)-20x0-1, 2, 3, 4- tetrahydropyrimidine-5-yl] propionic acid derivatives [22]

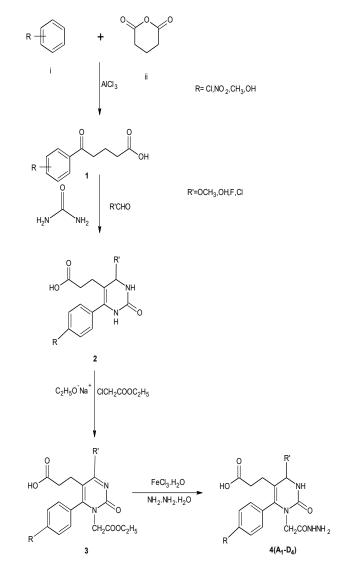
Mixture of compound 3 (0.01 mole), FeCl3.6H2O (0.02 mole) & 0.01 mole Hydrazine hydrate of was ground by pestle & mortar at room temp. The reaction mixture was digested with water. The resultant solid was filtered, washed with water & the crude material is purified by recrystallization from methanol to afford compound 4.

### ANTI TUMOR ACTIVITY: Trypan Blue Assay

### Viable Cell Counts Using Trypan Blue

Trypan Blue is a vital dye. The reactivity of trypan blue is based on the fact that the chromophore is negatively charged and does not interact with the cell unless the membrane is damaged. Therefore, all the cells which exclude the dye are viable. HL-60 cell line is taken for antitumor studies

### Scheme 1: Synthesis of 1, 2, 3, 4-Tetrahydropyrimidine derivatives



#### Procedure:

Trypan Blue Staining of Cells Place 0.5 ml of a suitable cell suspension (dilute cells in complete medium without serum to an approximate concentration of  $1 \times 105$  to  $2 \times 105$  cells per ml in a screw cap test tube. Add 0.1 ml of 0.4% Trypan Blue Stain. Mix thoroughly. Allow to stand 5 min at 15 to 30°C (room temperature). Fill a hemocytometer as for cell counting. Under a microscope, observe if non-viable are stained and viable cells excluded the stain.

Compound	R	R1	Mol. formula	Mol. wt	M.P	% yeild	R.f value (Chloroform: Ethanol)
A1	C1	н <sub>3</sub> со	C <sub>22</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>5</sub>	458.89	138.5	73.25	0.473
A2	Cl	но	$C_{21}H_{21}ClN_4O_5$	444.86	134.2	75.23	0.356
A3	Cl	F	$C_{21}H_{20}FN_4O_4$	446.85	136.9	71.58	0.436
A4	Cl	CI-	$C_{21}H_{20}Cl_2N_4O_4\\$	463.31	142.3	68.22	0.388
B1	OH	н <sub>3</sub> со	$C_{22}H_{24}N_4O_6$	440.44	133.0	76.59	0.412
B2	OH	но	$C_{21}H_{22}N_4O_6$	426.44	125.8	74.32	0.362
B3	OH	F	$C_{21}H_{21}FN_4O_5$	428.41	127.4	69.33	0.268
B4	OH	ci	$C_{21}H_{21}ClN_4O_5$	444.86	134.8	75.66	0.387
C1	CH3	н₃со	$C_{23}H_{26}N_4O_5$	438.36	131.5	81.57	0.425
C2	CH3	но	$C_{22}H_{24}N_4O_5$	424.44	123.6	69.56	0.389
C3	CH3	F	C <sub>22</sub> H <sub>23</sub> FN <sub>4</sub> O <sub>4</sub>	426.44	125.7	73.66	0.296
C4	CH3	ci	C <sub>22</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>4</sub>	442.89	134.6	78.32	0.432
D1	NO2	н <sub>3</sub> со—	C22H23N5O7	469.44	144.5	68.72	0.459
D2	NO2	но	$C_{21}H_{21}N_5O_7$	455.22	141.2	76.52	0.342
D3	NO2	F	$C_{21}H_{21}FN_5O_6$	457.41	142.6	77.25	0.267
D4	NO2	н <sub>з</sub> со	$C_{21}H_{20}ClN_5O_6$	473.86	148.3	68.74	0.496

 Table 1. Synthesized compounds

### SPECTRAL DATA

A1-IR (KBr,cm-1) :3430 (O-H)str., 3230 (N-H str.)3068 (Ar-C-H)str,2966 (C-H)str,1697(C=O)str,1650(C=O amide) str,1574 (C=C)str, 1280 (C-O)str.,1109(C-N)str. 667 (C-CI)str.,

**1H NMR (CDCl3)** :7.2-8 (8H (s) of Ar-H), 6.9 (2 H (s) of pyrimidine), 10.8 (1 H (s) of -COOH), 3.2(3 H (s) of OCH3), 1.42 (2H (t) of CH2), 2.82 (2H (t) of CH2).3.8(1H(S) of -CONH), 3.6(2H(S) of -NH2).2.1(2H (s) of CH2).

A2-IR (KBr,cm-1) :3450 (O-H)str., 3280 (N-H str.)3030 (Ar-C-H)str,2940 (CH)str,1720(C=O)str, 1650(C=O amide) str,1550(C=C)str, 1278 (C-O)str.,1115(C-N)str. 662 (C-CI)str.

A3-IR (KBr,cm-1) :3480 (O-H)str., 3250 (N-H str.)3047 (Ar-C-H)str,2957 (CH)str,1715(C=O)str, 1650(C=O amide) str,1558(C=C)str, 1275 (C-O)str.,1120(C-N)str. 662 (C-CI)str. 710(C-F)str.

A4-IR (KBr,cm-1) :3465 (O-H)str., 3278 (N-H str.)3056 (Ar-C-H)str,2950 (CH)str,1715(C=O)str, 1650(C=O amide) str,1556(C=C)str, 1270 (C-O)str.,1110(C-N)str. 662 (C-CI)str.

**B1-IR (KBr,cm-1)** :3460(O-H)str., 3245 (N-H str.)3030 (Ar-C-H)str,2960 (C-H)str,1715(C=O)str, 1650(C=O amide) str,1574 (C=C)str, 1280 (C-O)str.,1109(C-N)str. 667 (C-CI)str.,

**1H NMR (CDCl3)** :7.2-8 (8H (s) of Ar-H), 6.9 (3 H (s) of pyrimidine), 10.8 (1 H (s) of -COOH), 3.2(3 H (s) of OCH3), 1.42 (2H (t) of CH2), 2.82 (2H (t) of CH2).3.8(1H(S) of -CONH), 3.6(2H(S) of -NH2).3.9(1H(S) of -OH).

**B2-IR (KBr,cm-1)** :3450 (O-H)str., 3280 (N-H str.)3030 (Ar-C-H)str,2940 (CH)str,1720(C=O)str, 1650(C=O amide) str,1550(C=C)str, 1278 (C-O)str.,1115(C-N)str.

**B3-IR (KBr,cm-1)** :3476 (O-H)str., 3264 (N-H str.)3047 (Ar-C-H)str,2957 (CH)str,1715(C=O)str, 1650(C=O amide) str,1558(C=C)str, 1275 (C-O)str.,1115(C-N)str, 710(C-F)str.

**B4-IR** (KBr,cm-1) :3465 (O-H)str., 3278 (N-H str.)3056 (Ar-C-H)str,2950 (CH)str,1715(C=O)str, 1650(C=O amide) str,1556(C=C)str, 1270 (C-O)str.,1110(C-N)str. 660 (C-CI)str.

C1-IR(KBr,cm-1):3430 (O-H)str., 3230 (N-H str.)3068 (Ar-C-H)str,2966 (C-H)str,1697(C=O)str,1650(C=O amide) str,1574 (C=C)str, 1280 (C-O)str., 1109(C-N)str. 1H NMR (CDCl3) :7.2-8 (8H (s) of Ar-H), 6.9 (3 H (s) of pyrimidine), 10.8 (1 H (s) of -COOH), 3.2(3 H (s) of OCH3), 1.42 (2H (t) of CH2), 2.82 (2H (t) of CH2), 3.8(1H(S) of -CONH), 3.6(2H(S) of -NH2). C2-IR (KBr,cm-1) :3450 (O-H)str., 3280 (N-H str.)3030 (Ar-C-H)str,2940 (CH)str,1720(C=O)str, 1650(C=O amide) str,1550(C=C)str, 1278 (C-O)str.,1115(C-N)str. C3-IR (KBr,cm-1) :3476 (O-H)str., 3264 (N-H str.)3047 (Ar-C-H)str,2957 (CH)str,1715(C=O)str, 1650(C=O amide) str,1558(C=C)str, 1275 (C-O)str.,1115(C-N)str, 710(C-F)str. C4-IR (KBr,cm-1) :3465 (O-H)str., 3278 (N-H str.)3056 (Ar-C-H)str,2950 (CH)str,1715(C=O)str, 1650(C=O amide) str,1556(C=C)str, 1270 (C-O)str.,1110(C-N)str. 660 (C-CI)str. D1-IR(KBr,cm-1):3430(O-H)str., 3230 (N-H str.)3068 (Ar-C-H)str,2966 (C-H)str,1697(C=O)str,1650(C=O amide) str,1574 (C=C)str,1344(NO2)str, 1280 (C-O)str.,1109(C-N)str. 1H NMR (CDCl3) :7.2-8 (8H (s) of Ar-H), 6.9 (3 H (s) of pyrimidine), 10.8 (1 H (s) of -COOH), 3.2(3 H (s) of OCH3), 1.42 (2H (t) of CH2), 2.82 (2H (t) of CH2), 1.9(3H (S)of-CH3), 3.8(1H(S) of -CONH), 3.6(2H(S) of -NH2). D2-IR (KBr.cm-1) :3450 (O-H)str., 3280 (N-H str.) 3030 (Ar-C-H)str, 2940 (CH)str, 1720(C=O)str, 1650(C=O amide) str,1550(C=C)str, 1325(NO2)str, 1278 (C-O)str.,1115(C-N)str. D3-IR (KBr,cm-1) :3476 (O-H)str., 3264 (N-H str.)3047 (Ar-C-H)str,2957 (CH)str,1715(C=O)str, 1650(C=O amide) str,1558(C=C)str, 1328(NO2)str, 1275 (C-O)str.,1115(C-N)str, 710(C-F)str. D4-IR (KBr,cm-1) :3465 (O-H)str., 3278 (N-H str.)3056 (Ar-C-H)str,2950 (CH)str,1715(C=O)str, 1650(C=O amide)

**D4-IK (KBr,cm-1)** :3465 (O-H)str., 3278 (N-H str.)3056 (Ar-C-H)str,2950 (CH)str,1715(C=O)str, 1650(C=C)str,1556(C=C)str, 1344(NO2)str, 1270 (C-O)str.,1110(C-N)str. 660 (C-CI)str.

### Table 2:Antitumor activity of [4, 6 (4- substituted aryl)-2-oxo-1, 2, 3, 4-tetrahydropyrimidine- 5-yl] propionic acid derivatives

Sample	Conc. (ug/ml)	Observed Viable Cell	Total cell count	% Viability	Mean ±SEM
Control	-	89	113	78.76	
	1000	33	117	28.20	28.62 ± 0.4250*
	100	43	103	41.74	45.62 ± 3.8850
$A_1$	10	48	99	48.48	47.99 ± 0.4850
	1	55	92	59.78	65.52 ± 5.740
	1000	28	119	23.52	23.42 ± 0.0953*
•	100	36	108	33.33	32.80 ± 0.5299
$\mathbf{A}_{2}$	10	47	104	45.19	45.24 ± 0.0432
	1	59	97	60.82	59.79 ± 1.0300
	1000	26	113	24.52	24.42 ± 0.0832*
	100	33	102	34.33	35.80 ± 0.466
$A_3$	10	56	135	55.12	48.56 ± 0.0657
	1	59	97	60.82	59.79 ± 1.0300
	1000	18	112	16.07	$16.44 \pm 0.370^*$
•	100	21	126	16.66	$17.20 \pm 0.540^{*}$
$\mathbf{A}_4$	10	27	123	21.95	$21.38 \pm 0.574^*$
	1	22	112	19.64	$20.18 \pm 0.540^*$
	1000	25	112	22.32	22.97 ± 0.6550*
$B_1$	100	43	108	39.81	41.33 ± 1.5200
<b>D</b> 1	10	54	96	56.96	56.69 ± 0.2647
	1	60	89	67.41	67.04 ± 0.3746
	1000	45	116	32.56	32.09± 0.4250
<b>B</b> <sub>2</sub>	100	47	112	34.45	34.12 ± 2.8850
<b>D</b> <sub>2</sub>	10	44	95	42.76	42.65 ± 0.5640
	1	45	82	59.45	58.12 ± 3.687
	1000	33	103	32.03	32.85 ± 0.8149
B <sub>3</sub>	100	46	95	48.42	49.21± 0.7900
<b>D</b> 3	10	59	93	63.44	63.82 ± 0.3855
	1	63	85	74.41	71.27 ± 3.1400

Int. Res. J. of Science & Engineering, Special Issue A3, January, 2018:

	1000	23	108	23.42	23.32 ± 0.0562
B <sub>4</sub>	1000	34	105	33.32	$32.80 \pm 0.345$
	100	54	105	45.42	44.43± 0.0873
	10	43	98	52.82	51.79 ± 1.0546
	1 1000	25	110	22.72	$23.97 \pm 0.5640^*$
	1000	42	105	40.00	$40.33 \pm 1.5620$
	100	58	98	40.00 59.18	$40.33 \pm 1.3820$ 59.69 ± 0.1345
	10	56	98 89	62.91	$64.04 \pm 0.3987$
C <sub>2</sub>	1000	45	116	32.56	32.09± 0.4250
	100	47	112	34.45	34.12 ± 2.8850
	10	44	95	42.76	$42.65 \pm 0.5640$
	1	45	82	59.45	58.12 ± 3.687
C <sub>3</sub>	1000	18	112	16.07	$16.44 \pm 0.370^*$
	100	21	126	16.66	$17.20 \pm 0.540^*$
-3	10	27	123	21.95	$21.38 \pm 0.574^*$
	1	22	112	19.64	$20.18 \pm 0.540^*$
	1000	25	110	22.72	23.97 ± 0.5640*
C <sub>4</sub>	100	42	105	40.00	40.33 ± 1.5620
<b>C</b> 4	10	58	98	59.18	59.69 ± 0.1345
	1	56	89	62.91	64.04 ± 0.3987
	1000	18	113	16.54	$16.13 \pm 0.845^*$
$D_1$	100	27	132	20.23	$20.43 \pm 0.675^*$
$D_1$	10	30	139	23.48	$24.24 \pm 0.459^*$
	1	48	125	38.58	$38.64 \pm 0.794^*$
	1000	23	108	23.42	$23.32 \pm 0.0562$
$D_2$	100	34	105	33.32	32.80 ± 0.345
$D_2$	10	54	137	45.42	44.43± 0.0873
	1	43	98	52.82	51.79 ± 1.0546
D <sub>3</sub>	1000	25	110	22.72	23.97 ± 0.5640*
	100	42	105	40.00	40.33 ± 1.5620
	10	58	98	59.18	59.69 ± 0.1345
	1	56	89	62.91	64.04 ± 0.3987
	1000	16	110	14.54	$14.13 \pm 0.845^*$
D.	100	25	130	19.23	$19.43 \pm 0.675^*$
$D_4$	10	29	135	21.48	$22.24 \pm 0.459^*$
	1	45	123	36.58	$36.64 \pm 0.794^*$
Cyclophosphami de	1000	16	102	15.68	$16.42 \pm 0.324^*$
	100	19	113	16.81	17.32±0.583*
	10	25	120	20.83	$21.48 \pm 0.276^{*}$
	1	21	108	19.44	$20.08 \pm 0.470^*$

### **RESULTS AND DISCUSSION**

In scheme all compounds have shown moderate anticancer activity at both the concentration. (100  $\mu$ g /mL, 1000  $\mu$ g /mL). Compounds A<sub>4</sub>, C<sub>3</sub>, D<sub>1</sub>, D<sub>4</sub> have shown excellent anticancer activity at all concentrations (1  $\mu$ g /mL, 10 $\mu$ g/mL, 100  $\mu$ g/mL, 1000  $\mu$ g/mL). Cyclophosphamide was used as standard drug.

Increased antitumor activity possessed by derivatives may be because of presence of tetrahydropyrimidine ring. Cl, F derivatives shows increased antitumor activity.

### CONCLUSION

Sixteen [3, 6 (4- substituted aryl)1-(1,2 hydrazino-2oxoethyl)-20x0-1, 2, 3, 4-tetrahydropyrimidine-5-yl] propionic acid derivatives have been synthesized in a four-step reaction. Their structures are confirmed by IR, <sup>1</sup>H NMR and TLC. The antitumor activity of all compounds has been recorded on the basis of reference standard Cyclophosphamide. All the compounds showed antitumor activity. The antitumor data shows that use of cloro and fluoro group in product shows increase in activity. The other compounds shows moderate antitumor activity.

### REFERENCES

- Holmes JT. Prostaglandins, Leukotrienes and other Eicosanoids in Wilson and Giswold's Textbook of Organic Medicinal and Pharmaceutical Chemistry. Lippincott Williams and Wilkins, PA, 2004;11:818-829.
- Huskisson EC. Antirheumatic drugs. Clinical pharmacology and therapeutic series. Praeger Publishers, California, 1983;3:117-125.
- 3. Borne RF. Nonsteroidal Anti-inflammatory agents, Foye's Principles of Medicinal Chemistry, New York, 2002;5:751-793.
- 4. Kadam SS, Mahadik KR, Bothara KG. Nonsteroidal Antiinflammatory agents, Principles of Medicinal Chemistry, 15th edition, Nirali Prakashan, Pune, 2002; 2:138-139.
- Wallace JL, Bure AG. GI-Sparing Anti-inflammatory Drugs: A Promising Future. Inflammation Research Network, University of Calgary, Calgary, Alberta, Canada 1993;3:154-158.
- 6. Bhosale D, Bharambe S, Gairola N, Dhaneshwar SS., Mutual prodrug concept: Fundamentals and applications, *Ind.J.Pharm.Sci*, 2006;68, 3: 286-294.
- Bartsch H, Nair J. Chronic inflammation and oxidative Stress I the genesis and perpetuation of cancer: role of lipid peroxidation, DNA damage and repair. Langenbecks Arch Surg, 2006;391:499–510.
- Lu H, Ouyang W, Huang C. Inflammation, a key event in cancer development. Mol Cancer Res, 2006;4:221-233.
- 9. Winter C A. GI-Sparing Anti-inflammatory Drugs. Proc Soc Exp Bio Med, 1962;1:544.
- 10. Pirsino R, Bainchini F, Banchelli G. 2phenylpyrazolo- 4-ethyl- 4, 7-dihydro [1, 5-a] pyrimidine-7-one for its analgesic, antipyretic and anti-inflammatory activities. Pharmacol Res Comm, 1996;18:241.
- 11. Modica M, Santagati M, Santagat A. Some new thiadiazolothieno-pyrimidinones for antiinflammatory and analgesic activities. Pharmazie, 2000;55:500.

- Cenicola ML, Donnoli D, Stella L. Someimidazolo [1, 2]- pyrimidines for anti-inflammatory, analgesic and antipyretic activities. Pharmacol Res,1990;22:80.
- Nargund LVG, Badiger VV and Yarnal SM. Substituted 2- mercapto-3-(N-alkyl) pyrimido [5, 4-c] cinnolin-4-(3H)-ones for anti-inflammatory and antimicrobial activities. *J Pharm Sci*, 1992;81:365.
- 14. Cottam HB, Wasson DB, Shikh HC. Pyrazolo [3, 4-d] pyrimidine derivatives as potential inhibitor of adenosine kinase. *J Med Chem*, 1993:36.
- Bruni F, Costanzo A, Selleri S. Series of pyrazolo [1, 5- a] pyrimidin-7-one for anti-inflammatory activity. *J Pharm Sci*, 1993; 82:480.
- Tozkoparan B, Ertan M, Kelicen P and Demirdamar. New 2- benzylidene -7 -methy 1- 3- oxo - 5-(substitutedphenyl)-2,3-dihydro-5H thiazolo [3,2] pyrimidine-6-carboxylic acid esters for antiinflammatory activity R. I L Farmaco, 1999;54:588.
- 17. Lee C H, Jiang M, Cowart M. Some 6-substituted pyridopyrimidine analogues as potential AK inhibitors. *J Med Chem*, 2001;44:2133, 2001.
- 18. Boyle DL, Kowluk EA, Jarvis MF. Pyrazolopyrimidine derivatives as potential inhibitor of adenosine kinase. *J Pharmacol Exp Ther*, 2001;296:495.
- 19. Molina P, Aller E, Lorengo A. Pyrido [1, 2] pyrimidines on solid phase using the iminophosphorane methodology and tested for effects on leukocyte function *in-vitro* and anti-inflammatory activity. *J Med Chem*, 2001;44:1011.
- 20. Vidal A, Ferrandiz ML, Ubeda A. Hexahydroimidazo [1, 2] pyrimidine derivatives on leukocyte functions. *J Pharm Pharmacol*, 2001;53:1379.
- 21. Furniss BS, Hannaford AJ, Smith PWG, Tatchel AR. Vogel's textbook of practical organic chemistry, 2004;5:1015-1016.
- 22. Sodhi SM, Dinodia M, Rani R, Shukla R. Synthesis, anti-inflammatory and analgesic activity of some pyrimidine derivatives. *Indian J of Chemistry*, 2009;49B:273-281.
- 23. Pattan SR, Hole MB. Synthesis and evaluation of some novel [1,2,4] triazolo[1,5-α] pyrimidine derivatives for anticancer activity. *Indian J of Chemistry*, 2012;51B:774-779.

© 2018 | Published by IRJSE