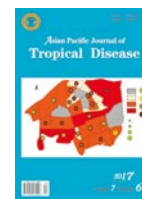


## Asian Pacific Journal of Tropical Disease

journal homepage: <http://www.apjtc.com>Original article <https://doi.org/10.12980/apjtd.7.2017D6-160> ©2017 by the Asian Pacific Journal of Tropical Disease. All rights reserved.Validated HPTLC technique for simultaneous evaluation of biomarkers gallic acid and quercetin in *Trichosanthes dioica* Roxb.: A systematic approach for quality control of herbalsMd. Sarfaraj Hussain<sup>1\*</sup>, Hend Ismail<sup>2</sup>, Sheeba Fareed<sup>3</sup>, Mohammad Ali<sup>4</sup><sup>1</sup>Department of Pharmacognosy, Faculty of Pharmacy, Misurata University, Misurata, Libya<sup>2</sup>Pharmaceutical Technical Department, Faculty of Medical Technology, Misurata University, Misurata, Libya<sup>3</sup>Faculty of Pharmacy, Integral University, Dasauli, Kursi Road, Lucknow 226026, Uttar Pradesh, India<sup>4</sup>Department of Pharmacognosy & Phytochemistry, Faculty of Pharmacy, Hamdard University, Hamdard Nagar, New Delhi 110062, India

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## ABSTRACT

**Objective:** To investigate the simultaneous quantification of two biologically active elements, flavonoids gallic acid (GA) and quercetin (QE), in methanolic extract of *Trichosanthes dioica* that was customarily used to lower blood sugar and cholesterol.**Methods:** Silica gel 60F254 having receded with aluminum plate was applied for the separation. We used toluene-ethyl acetate-formic acid (5:4:1, v/v/v) as a solvent system for the separation.**Results:** At 270 nm in reflectance mode densitometric analysis was performed. Squeezed bands for GA and QE were achieved at  $R_f$  (0.31 ± 0.03) and (0.50 ± 0.02). The scheme was validated for sensitivity, specificity, limit of detection and quantification. Linear relationship was found between the concentration range of 100–1000 ng spot<sup>-1</sup> for GA and 150–900 ng spot<sup>-1</sup> for QE, respectively. The instrumental precision was noted to be 1.03–1.96 (% RSD) and 1.09–1.98 (% RSD) for GA and QE, separately. Methodology specificity was determined at three different study point with varying concentration for recovery studies. Mean % recovery was noted to be 99.4%–99.9% and 98.7%–99.3% and the content estimated as (31.00 ± 1.18)% and (39.00 ± 1.24)% for GA and QE, respectively.**Conclusions:** The developed high performance thin layer chromatography scheme can be used for routine quality control analysis of *Trichosanthes dioica* and several other formulations containing these markers.

## 1. Introduction

Gallic acid (GA) and quercetin (QE) are present in different range of flora. GA and QE are well recognized for its multiple pharmacological effects ranging from anti-inflammatory to anti-hypertension activities[1]. Regular consumption of GA and QE has been shown protective effects in developing cancer and heart diseases[2-4]. Research interest increases in GA and QE because of its potential in improving public health. Concerning the estimation of GA and QE very tiny data are accessible for both qualitative and quantitative analysis. QE 3, 4-diglucoside, QE 4-monoglucoside and QE 3-monoglucoside are the most known glycoside and found

in nature in the form of conjugated glycosides[5]. Many studies have reported QE as a major flavanol in apple, berries, carpers, onions and tea[6], and its protective role towards cardiovascular, cancer and inflammatory conditions[7]. The previous reports reveal that the GA and its derivatives possess wide spectrum of biological activities like antimicrobial[8], anticancer[9], antiviral, anti-inflammatory[10], analgesic[11] and anti-HIV activities[12]. These days, high performance thin layer chromatography (HPTLC) is used as conventional analytical technique due to some of its unique features such as high sample throughput, minimum sample requirement, multiple compounds can be analyzed at a time by using small volume of mobile phase that makes the process fast and economic as well[13-17]. Finger print analysis through HPTLC has appeared as an influential and current instrument that linking the botanical identity and chemical constituent's profile of the plant for the estimation of the markers (chemical and biochemical)[18]. The HPTLC-densitometric method is a renowned scheme for the partition and quantification of natural products. GA and QE have

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been estimated through HPTLC, LC-MS/MS and gas chromatography as an hyphenated techniques[19-24].

*Trichosanthes dioica* Roxb. (Cucurbitaceae) (*T. dioica*) is a vegetative herb usually cultivated in Malaysia, throughout Indian subcontinent and additionally in South Asia. It is usually familiar as “sespadula” in English or “parwal” in Hindi. Vitamin C and minerals like calcium, magnesium, sodium, potassium, copper and sulphur are found in rich source in *T. dioica*. Apart from fruits, leaves and tender shoots have also been used since antique era[24-27]. Medicinal properties like hypocholesterolemic, hypoglyceridemic, and hypophospholipidemic, have been identified in animals[28,29] and also in normal and diabetic human volunteers[30]. Direct feeding of *T. dioica* seeds have shown significant changes in serum lipid profile of normal and mild diabetic human and albino rabbits[31,32]. *T. dioica* effects in the powder form have also been studied on blood sugar and lipid profile of normal albino rabbits[33]. Seeds of the plants also showed antifungal and antibacterial activity and are extensively used in the management of acid dyspeptic ailments.

Being the constituent of this scientific study on the quantification of markers compound of the herbs, this paper explains development, validation and simultaneous estimation of GA and QE in *T. dioica* by HPTLC technique.

## 2. Materials and methods

### 2.1. Plant materials and chemicals

For research and experiment, we took fresh leaves of *T. dioica* from a district of the state of Bihar in India exactly from the basin of the Koshi river which was authenticated by Prof. (Dr.) Anjani Kumar Sinha, Principal, M L T Saharsa College Saharsa, Saharsa, Bihar and for the records a voucher specimen No. SHC 55/05 has been submitted in the herbarium, Department of Botany, M L T Saharsa College Saharsa, 852201, India. Marker drug pure GA (97%) and QE (99%) were procured as samples from Natural Remedies Pvt. Ltd, Bangalore, India. Toluene, ethyl acetate and formic acid all are analytical grade obtained from E. Merck, Mumbai, India. Silica gel 60F254 precoated with aluminum plate were purchased from E. Merck, Darmstadt, Germany.

### 2.2. HPTLC chromatographic conditions

The drugs were implicated by Linomat V (CAMAG, Muttenz, Switzerland) on precoated silica gel 60F254 (E. Merck, Darmstadt, Germany) with a thickness of 0.2 mm aluminium backed plate in the form of bands of thickness six (6 mm) with a CAMAG  $\mu$ L syringe. The solvent system (mobile phase) resided of toluene-ethyl acetate-formic acid (5:4:1, v/v/v). Linear ascending development was performed out in a 20 cm  $\times$  10 cm twin trough glass chamber (CAMAG, Muttenz, Switzerland), earlier inundated & boosted with solvent system for 15 min at normal temperature ( $25 \pm 2$  °C) and comparative moistness of  $60\% \pm 5\%$ . The development comprises

chromatogram track of 8 cm, 20 mL of solvent system, and duration time of 10 min. Five separate volumes (0.2, 0.4, 0.8, 1.0, 2.0  $\mu$ L) of the standard solution of GA were applied on a 20  $\times$  10 cm TLC plate for the preparation of the standard plot of GA. Similarly, five different volumes (0.3, 0.6, 1.2, 2.4, 4.8  $\mu$ L) of a standard solution of QE were applied on a separate 20 cm  $\times$  10 cm silica gel aluminum backed plate for the making of the standard curve of QE. The slit measurement 3 mm  $\times$  0.45 mm and scanning speed were chosen as enhanced apparatus limitations. A persistent use or spraying rate of 160 nL s<sup>-1</sup> and speed of scanning 20 mm s<sup>-1</sup> were fixed for both GA and QE. Subsequent to the development, TLC plates were dried in a flow of air with the use of an air dryer. For densitometric scanning CAMAG TLC scanner III, (absorbance 270 nm) and run by winCATS software[13-17].

### 2.3. Preparation of marker drugs solutions

Ten milligrams of reference drugs GA & QE was weighed accurately transferred to 2 separate 10 mL volumetric flask and dissolve in methanol. For standard plot, GA reference solution (1–10  $\mu$ L) was used to a HPTLC plate to furnish quantities in the range 100–1000 ng spot<sup>-1</sup>; however, QE reference solution (0.5–5  $\mu$ L) was used to provide quantities in the range 150–900 ng spot<sup>-1</sup>. These solutions were used for the preparation of standard curve of GA & QE, respectively.

### 2.4. Preparation of sample solutions

The leaves of *T. dioica* were air-dried and roughly grinded. A total of 500 g of the powdered material extracted thoroughly with a Soxhlet apparatus with methanol for 72 h. For obtaining a dark brown viscous mass (14.9%), the methanolic extract was concentrated at low pressure.

#### 2.4.1. Method validation

The established scheme was validated according to the ICH guidelines (ICH, 2005) parallel to the other chromatographic HPTLC approaches reported by laboratory[25,34] which are in custom for the quality control of plant medicines.

#### 2.4.2. Precision

The precision of the scheme was resulted by measuring the repeatability of the sample application and the dimension of peak areas for six imitates of the marker compound solution at the 3 amounts (150, 300 and 600 ng spot<sup>-1</sup> for GA & 200, 400 and 800 ng spot<sup>-1</sup> for QE correspondingly). Inter-analyst precision was carried out by quoting the identical method by a different analyst. The intermediate precisions were determined regarding % RSD of the area.

#### 2.4.3. Robustness

The robustness of the suggested scheme was planned at 3

concentration, stages (100, 200 & 400 ng/spot for GA & QE respectively) in two dissimilar ways, *i.e.* by altering the composition of solvent phase and the UV wavelength. The % RSD of the area of peaks was intended to consider the robustness of the method. Solvent system (mobile phase) developed from toluene: ethyl acetate: formic acid (5:4:1, v/v/v) in diverse extents (5.6:3.4:1, v/v/v, 5.5:3.5:1, v/v/v, 5.3:5:1.5, v/v/v, and 6.5:2.5:1, v/v/v) were used for chromatography. The volume of solvent system and the time of saturated exploration were (20 ± 2) mL (18, 20 and 22 mL) and (20 ± 10) min (10, 20 and 30 min), sequentially. The TLC plates were initiated at (60 ± 5) °C for 2, 5 and 7 min earlier chromatography.

#### 2.4.4. Limit of detection (LOD) and limit of quantification (LOQ)

To calculate the LOD and LOQ, blank methanol ( $n = 6$ ) was applied and the standard deviation ( $\sigma$ ) of the methodical rejoinder was ascertained.

$$\text{LOD} = 3.3(\text{SD})/S$$

$$\text{LOQ} = 10(\text{SD})/S$$

where SD is the official deviation of rejoinder (peak area) and S is the aggregate of the slope of the standard curve.

#### 2.4.5. Recovery studies experiment

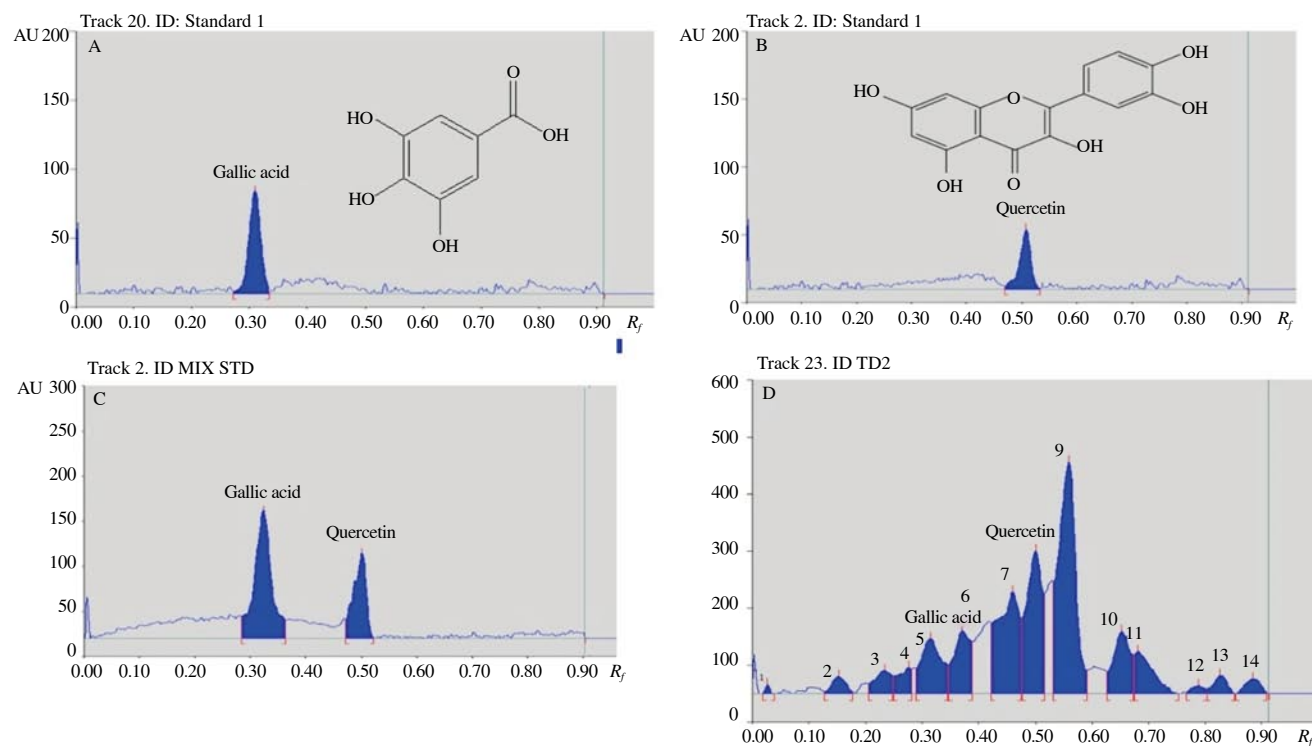
The accuracy of the current study was examined by using the scheme to drug samples to which known quantity of the standard drugs 50%, 100%, and 150% of the GA & QE had been added. Each concentration was examined in three times. The recovery of the standard drug at various concentration in the extract was estimated.

### 3. Results

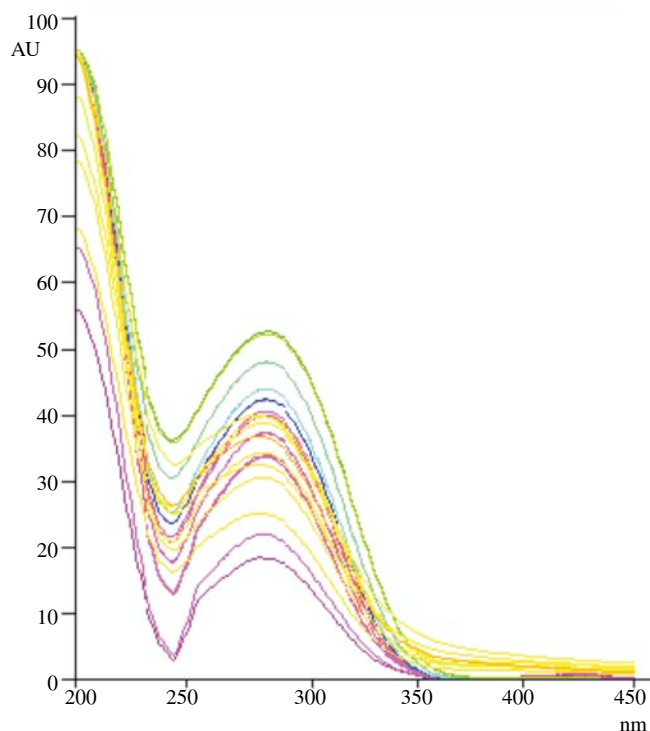
#### 3.1. Optimization of mobile phase

Under chamber saturated conditions chromatogram was developed for both GA and QE. Toluene: ethyl acetate: formic acid (5:4:1, v/v/v) was used as solvent phase (mobile phase) (Figure 1A and B) and was too used for the separation of *T. dioica* methanolic extracts (Figure 1D). The adjusted saturation period was found to be 10 min. Maximum absorbance's for the spots were observed at 270 nm under UV spectra. At 270 nm in reflectance mode densitometric analysis was performed (Figure 2). For GA & QE high resolution sharp, symmetrical, compact bands were obtained at  $R_f$  (0.31 ± 0.03) and (0.50 ± 0.02) separately (Figure 1C).

As per the author knowledge, to date, no studies have reported any HPTLC technique to concurrently estimate GA and QE in *T. dioica* herb or extracts. So, we made an attempt to produce a price operative HPTLC scheme to estimate natural constituent's in this herb. At  $R_f$  0.31 and 0.50, GA and QE were well resolved (Figure 1A and B) from *T. dioica* methanolic extract. Visualization of the plates was done at dual altered wavelengths 254 and 270 nm, which shows the adaptable spectrum range. In accumulation, this also aids in differentiation of species by providing a better fingerprint data which helps in graphical documentation of separate compounds. This scheme was found to be reasonably effective because of its high baseline resolution. The uniqueness of the peaks of compounds 1–14 in the methanolic extracts was established by overlapping their UV absorption spectra at 270 nm with standards.



**Figure 1.** HPTLC chromatogram of (A) standard GA at  $R_f$  0.31; (B) standard QE at  $R_f$  0.50; (C) GA and QE simultaneously determined in by using toluene: ethyl acetate: formic acid (5:4:1, v/v/v) as solvent system scanned at 270 nm [GA (0.31) and QE (0.50)] and (D) methanolic extract of *T. dioica* scanned at 270 nm [peak 1–14; GA (0.31) and QE (0.50)].



**Figure 2.** Overlay UV absorption spectra of standard GA, QE in sample methanolic extract of *T. dioica* and sample track at start, apex and end positions.

### 3.2. Calibration curve of GA and QE

Linear regression calculation and correlation factors were used to validate the linearity of compounds (GA and QE). A linear six-point calibration curve was obtained in the range of 100–1 000 ng spot<sup>-1</sup> and 150–900 ng spot<sup>-1</sup> for GA and QE, respectively. Developed method was found to be linear in nature as reflected from its regression equation and correlation coefficient of marker compound,  $Y = 0.0048X + 0.012$  ( $r^2 = 0.9991$ ) for GA and  $Y = 0.033X - 0.017$  for QE ( $r^2 = 0.9956$ ) (Table 1). The mean values ( $\pm$  SD) of the slope were  $0.0048 \pm 0.0003$  and  $0.033 \pm 0.008$  and intercepts were  $0.012 \pm 0.007$  and  $0.017 \pm 0.002$  for GA and QE, individually. No noteworthy alteration was perceived in the slopes of standard plots (ANOVA,  $P > 0.05$ ).

### 3.3. Method validation

#### 3.3.1. Precision

Precisions were determined and reported in terms of % RSD particularly intermediate. This incorporates data of inter-day, intra-day and inters analyst precisions. Accuracy of the developed methods indicated that the scheme was well adaptable for the repetitive study of GA and QE which was further supported by the low % RSD. The

**Table 1**

$R_p$  linear regression data for the calibration curve and sensitivity parameter for GA and QE.

Parameter	$R_f$	Linearity range (ng/spot)	Regression equation	Correlation coefficient ( $r^2$ )	Slope $\pm$ SD	Intercept $\pm$ SD	SE of slope	SE of intercept	LOD (ng/spot)	LOQ (ng/spot)
GA	0.31	100–1 000	$Y=0.0048X + 0.012$	0.9991	$0.0048 \pm 0.0003$	$0.012 \pm 0.007$	0.00017	0.0040	37.5	112.5
QE	0.50	150–900	$Y=0.033X - 0.017$	0.9956	$0.0330 \pm 0.0080$	$0.017 \pm 0.002$	0.00460	0.0011	69.0	207.0

**Table 2**

Precision of inter-day, intra-day and inters system analysis.

Elements	Conc. (ng spot <sup>-1</sup> )	Inter-day precision		Intra-day precision		Inter-system precision	
		Mean peak area $\pm$ SD	% RSD	Mean peak area $\pm$ SD	% RSD	Mean peak area $\pm$ SD	% RSD
Gallic acid	150	3 794.90 $\pm$ 34.78	1.90	3 908.50 $\pm$ 51.06	1.65	3 817.30 $\pm$ 63.18	1.80
	300	5 127.40 $\pm$ 27.49	1.54	5 190.10 $\pm$ 54.71	1.10	5 211.90 $\pm$ 49.34	1.77
	600	6 207.70 $\pm$ 39.77	1.07	6 311.70 $\pm$ 54.45	1.12	6 388.50 $\pm$ 67.98	1.03
Quercetin	200	1 708.70 $\pm$ 39.36	1.98	1 790.10 $\pm$ 49.96	1.73	3 817.00 $\pm$ 63.18	1.70
	400	2 582.90 $\pm$ 72.49	1.76	2 457.70 $\pm$ 44.43	1.89	5 211.00 $\pm$ 49.34	1.65
	800	2 867.50 $\pm$ 79.83	1.09	2 835.40 $\pm$ 54.45	1.17	6 388.00 $\pm$ 67.98	1.39

SD: Standard deviation; RSD: Relative standard deviation.

**Table 3**

Robustness of the method.

Mobile phase change (toluene-ethyl acetate-formic acid)			Mean peak area $\pm$ SD		% RSD	
Actual (v/v/v)	Used (v/v/v)	Level	GA	QE	GA	QE
5:4:1	5.5:3.5:1	-2	3 490.30 $\pm$ 59.62	1 880.70 $\pm$ 17.34	1.12	1.19
	5:4.5:0.5	0	3 375.80 $\pm$ 19.27	1 795.90 $\pm$ 39.55	0.53	0.61
	6:3:1	+2	3 205.50 $\pm$ 19.03	1 949.30 $\pm$ 40.36	1.09	1.22
Wavelength change						
Actual (270 nm)	Used (nm)	Level				
	268	-2	3 629.80 $\pm$ 39.36	1 859.70 $\pm$ 89.26	0.79	1.02
	270	0	3 567.30 $\pm$ 39.36	1 798.90 $\pm$ 73.69	0.98	0.69
	272	+2	3 709.50 $\pm$ 39.36	1 718.50 $\pm$ 94.98	1.73	1.87

RSD: Relative standard deviation.

results of the analysis are depicted in Table 2.

### 3.3.2. Robustness

The outcome of deliberate changes in the composition of solvent system (mobile phase) and detection wavelength was analyzed as % RSD and is presented in Table 3. Robustness of method was indicated by low percentage of RSD.

### 3.3.3. LOD and LOQ

We found LOD values 37.5 and 69.0 ng spot<sup>-1</sup> for GA & QE respectively; however, LOQ values were found to be 112.5 and 207.0 ng spot<sup>-1</sup> for GA and QE, respectively (Table 1), which shows the adequate assay sensitivity. The value of the lowest part of the calibration curve was used to represent LOD and LOQ. This proves good sensitivity of the proposed methods for the estimation of above marker drugs.

### 3.3.4. Recovery studies

At three quality control levels fortification of the samples were done which gives good repossessions for GA and QE. Afterward sample dispensation and spreading over percent recoveries for both markers were in the range of 99.4%–99.9% (GA) and 98.7%–99.4% (QE) as shown in Table 4.

**Table 4**

Recovery studies of GA and QE.

Concentration added to analyte (%)	Theoretical (ng)	Added (ng)	Detected (ng)	Recovery (%)	RSD (%)	
GA	50	200	497.3	99.4	1.92	
	100	300	400	698.6	99.80	1.51
	150		600	899.1	99.90	1.49
QE	50		100	197.5	98.7	1.74
	100	100	200	298.2	99.4	1.92
	150		300	397.4	99.3	1.27

### 3.4. Estimation of bioactive GA and QE in methanolic extract of TD

The percentages of GA and QE in methanolic extract of *T. dioica* estimated by developed method were (0.31 ± 1.18)% and (0.39 ± 1.24)%, respectively. A faster, precise, and cost effective HPTLC scheme has been established for the simultaneous quantitation of two biologically active markers in *T. dioica*.

## 4. Discussion

Quality of herbal medicines is judged by the presence of bioactive compounds. HPTLC fingerprint profile technique is considered as the golden standard to determine the bioactive components of the herbal medicine. HPTLC fingerprinting is accurate for herbal identification, authentication and standardization because of its liner and precise nature. This technique is also useful in quality control of herbal drugs and examination for the contaminant. It is also proved to be beneficial to the assessment of different pharmaceutical formulations

in the market and plant orderly revisions[35,36]. As the demand for herbal products as medicines and cosmetics is growing, there is an urgent need for standardization of plant products. The HPTLC method is an important tool for standardization, phytochemical outlining and quantitation of compounds present in herbs[37]. Chromatographic fingerprinting is one of the rationale tools to fulfill the essential for economical and powerful quality assessment to folklore medicine throughout the earth. The enhanced chromatographic fingerprinting technique is also used to characterize the different chemical constituents widely found in various plant medicine and to preserve such “catalogue” for further multifacial justifiable analysis. HPTLC fingerprinting is proved to be a simple and reliable tool for the identification, authentication and quality control of herbal drugs.

Introducing HPTLC in pharmaceutical analysis is a major step in terms of quality assurance. In addition to high sensitivity and specificity, the developed HPTLC technique is simple and cost-effective for the simultaneous quantification of GA and QE compounds. This method is an effective tool for routine quality control of herbal materials as well as formulations containing any or both of these compounds as proved by statistical analysis.

### Conflict of interest statement

We declare that we have no conflict of interest.

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