

CODEN [USA]: IAJPBB ISSN: 2349-7750

INDO AMERICAN JOURNAL OF PHARMACEUTICAL SCIENCES

http://doi.org/10.5281/zenodo.1217989

Available online at: http://www.iajps.com

Research Article

SYNTHESIS OF 2-[2-HYDROXY-5-(3-SUBSTITUTEDIMINO-1, 2, 4-DITHIAZOLO)] AMINO PHENYL INDOLES

R.D.Isankar and D.T.Tayade*

Department of Chemistry, Government Vidarbha Institute of Science and Humanities, Amravati 444 604, Maharashtra State, India.

Abstract:

2-[2-Hydroxy-5-(-2,4-substituted-dithiobiureto)]phenylindoles with iodine in ethanol as an oxidative cyclising agent was transformed into the 2-[2-hydroxy-5-(3-substitutedimino-1,2,4-dithiazolo)]aminophenylindoles in one step. Characterization and justification of structure of synthesized compounds were done on the basis of conventional elemental analysis, chemical characteristics and spectral studies.

Key Words: 2-[2-Hydroxy-5-(2,4-ethyldithiobiureto)]phenylindole, iodine, ethanol.

Corresponding Author:

D.T.Tayade,

Department of Chemistry,

Government Vidarbha Institute of Science and Humanities,

Amravati 444 604, Maharashtra State, India.

E-mail:skdtayade@gmail.com, ramisankarrd@gmail.com



Please cite this article in press R.D.Isankar and D.T.Tayade., Synthesis of 2-[2-Hydroxy-5-(3-Substitutedimino-1, 2, 4-Dithiazolo)] Amino Phenyl Indoles, Indo Am. J. P. Sci, 2018; 05(04).

INTRODUCTION:

Diverse biological and pharmacological activities have been reported for thiazoles, dithiazines and related compounds. For instance, thiazole derivatives are in clinical use [1,2] and many dithiazines exhibit antiprotozoal, antiviral, bactericidal and fungicidal properties [3,4], probably by virtue of the presence of the toxophoric (-N=C-S) group. Furthermore, many dithiols and dithiazines are patented as synthetic flavor compounds [5-9] and in photographic developing by a diffusion transfer process.In this laboratory Kawale has designed thiadiazoloacrilidine and studied its antimicrobial activities. Khobragade synthesized thiadizolopyridine and studied antibacterial activities and its effect on germination pattern of on jowar, Bhagwatkar

Taking all these things into consideration, it was thought interesting to investigate the oxidative cyclisation of 2-[-2-hydroxy-5-(-2,4-substituteddithiobiureto)]phenylindoles by making use of iodine in ethanol as an oxidative agent for synthesizing yet novel series of 2-[-2-hydroxy-5-(3-substitutedimino-1,2,4-dithiazolo) laminophenylindole. This is easiest, cheaper, suitable, convenient and less time consumable method for the synthesis of dithiazoles which is depicted below in (Scheme-I).

HO

2-[2-Hydroxy-5-(2, 4-substituted-dithiobiureto)] phenylindoles

2-[2-Hydroxy-5-(3-substitutedimino-1, 2, 4-dithiazolo)]aminophenylindoles

Where, R = -methyl, -ethyl, -t-butyl -phenyl, -p-Cl-phenyl.

Synthesis of 2-[-2-hydroxy-5-(3-ethylimino-1,2,4dithiazolo)]aminophenylindole

In a china dish paste of 2-[2-hydroxy-5-(2,4ethyldithiobiureto)]phenylindole was prepared in ethanol to it iodine in ethanol was added with constant stirring, till colour of iodine persists to reaction mixture. Then it was kept at room condition

for 4 hours. On basification with dilute ammonium hydroxide solution, dark yellow crystals were obtained, recrystallized from ethanol, yield 84%, m.p. 179°C.

The probable reaction and mechanism for the formation of is depicted below,

Reaction

2-[2-Hydroxy-5-(3-ethylimino-1, 2, 4-dithiazolo)]aminophenylindoles

RESULT AND DISCUSSION:

Molecular formula $C_{18}H_{16}N_4O_1S_2$ %Composition: found(calculated) C-57.32 (58.69), H-4.02 (4.34), N-15.20 (15.21), S-16.35 (17.39)., **IR** (**KBr Pellets**) **v cm**⁻¹- 3405.2 (O-H; s), 3356 (N-H; s), 3131 (Ar-CH; s), 1688 (N=C-N; s), 1636 (Ar-C=C), 1294 (C-N; s)., 2866.22(C-H), 1524 (C = N) 777(C-S; s) **The PMR Spectrum:** PMR Spectrum of compound was carried out in CDCl₃ and DMSO-d₆. This spectrum distinctly displayed the signals due to indole -NH proton at δ 12.8560, ppm, -NH proton at 9.6265 ppm flanked thioamido δ Ar-OH proton at 4.0025δ ppm, Ar-H protons at δ 7.4358-6.8254 ppm, -CH₂ protons at δ 2.5367-2.3605 ppm, -CH₃ protons at δ 1.6202 ppm.

Similarly, 2-[-2-hydroxy-5-(3-methylimino-1,2,4-dithiazolo)]aminophenyl-indole, 2-[-2-hydroxy-5-(3-methylimino-1,2

t-butylimino-1,2,4-dithiazolo)]aminophenylindole, 2-[-2-hydroxy-5-(3-phenylimino-1,2,4dithiazolo)]aminophenylindole and 2-[-2-hydroxy-5-(3-p-chlorophenyl imino-1,2,4dithiazolo)]aminophenylindole were synthesized by the oxidative cyclisation of 2-[2-hydroxy-5-(-2,4methyldithiobiureto)]phenylindole, 2-[2-hydroxy-5-(2,4-t-butyl)dithiobiureto)]phenylindole, hydroxy-5-(-2,4-phenyldithiobiureto)] phenylindole 2-[2-hydroxy-5-(2,4-pchlorophenyldithiobiureto)]phenyl respectively by making use of iodine in ethanol as oxidative cyclising agent by the above mentioned and enlisted in Table No.-1

Sr. No.	2-[-2-Hydroxy-5-(3- <i>substitued</i> imino-1,2,4-dithiazolo)]aminophenylindoles (VIa-e)	Yield (%)	M.P. °C
1.	methyl	89	178
3.	t-butyl	86	188
4.	phenyl	93	156
5.	p-chlorophenyl	87	187

REFERENCES:

- Martinez A., Alonso M., Castro A., Dorronsoro I., Gelpi J. L., Luque F. J., Perez C., Moreno F. J., J. Med. Chem. 2005, 48, 7103
- Kanda Y., Aria H., Yamaguchi H., Ashizawa T., Ikeda S., Murakata C., Tamaoki T., U. S. US 6,080,771 (2000), WOAppl 1997/JP 4,584 (1997); Chem. Abstr. 2000, 133, 674.
- 3. Al-Nakib T., Meegan M. J., Burke M. L., *J. Chem. Research* **1994**, S, 170; (M) 1042 (1994).
- 4. E. Brouhet, B. Dupon, *Arzneim.-Forsch. / Drug Research* **1992**, 42 705.
- Frenck R. W. (Jr.), Mansour A., Nakhla I., Sultan Y., Putnam S., Wierzba T., Morsy M., Knirsch C., Clinical Infectious Diseases 2004, 38, 951; Chem. Abstr. 2004, 141, 46.
- Pletz M. W. R., Rau M., Bulitta J., de Roux A., Burkhardt D., Kruse G., Kurowski M., Nord C. E., Lode H., Antimicr. Agents and Chemotherapy, 2004, 48, 3765; Chem. Abstr. 2004, 141, 29.

- Prithiviraj B., Vikram A., Kushalappa A. C., Yaylayan V., *Eur. J. Plant Pathology*, **2004**, 110, 371; Chem. Abstr. 2005, 142, 52335.
- K. Kawasaki, Jpn. Kokai Tokyo Koho Jp. 2005
 15, 684 (C11B9/00), Appl. 2003/184, 480 (2003), Jpn; Chem. Abstr. 2005, 142, 967.
- J. C. Jung, G. S. Lee, G. S. Shin, K. R 2003 45, 471 (C1G03F7/004) 2003, Appl. 76,195 (2001); Chem. Abstr. 2005, 142, 1886.
- 10. Singh A. K. Mishra G. and Jyoti K., *Journal of App. Pharm. Science*, 1(5), **2011**, 44-49.
- 11. Tao J., Duo Z. W. and Ling H. C., *J Chinese Chem Soc.*, 57, **2010**, 1077-80.
- 12. Madhav N. V., Nayak A. S., Rao A. J., and Sarangapani M., *J Pharm Res.*, 4(5), **2011**, 1396-97.
- 13. Patel D. and Singh A., *E-Jour Chem.*, 6(4), **2009**, 1017-1022.
- 14. Mahalle P. R. and Deshmukh S. P., *Int. J. of Pharmcy and Pharmcaeutical Science*, 3(4), **2011**, 277-279.
- 15. Saleh M. A., Sulfur letters, 25, **2002**, 235-245.